

The full infrared spectrum of molecular hydrogen [★]

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ABSTRACT

Context. The high spectral resolution $R \sim 45,000$ provided by IGRINS (Immersion Grating INfrared Spectrometer) at MacDonald Observatory and $R \sim 100,000$ achieved by CRIRES (CRyogenic high-resolution InfraRed Echelle Spectrograph) at VLT (Very Large Telescope) challenges the present knowledge of infrared spectra.

Aims. We aim to predict the full infrared spectrum of molecular hydrogen at a comparable accuracy.

Methods. We take advantage of the recent theoretical *ab initio* studies on molecular hydrogen to compute both the electric quadrupole and magnetic dipole transitions taking place within the ground electronic molecular state of hydrogen.

Results. We computed the full infrared spectrum of molecular hydrogen at an unprecedented accuracy and derive for the first time the emission probabilities including both electric quadrupole ($\Delta J = 0, \pm 2$) and magnetic dipole transitions ($\Delta J = 0$) as well as the total radiative lifetime of each rovibrational state. Inclusion of magnetic dipole transitions increases the emission probabilities by factors of a few for highly excited rotational levels, which occur in the 3-20 μ range

Key words. Physical data and processes – Molecular data – Infrared: general

1. Introduction

Molecular hydrogen (H_2) is a symmetric molecule, which prohibits electric dipole transitions to occur within its $X^1\Sigma_g^+$ ground electronic state. Then electric quadrupole and magnetic dipole transitions can take place, reciprocally submitted to $\Delta J = 0, \pm 2$ and $\Delta J = 0$ selection rules. Considerable attention has been paid to the *ab initio* studies of this simple molecule where the resolution of the one dimensional Schrödinger equation allows the energy spectrum of the ground electronic state to be derived, as described in the pioneering work by Kołos & Wolniewicz (1964). Various corrections have been further introduced in order to compare the theoretical *ab initio* values to the experimentally derived values by Dabrowski (1984) from the VUV (Vacuum Ultra-Violet) absorption and emission flash discharge spectra of H_2 in the Lyman and Werner bands. The rotation-vibration energy level measurements were derived for all vibrational levels $v = 0-14$, and a maximum rotational J value of $J = 29$ with a predicted accuracy of 0.1 cm^{-1} . These measurements challenged theoretical calculations as the discrepancy between experimentally derived values and theoretical calculations reached about 4 cm^{-1} for highly excited rotational levels. An additional important step in the ground level energy determinations was provided by the study of the quadrupole infrared spectrum at the laboratory (Bragg et al. 1982; Jennings & Brault 1983), which provides level energy terms with an accuracy of about 0.001 cm^{-1} . These studies, which are limited to energy terms of low vibrational and rotational values, have prompted detailed fundamental *ab-initio* studies of the H_2 spectrum where non-adiabatic, relativistic, quantum electrodynamic (QED) corrections are carefully introduced by Pachucki & Komasa (2009) and Komasa et al. (2011). In parallel, astrophysical observations in the in-

frared have allowed the detection of tens of highly H_2 excited emission rovibrational transitions occurring in so-called Photon dominated regions (PDRs) such as the Orion Bar (Kaplan et al. 2017) or NGC 7023 (Le et al. 2017) or in shocked regions such as the Orion KL outflow (Oh et al. 2016; Geballe et al. 2017) and Herbig-Haro objects (Pike et al. 2016). Forthcoming infrared facilities, such as CRIRES (CRyogenic high-resolution InfraRed Echelle Spectrograph) at VLT (Very Large Telescope) and JWST (James Webb Space Telescope) in space will allow the wavelength window and the sensitivity to be extended. We find that it is thus timely to provide the complete infrared spectrum of H_2 involving any possible transition linking all available rovibrational levels at the highest level of accuracy.

