

LETTER TO THE EDITOR

Space and laboratory observation of the deuterated cyanomethyl radical HDCCN*,**

C. Cabezas¹, Y. Endo², E. Roueff³, N. Marcelino¹, M. Agúndez¹, B. Tercero^{4,5}, and J. Cernicharo¹

- ¹ Grupo de Astrofísica Molecular, Instituto de Física Fundamental (IFF-CSIC), C/ Serrano 121, 28006 Madrid, Spain e-mail: carlos.cabezas@csic.es; jose.cernicharo@csic.es
- ² Department of Applied Chemistry, Science Building II, National Chiao Tung University, 1001 Ta-Hsueh Rd., Hsinchu 30010, Taiwan
- ³ LERMA, Observatoire de Paris, PSL Research University, CNRS, Sorbonne Universités, 92190 Meudon, France
- ⁴ Observatorio Astronómico Nacional (IGN), C/ Alfonso XII, 3, 28014 Madrid, Spain
- ⁵ Centro de Desarrollos Tecnológicos, Observatorio de Yebes (IGN), 19141 Yebes, Guadalajara, Spain

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ABSTRACT

Our observations of TMC-1 with the Yebes 40 m radio telescope in the 31.0-50.3 GHz range allowed us to detect a group of unidentified lines, showing a complex line pattern indicative of an open-shell species. The observed frequencies of these lines and the similarity of the spectral pattern with that of the $2_{0.2}-1_{0.1}$ rotational transition of H_2 CCN indicate that the lines arise from the deuterated cyanomethyl radical, HDCCN. Using Fourier transform microwave spectroscopy experiments combined with electric discharges, we succeeded in producing the radical HDCCN in the laboratory and observed its $1_{0.1}-0_{0.0}$ and $2_{0.2}-1_{0.1}$ rotational transitions. From our observations and assuming a rotational temperature of 5 K, we derive an abundance ratio H_2 CCN/HDCCN = 20 ± 4 . The high abundance of the deuterated form of H_2 CCN is well accounted for by a standard gas-phase model, in which deuteration is driven by deuteron transfer from the H_2 D+ molecular ion.

Key words. astrochemistry - ISM: molecules - ISM: individual objects: TMC-1 - line: identification - molecular data

1. Introduction

Recent developments of radio astronomical equipments allow for very sensitive line surveys of molecular sources, where weak lines arising from low-abundance species can be detected. These surveys are the best tool to map out the molecular makeup of astronomical sources by identifying new chemical species. Discovering spectral features of new molecules initially requires a detailed analysis of the spectral contribution of the already identified species, including their isotopologs and vibrationally excited states. In this manner, once all the lines coming from known species are assigned, unidentified lines emerge and open the opportunity to discover new molecules and to deepen in our understanding of the chemical complexity of molecular clouds.

Deuterated molecules show remarkably high D/H abundance ratios (Ceccarelli et al. 2007) even though the deuterium abundance in space is 1.5×10^{-5} relative to hydrogen (Linsky 2003). Hence, deuterated species of abundant interstellar molecules can also contribute to the spectral richness of line surveys. This fact makes the astronomical identification of these deuterated

species of utmost importance, not only to gain knowledge about unidentified lines in the lines surveys but also to understand how the deuterium fractionation works and ultimately to constrain molecular formation pathways.

Cyanomethyl radical, H_2CCN , has been detected in the cold dark cloud TMC-1 (Saito et al. 1988; Irvine et al. 1988), where its abundance is relatively high. Hence, the deuterated isotopologs of H_2CCN are good candidates to be observed in this source. As has been shown previously, a new high sensitivity line survey on TMC-1 has allowed the identification of several transient species such as the anions C_3N^- and C_5N^- (Cernicharo et al. 2020a) and the protonated molecules HC_5NH^+ (Marcelino et al. 2020), HC_3O^+ (Cernicharo et al. 2020b), CH_3CO^+ (Cernicharo et al. 2021a), and HC_3S^+ (Cernicharo et al. 2021b).

In this Letter, we report the identification of spectral lines of the deuterated species HDCCN in TMC-1 based on ab initio calculations and accurate line frequencies from laboratory experiments. Column densities for H₂CCN and HDCCN species are derived from our observations and the most updated chemical models are used to diagnose the deuterium fractionation ratios for the most abundant molecules in TMC-1.