2. Infrared spectrum of ground state H_2

2.1. Quantum mechanical calculations of line positions

Individual energy levels and their wave functions were determined in the framework of the nonrelativistic quantum electrodynamics (NRQED) (Caswell & Lepage 1986; Pachucki 2005). This theory is suitable for bound energy levels of atomic and molecular systems composed of light nuclei. Within this approach, the energy is expressed in the form of an expansion in powers of the fine structure constant α

$$E(\alpha) = E^{(2)} + E^{(4)} + E^{(5)} + E^{(6)} + E^{(7)} + O(\alpha^8), \quad (1)$$

where $E^{(i)}$ is proportional to α^i and may contain powers of $\ln \alpha$. The subsequent terms of this expansion are commonly known as the nonrelativistic energy $E^{(2)}$, the relativistic $E^{(4)}$, the QED $E^{(5)}$, and higher order $E^{(i)}$, $i > 5$, corrections. Each term, in

turn, can be expanded in the small electron-to-proton mass ratio m_e/M_p as implemented in the nonadiabatic perturbation theory (NAPT) by Pachucki & Komasa (2008, 2015). The nonrelativistic energy $E^{(2)}$ is then obtained by solving the nonadiabatic radial Schrödinger equation composed of the Born-Oppenheimer, the adiabatic correction, and the nonadiabatic correction potentials (Pachucki & Komasa 2009). For H_2 these electronic potentials are known to the relative accuracy of $10^{-9} - 10^{-14}$ (Pachucki 2010; Pachucki & Komasa 2014, 2015), which enables the non-relativistic dissociation energy of an individual level to be determined with the accuracy limited only by the missing higher order nonadiabatic corrections $\sim (m_e/M_p)^3$. The latter have recently been found to be of the order of $10^{-4} - 10^{-5} \text{ cm}^{-1}$ (Pachucki & Komasa 2018).

The hydrogen molecule in its electronic ground state ($X^1\Sigma_g^+$) accommodates 302 bound states¹. An apparent advantage of the NAPT approach is that once the necessary potentials are constructed all the energy levels can be obtained simultaneously. An alternative to the NAPT and significantly more accurate approach to the nonrelativistic energy has recently been developed by Pachucki & Komasa (2018). In this approach, the $E^{(2)}$ of a selected energy level is evaluated directly in a four-body calculation without separation of electronic and nuclear movements. The accuracy of such calculations is limited by the current precision of the physical constants and amounts to 10^{-7} cm^{-1} for the dissociation energy of an individual level. At present, for about one-fifth of all the bound levels much more accurate nonrelativistic energies are available.

The NAPT was also applied to evaluate the relativistic correction $E^{(4)}$. The electronic relativistic potential for H_2 was determined with accuracy higher than 10^{-6} cm^{-1} by Puchalski et al. (2017). Recently, Czachorowski et al. (2018) evaluated the leading order recoil $\sim m_e/M_p$ correction to this potential so that currently its accuracy is limited by the unknown higher order terms $\sim (m_e/M_p)^2$ and is of the order of 10^{-6} cm^{-1} . The uncertainty introduced by the $E^{(4)}$ term to the dissociation energy of a level is estimated as 10^{-5} cm^{-1} .

The QED term $E^{(5)}$ was obtained from the electronic QED potential determined by Piszczatowski et al. (2009) and later refined by Puchalski et al. (2016) and Puchalski et al. (2017). So far, no numerical values for the finite nuclear mass (i.e. recoil) correction to the QED potential are known. This missing correction currently limits the overall accuracy of the molecular levels energy predicted by theory. The uncertainty resulting from this missing contribution is estimated as $2 E^{(5)}/M_p \sim 10^{-4} \text{ cm}^{-1}$.

The higher order $E^{(6)}$ QED correction was evaluated in the non-recoil limit by Puchalski et al. (2016). The accuracy of this correction is limited by two factors that contribute errors of the same order—the unknown finite-nuclear-mass corrections and the numerical convergence. The final uncertainty on the energy of a level is estimated as 10^{-6} cm^{-1} or less.

Complete expressions for the higher order corrections $E^{(i)}$, $i > 6$, are unknown. For this reason, their numerical values were estimated by Czachorowski et al. (2018), on the basis of the dominating terms selected by analogy to atomic hydrogen, with relatively large uncertainties. However these corrections, as well as the other tiny contributions like the finite nuclear size correction, contribute at the level of 10^{-4} cm^{-1} or less to the total dissociation energy.

As mentioned above, the final accuracy of a single energy level is currently restricted by the accuracy of QED contribution and changes slowly from state to state. There is, however, a

Table 1. Calculated contributions (and uncertainties) to S(1) transition energy in H_2 fundamental band. Values are given in cm^{-1} .

Contribution	1-0 S(1)
$E^{(2)}$	4 712.895 971 8(1)
$E^{(4)}$	0.034 872 6(2)
$E^{(5)}$	-0.025 99(3)
$E^{(6)}$	-0.000 231 9(7)
$E^{(7)}$	0.000 013(3)
Total	4 712.904 64(3)

Notes. 1-0 S(1) corresponds to $v_u=1$, $J_u=3$, $v_l=0$, $J_l=1$, where $v_u, J_u; v_l, J_l$ are the vibrational and rotational quantum numbers of the upper (u) and lower (l) levels of the transition.

significant cancellation of the errors carried by individual levels involved in a transition energy. This is particularly pronounced for near-lying levels. This cancellation was controlled for each energy contribution separately. The final uncertainty determined this way accompanies each line position (see $\Delta\sigma$ column in Table 2). The above uncertainty discussion is illustrated by the numerical data displayed in Table 1 concerning the 1-0 S(1) transition. More details on the theoretical and computational procedures outlined above can be found in Komasa et al. (2019).

Both electric quadrupole and magnetic dipole transitions may occur within the X electronic ground state. Below we discuss separately both types of the transitions and supply working formulas for them.

2.2. Electric quadrupole transitions

Quadrupole vibration-rotation transition probabilities of molecular hydrogen were first computed by Turner et al. (1977) for vibrational levels up to $v=14$ and rotational levels up to $J=20$. These calculations were subsequently improved and extended by Wolniewicz et al. (1998) for all available bound rovibrational levels of H_2 with a more accurate quadrupole moment function $Q(r)$. The electric quadrupole emission probability $W_{v'J' \rightarrow v''J''}$ in s^{-1} is formulated as given in theoretical textbooks (e.g., Sobelman (2006)):

$$W_{v'J' \rightarrow v''J''} = \frac{\alpha}{15} \cdot \frac{\omega^5}{c^4} \cdot \frac{1}{(2J' + 1)} \cdot \sum_{M' M''} \left| \langle J' M' f_{v', J'}(r) | r^2 P_2(\cos \theta) | J'' M'' f_{v'', J''}(r) \rangle \right|_{CGS}^2, \quad (2)$$

where all quantities are expressed in CGS units. $f_{v, J}(r)$ is the radial wavefunction of the ground state H_2 molecule, solution of the radial Schrödinger equation, corresponding to the discrete eigenvalue $E_{v, J}$. The symbols J, M stand for the Y_J^M spherical harmonics, solution of the angular part of the Schrödinger equation. The ω is the angular frequency of the transition. Introducing the wavenumber $\sigma = \frac{\omega}{2\pi c}$ and numerical values for the physical constants, the previous expression becomes

$$W_{v'J' \rightarrow v''J''} = 1.4286 \times 10^{11} \cdot \sigma^5 \cdot \frac{1}{(2J' + 1)} \cdot \sum_{M' M''} \left| \langle J' M' f_{v', J'}(r) | r^2 P_2(\cos \theta) | J'' M'' f_{v'', J''}(r) \rangle \right|_{CGS}^2. \quad (3)$$

The sum over the various substates M', M'' gives rise to the $f(J', J'')$ angular coefficients, as reported in Wolniewicz et al. (1998), with specific selection rules corresponding to O ($\Delta J = -2$), Q ($\Delta J = 0$) and S ($\Delta J = 2$) transitions.

¹ The $v'=14, J'=4$ level is considered as the highest bound level.

To evaluate the electric quadrupole moments we employed the radial function (in a.u.)

$$Q(r) = \frac{r^2}{2} - \frac{1}{2} \langle \phi | \sum_a r_a^2 (3 \cos^2 \theta_a - 1) | \phi \rangle \quad (4)$$

with the expectation value evaluated with the Born-Oppenheimer wave function ϕ . The calculated values of $Q(r)$, employed already in Pachucki & Komasa (2011) and in Campargue et al. (2012), are in agreement with those obtained by Wolniewicz et al. (1998) except that the latter represent twice the $Q(r)$. The working equation used for computing the emission probabilities is

$$W_{v'J' \rightarrow v''J''} = 1.4286 \times 10^{11} \cdot a_0^4 \cdot \sigma^5 \cdot \frac{1}{(2J' + 1)} \cdot \sum_{MM'} |\langle JM f_{v',J'}(r) | Q(r) | J' M' f_{v'',J''}(r) \rangle|^2, \quad (5)$$

$$W_{v'J' \rightarrow v''J''} = 1.12 \times 10^{-22} \cdot \sigma^5 \cdot \frac{1}{(2J' + 1)} \cdot \sum_{MM'} |\langle JM f_{v',J'}(r) | Q(r) | J' M' f_{v'',J''}(r) \rangle|^2, \quad (6)$$

where the wavenumbers σ are expressed in reciprocal centimeters (cm^{-1}).