2. Observations

The observations of TMC-1 ($\alpha_{J2000} = 4^{\rm h}41^{\rm m}41.9^{\rm s}$ and $\delta_{J2000} = +25^{\circ}41'27''.0$) on the Q band presented in this work were performed in several sessions between November 2019 and October 2020. They were carried out using a set of new receivers, which

^{*} Data are only available at the CDS via anonymous ftp to cdsarc.u-strasbg.fr (130.79.128.5) or via http://cdsarc.u-strasbg.fr/viz-bin/cat/J/A+A/646/L1

^{**} Based on observations carried out with the Yebes 40 m telescope (projects 19A003, 20A014, and 20D15) and the Institut de Radioastronomie Millimétrique (IRAM) 30 m telescope. The 40 m radiotelescope at Yebes Observatory is operated by the Spanish Geographic Institute (IGN, Ministerio de Transportes, Movilidad y Agenda Urbana).

were built within the Nanocosmos project¹ and installed at the Yebes 40 m radio telescope (hereafter, Yebes 40 m).

The Q band receiver consists of two HEMT cold amplifiers covering the 31.0–50.3 GHz band with horizontal and vertical polarizations. The temperatures of the receiver vary from 22 K at 32 GHz to 42 K at 50 GHz. The spectrometers formed by 2 × 8 × 2.5 GHz FFTs provide a spectral resolution of 38.1 kHz and cover the whole Q band in both polarizations.

We observed different frequency coverages of $31.08-49.52\,\mathrm{GHz}$ and $31.98-50.42\,\mathrm{GHz}$. This permitted us to check for spurious ghosts produced in the down-conversion chain, in which the signal coming from the receiver was down-converted to $1-19.5\,\mathrm{GHz}$ and then split into eight bands with a coverage of $2.5\,\mathrm{GHz}$, each of which were analyzed by the FFTs. Frequency switching with a frequency throw of $10\,\mathrm{MHz}$ or $8\,\mathrm{MHz}$ was chosen as the observing procedure. The nominal spectral resolution of $38.1\,\mathrm{kHz}$ was left unchanged for the final spectra. The sensitivity, σ , was derived by removing a polynomial baseline in velocity windows of $24\,\mathrm{km\,s^{-1}}$ wide, centered on each observed line, and varied along the Q band between $\sim 0.4\,\mathrm{mK}$ ($31\,\mathrm{GHz}$), $\sim 1.0\,\mathrm{mK}$ ($43\,\mathrm{GHz}$), and $\sim 2.5\,\mathrm{mK}$ ($50\,\mathrm{GHz}$).

The intensity scale, antenna temperature $(T_{\rm A}^*)$ used in this work was calibrated using two absorbers at different temperatures and the atmospheric transmission model ATM (Cernicharo 1985; Pardo et al. 2001). Calibration uncertainties were adopted to be 10% based on the observed repeatability of the line intensities between different observing runs. All data were analyzed using the GILDAS package².

3. Results

3.1. TMC-1 observations

Among the unidentified lines of the line survey carried out by our group toward TMC-1 using the Yebes 40 m radio telescope, we found a group of lines with a spectral pattern reminiscent of that produced by a radical with a complex hyperfine structure (see Fig. 1a). The pattern is very similar to the $2_{0,2}-1_{0,1}$ rotational transition of H₂CCN (see Fig. 1b), which lies at a slightly higher frequency. These facts suggest that the group of unidentified lines could arise from the deuterated species of cyanomethyl radical. In order to check this hypothesis we performed ab initio calculations (see Appendix A) for HDCCN using the spectroscopic available information on H₂CCN (Saito & Yamamoto 1997; Ozeki et al. 2004) to scale the spectroscopic parameters calculated. The frequency of the $2_{0,2}$ – $1_{0,1}$ transition of HDCCN was predicted in the same frequency region where the group of unidentified lines appear. The hyperfine structure of this transition of HDCCN is slightly different from that of H₂CCN resulting from the replacement of a hydrogen nucleus by a deuteron. Motivated by the good agreement between calculated and observed frequency and spectral pattern, we performed additional ab initio calculations and laboratory experiments to assign the complicated hyperfine structure of the rotational spectrum of HDCCN (see Sect. 3.2 and Appendix A). These experiments fully confirm that the carrier of the lines shown in Fig. 1a is HDCCN.