2.3. Magnetic dipole transitions

The possibility of magnetic dipole transitions was raised by Pachucki & Komasa (2011) who reported the corresponding transition moment $g(r)$ as a function of the internuclear distance r and computed the corresponding emission probabilities within the $v = 1 \rightarrow 0$ transition. The emission probability, expressed in s^{-1} is given by:

$$W_{v'J' \rightarrow v''J''} = \frac{4\alpha\omega^3}{3c^2} \cdot \frac{\alpha^2 a_0^2}{4} \cdot \left(\frac{m_e}{M_P} \right)^2 \cdot \frac{1}{(2J' + 1)} \cdot \sum_{M'M''} |\langle J' M' f_{v',J'}(r) | g(r) \cdot \mathbf{J} | J'' M'' f_{v'',J''}(r) \rangle|_{CGS}^2 \quad (7)$$

(Sobelman 2006). Only $\Delta J = 0$ and $M' = M''$ transitions are allowed in this case from the \mathbf{J} matrix element:

$$\langle JM | \mathbf{J} | JM \rangle = \sqrt{J(J+1)(2J+1)}. \quad (8)$$

Introducing the wavenumber σ as previously and using the transition moment function $g(r)$ in atomic units we obtain:

$$W_{v'J \rightarrow v''J} = 8.00 \times 10^{-18} \cdot \sigma^3 \cdot J(J+1) \cdot |\langle f_{v',J}(r) | g(r) | f_{v'',J}(r) \rangle|^2. \quad (9)$$

2.4. Present computations

The emission probabilities involve both the transition wavenumbers and the radial integration of the corresponding matrix elements. We computed the matrix elements from the $f_{v,J}(r)$ solutions of the one dimension radial Schrödinger equation corresponding to the H_2 ground state, by using the renormalized Numerov method (Johnson 1977):

$$-\frac{\hbar^2}{2M} \frac{d^2 f_{v,J}(r)}{dr^2} - \frac{\hbar^2}{2M} \frac{J(J+1)}{r^2} f_{v,J}(r) + V(r) f_{v,J}(r) = E_{v,J} f_{v,J}(r),$$

(10)

where $M = M_P/2$ is the nuclear reduced mass of H_2 , $V(r)$ is the adiabatic potential function composed of the Born-Oppenheimer potential reported in Pachucki (2010) and the adiabatic correction function presented in Pachucki & Komasa (2014).

The $E_{v,J}$ eigenvalues correspond to the discrete rovibrational energies of H_2 . We have verified that the computed emission probabilities are in excellent agreement with those reported previously in Wolniewicz et al. (1998) and in Pachucki & Komasa (2011). Here, we extend the computations of the magnetic dipole transitions of Pachucki & Komasa (2011) to all possible $v'J - v''J'$ transitions.

3. Results and discussion

The informations concerning *all* possible transitions within the ground electronic state of H_2 (4711) are given in electronic format. Table 2 provides the first rows of the datafile. In addition to the quantum numbers involved in the transitions, we display the transition wavenumbers σ in cm^{-1} and their theoretical estimated accuracy (see Section 2.1), the wavelengths in micron and the resulting accuracy, the electric quadrupole transition probability A_{qu} , the magnetic dipole transition probability A_{ma} , the sum $A = A_{qu} + A_{ma}$ giving the transition probability of the transition, the inverse of the total radiative lifetime τ of the upper energy level of the transition $A_{tot} = \sum_l A_{u \rightarrow l}$ in s^{-1} . We then report the energy terms of the upper level where the origin 0 is taken for the infinite separation of the two hydrogen atoms and the estimated accuracy in the same units. The next column gives the energy of the upper level expressed in Kelvin, when measured from the ground rovibrational state, as this value is used by astrophysicists when analysing the observed emission spectrum in order to derive excitation temperature, and the last column stands for the statistical weight of the upper level of the transition. We recall that the statistical weight of a particular level is given by $(2J+1) \times g_l$ where $g_l = 1$ for even values of J (para levels) and $g_l = 3$ for odd values of J (ortho levels).

A first comment concerns the wavelengths of the actual transitions taking place within the ground electronic state of H_2 . As an example, Pike et al. (2016) report the (2-1) S(27) transition of H_2 at 2.1790 μ detection towards Herbig-Haro 7. The wavelength is computed from the predictions given in Dabrowski (1984). Former (Komasa et al. 2011) and present calculations show that the energy of the $v=2, J=29$ level is not bound so that the transition is misidentified. Our present computations report a closeby transition at 2.1784 μ corresponding to (8-5) S(14) with a very low emission probability of $2.52 \cdot 10^{-11} \text{ s}^{-1}$, which appears not realistic. The H_2 quadrupole transition wavenumbers and corresponding emission probabilities are also displayed in the HITRAN database (Gordon et al. 2017). We have checked the overall agreement between our computations and those displayed in the HITRAN database. The transition wavenumbers values are restricted to four significant digits in HITRAN whereas the quoted uncertainty reported in the present work is variable and spans an interval between a few 10^{-3} and a few 10^{-6} cm^{-1} , so that some differences in the last digits can be obtained. The number of transitions reported in HITRAN2016 is not fully complete as a result of possible numerical difficulties linked to spline interpolations of the transition moments such as those arising for CO overtone transitions, as discussed in Medvedev et al. (2016). The accuracy of our potential function and electric quadrupole as well as magnetic dipole moments prevents the occurrence of such difficulties.

Table 2. List and properties of rovibrational transitions of H₂. (1) vibrational quantum number of the upper level; (2) rotational quantum number of the upper level; (3) vibrational quantum number of the lower level; (4) rotational quantum number of the lower level; (5) wavenumber of the transition $\nu_u, J_u \rightarrow J_l, \nu_l$; (6) uncertainty on the wavenumber; (7) wavelength of the $\nu_u, J_u \rightarrow J_l, \nu_l$ transition; (8) uncertainty on the wavelength; (9) electric quadrupole transition emission probability; (10) magnetic dipole transition emission probability; (11) full radiative transition emission probability; (12) total radiative decay probability of the upper level; (13) energy of the upper level measured from the dissociation limit of H₂; (14) uncertainty on the energy; (15) energy term of the upper level measured from the ground $v=0, J=0$ level, computed from the dissociation value of 36 118.0695 cm⁻¹ (present calculations); (16) statistical weight of the upper level.

ν_u	J_u	ν_l	J_l	σ	$\Delta\sigma$	λ	$\Delta\lambda$	A_{qu}	A_{ma}	A	A_{tot}	E_u	ΔE_u	E_u	g_u
(1)	(2)	(3)	(4)	(5)	(6)	(7)	(8)	(9)	(10)	(11)	(12)	(13)	(14)	(15)	(16)
0	2	0	0	354.373130	3.7E-06	28.218843793	2.9E-07	2.943E-11	0.000E+00	2.943E-11	2.943E-11	-35763.696396	2.3E-04	509.9	5
0	3	0	1	587.032025	6.1E-06	17.034845756	1.8E-07	4.761E-10	0.000E+00	4.761E-10	4.761E-10	-35412.550688	2.3E-04	1015.1	21
0	4	0	2	814.424302	8.4E-06	12.278611991	1.3E-07	2.755E-09	0.000E+00	2.755E-09	2.755E-09	-34949.272093	2.2E-04	1681.6	9
0	5	0	3	1034.670685	1.1E-05	9.664910918	1.0E-07	9.836E-09	0.000E+00	9.836E-09	9.836E-09	-34377.880003	2.2E-04	2503.7	33
0	6	0	4	1246.099547	1.3E-05	8.025041036	8.4E-08	2.643E-08	0.000E+00	2.643E-08	2.643E-08	-33703.172546	2.1E-04	3474.5	13
0	7	0	5	1447.280936	1.4E-05	6.909508549	6.7E-08	5.879E-08	0.000E+00	5.879E-08	5.879E-08	-32930.599067	2.0E-04	4586.1	45
0	8	0	6	1637.046000	1.6E-05	6.108563840	6.0E-08	1.142E-07	0.000E+00	1.142E-07	1.142E-07	-32066.126547	2.0E-04	5829.8	17
0	9	0	7	1814.492375	1.7E-05	5.511183259	5.2E-08	2.001E-07	0.000E+00	2.001E-07	2.001E-07	-31116.106692	1.9E-04	7196.7	57
0	10	0	8	1978.977263	1.9E-05	5.053115155	4.9E-08	3.236E-07	0.000E+00	3.236E-07	3.236E-07	-30087.149284	1.8E-04	8677.1	21