3.2. Laboratory measurements

The HDCCN radical was produced in a supersonic jet by a pulsed electric discharge of a gas mixture of CH₃CN (0.2%)

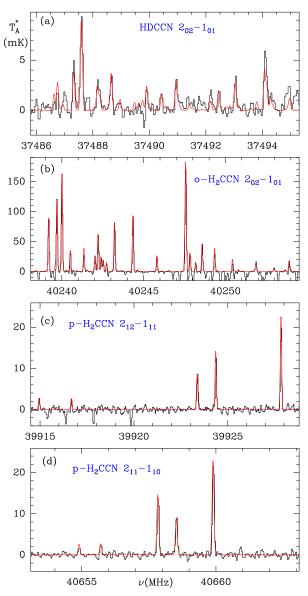


Fig. 1. (a) Observed group of lines assigned to the hyperfine structure of the $2_{0,2}-1_{0,1}$ transition of HDCCN (black histogram). (b) The observed hyperfine structure of the same transition of o-H₂CCN. (c) Selected hyperfine components of the $2_{12}-1_{11}$ transition of p-H₂CCN. (d) Selected hyperfine components of the $2_{11}-1_{10}$ transition of p-H₂CCN. In all panels, the abscissa represents the rest frequency (in MHz) assuming a local standard of rest velocity for the source of $5.83 \,\mathrm{km \, s^{-1}}$ (Cernicharo et al. 2020c). The ordinate is the antenna temperature in mK. The red curve in each panel represents the best model fit to the emission of this species in TMC-1 for a rotational temperature of $10 \,\mathrm{K}$. A similar fit is obtained for other rotational temperatures (see Sect. 4.1 and Table 2). Negative features are due to the folding of the frequency-switching data.

and D₂ (5%) diluted in Ar. The gas mixture was expanded into a Fabry-Perot cavity of a Fourier transform microwave (FTMW) spectrometer (Endo et al. 1994; Cabezas et al. 2016) with a backing pressure at 2 atm. Synchronized to the gas expansion, a pulse voltage of 1.0 kV with a duration of 450 μs was applied between stainless steel electrodes attached to the exit of the pulsed discharge nozzle (PDN). For measurements of the paramagnetic lines, the Earth's magnetic field was canceled using three sets of Helmholtz coils placed perpendicularly to one another. Since the PDN is arranged parallel to the cavity of the

https://nanocosmos.iff.csic.es/

http://www.iram.fr/IRAMFR/GILDAS

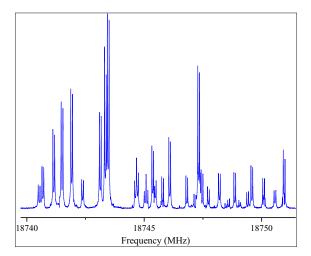


Fig. 2. Section of 10 MHz of the FTMW spectrum of HDCCN showing 36 hyperfine components of the $N_{K_a,K_c} = 1_{0,1} - 0_{0,0}$ pure rotational transition. The spectrum was achieved by 100 shots of accumulation and a step scan of 0.5 MHz with a repetition rate of 10 Hz. The coaxial arrangement of the adiabatic expansion and resonator axis produces an instrumental Doppler doubling. The resonances frequencies are calculated as the average of the two Doppler components.

spectrometer, it is possible to resolve small hyperfine splittings by suppressing the Doppler broadening of spectral lines.

The $N_{K_a,K_c} = 1_{0,1}-0_{0,0}$ pure rotational transition was observed by FTMW spectroscopy while the $N_{K_a,K_c} = 2_{0,2}-1_{0,1}$ transition was measured utilizing a MW-MW double-resonance (DR) technique (Sumiyoshi et al. 2005). In DR measurements, the signal of a known rotational transition is monitored with ordinary FTMW spectroscopy and when the pump radiation is resonant to a transition, connecting the upper or lower level of the monitored microwave transition, a DR spectrum is observed as a change of the intensity of the monitored microwave signal. First, a spectral scan of 100 MHz was done around the predicted frequency region for the $N_{K_a,K_c} = 1_{0,1} - 0_{0,0}$ pure rotational transition to observe all the expected hyperfine components. The recorded spectrum for $N_{K_a,K_c} = 1_{0,1} - 0_{0,0}$ transition is shown in Fig. 2, where a total of 66 hyperfine components were observed. Other 63 hyperfine components for the $N_{K_a,K_c} = 2_{0,2}-1_{0,1}$ rotational transition could be measured using the DR technique. Hence, a total of 129 hyperfine components (see Appendix B and Table B.1) were analyzed using a Hamiltonian with the following form:

$$H = H_{\text{rot}} + H_{\text{sr}} + H_{\text{mhf}} + H_{\text{Q}},\tag{1}$$

where $H_{\rm rot}$ contains rotational and centrifugal distortion parameters, $H_{\rm sr}$ is the spin-rotational term, $H_{\rm mhf}$ represents the magnetic hyperfine coupling interaction term due to the nitrogen, hydrogen, and deuterium nuclei, and finally H_Q represents the nuclear electric quadrupole interaction due to nitrogen and deuterium nuclei. The coupling scheme used is J = N + S, $F_1 = J + I(N)$, $F_2 = F_1 + I(H)$ and $F = F_2 + I(D)$, and thus each energy level is denoted by seven quantum numbers; N, J, K_a , K_c , F_1 , F_2 , and F. The results obtained from the least-squares analysis are shown in Table 1. We floated the (B + C)/2 parameter, Δ_N centrifugal constant, ε_{bb} spin-rotation coupling constant, a_F and T_{aa} magnetic hyperfine constants for nitrogen, hydrogen, and deuterium nuclei, and the χ_{aa} nuclear electric quadrupole constant for nitrogen nucleus. The A and (B - C)/2 rotational parameters, ε_{aa} and ε_{cc} spin-rotation coupling constants, and the χ_{aa} and χ_{bb} nuclear electric quadrupole constants for deuterium nucleus

Table 1. Spectroscopic parameters for HDCCN (all in MHz).