A second comment concerns the relevance of the magnetic dipole contribution in the Q transitions of H_2 which has been overlooked so far in astrophysical and plasma studies. In order to quantify their possible impact we have plotted both electric quadrupole A_{qu} and magnetic dipole A_{ma} for all Q transitions by separating $\Delta v = 1$, $\Delta v = 2$ and $\Delta v > 2$ transitions. The largest contribution of the magnetic dipole compared to electric quadrupole transition probabilities is obtained for $\Delta v = 1$ transitions and λ above 3.5μ , which involve high J rotational quantum numbers. This trend was already pointed out for the $v = 1 \rightarrow 0$ fundamental band Q transitions by Pachucki & Komasa (2011). The differences can reach more than one order of magnitude. The available spectroscopy measurements of line strengths from Bragg et al. (1982) were restricted to low J values ($J = 1, 2, 3$ of the 1-0 band) and are insensitive to the magnetic dipole contribution. However, the state of the art techniques of intensity determination could allow to further check our derivations. In particular, the Q(6)-Q(10) transitions of the fundamental band show already a 6 to 20 % magnetic dipole contribution to the radiative transition probability, which could be challenged through experiments. Finally, we think that the present computations provide the most accurate H_2 transition wavenumbers, wavelengths, emission probabilities which should be used in the analysis of high temperature plasma and astrophysical conditions.

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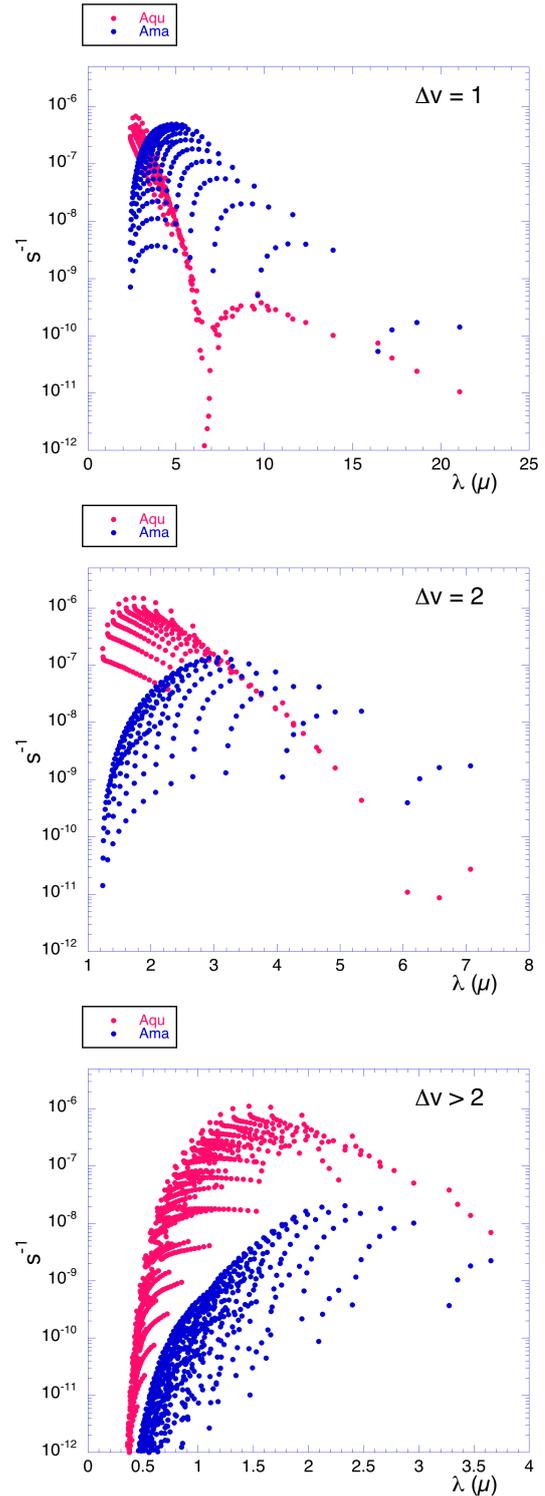


Fig. 1. Electric quadrupole A_{qu} (red) and magnetic dipole A_{ma} (blue) contributions to the transition probabilities in s^{-1} for the Q branches of the H_2 rovibrational spectrum within the ground state for $\Delta v = 1$, $\Delta v = 2$, $\Delta v > 2$.