De	etermined (a)	Scale	ed (b)	Fixed (c)		
Param.	Value	Param.	Value	Param.	Value	
(B+C)/2	9373.36479(35) ^(d)	A	195819.2	Δ_{NK}	0.41626	
Δ_N	0.004379(48)	(B-C)/2	230.594	Δ_K	13.6709	
$oldsymbol{arepsilon}_{bb}$	-22.6229(16)	$oldsymbol{arepsilon}_{aa}$	-454.559	δ_J	0.0001611	
$a_F(N)$	9.4965(11)	$oldsymbol{arepsilon}_{cc}$	-1.883	δ_K	0.263	
$T_{aa}(N)$	-15.6405(28)	$T_{bb}(D)$	1.9721	Φ_{KN}	-0.000719	
$\chi_{aa}(N)$	-4.1906(30)	$\chi_{aa}(D)$	-0.010	ϕ_K	-0.0024	
$a_F(H)$	-59.8339(41)	$\chi_{bb}(D)$	0.105	$^{S}\Delta_{KN}$	0.0105	
$T_{aa}(H)$	-17.7902(25)			$^{S}\Delta_{K}$	0.14	
$a_F(D)$	-9.1500(10)			$T_{bb}(N)$	-12.4519	
$T_{aa}(D)$	-2.1833(22)			$C_{aa}(N)$	0.0273	
				$C_{bb}(N)$	0.0029	
				$\chi_{bb}(N)$	1.8327	
				$T_{bb}(H)$	14.6	

Notes. ^(a)This work. ^(b)Fixed to the scaled values calculated in this work (see text). ^(c)Fixed at the values obtained by Saito & Yamamoto (1997) and Ozeki et al. (2004). ^(d)Numbers in parentheses represent the derived uncertainty (1 σ) of the parameter in units of the last digit.

were kept fixed to the theoretical values calculated in this work. The rest of parameters were fixed to the values obtained by Saito & Yamamoto (1997) and Ozeki et al. (2004) for H₂CCN radical. The standard deviation of the fit is 4.1 kHz.

4. Discussion

4.1. H₂CCN and HDCCN column densities

The N=2-1 line of $\rm H_2CCN$ presents a very complex hyperfine structure in addition to the presence of two spin symmetry species, ortho and para. Figure 1b shows the 2_{02} – 1_{01} ortho line; the ortho species corresponds to even values of the quantum number K_a . Figures 1c and d show a selected range of hyperfine components of the $N_{K_a,K_c}=2_{1,2}$ – $1_{1,1}$ and $2_{1,1}$ – $1_{1,0}$ para lines (K_a odd). Observed line intensities for the ortho species agree very well with those reported by Kaifu et al. (2004). No lines of the para species are reported in that work, which probably results from the limited sensitivity in their spectra. However, some of their unidentified features agree in frequency with the strongest components of this species (see also Ozeki et al. 2004). The detection of all hyperfine components above 1.5 mK of the two para transitions within our line survey confirms the presence of the para species in TMC-1.

Frequency predictions for the ortho and para species of $\rm H_2CCN$ are from the CDMS catalog (Müller et al. 2005), which were implemented in the MADEX code (Cernicharo 2012) to compute column densities for these species. Both species have to be considered separately as there are no radiative or collisional transitions between them. The lowest energy level of the para species $(1_{1,1})$ is 14.1 K above the ortho ground level (0_{00}) .

In our line survey we only observed one rotational transition of the ortho H_2CCN ($2_{0,2}-1_{0,1}$) and two transitions for the para species ($2_{1,2}-1_{1,1}$ and $2_{1,1}-1_{1,0}$), which involve levels with similar energies. Taking into account that collisional rates are not available for this molecule, column density determinations rely on the assumption of a rotational temperature, T_r , for the two species. Similar assumptions have been adopted in previous studies of this molecule (Irvine et al. 1988; Vastel et al. 2015). In TMC-1, the rotational temperature found for different

molecules varies between 5 K and 10 K (see, e.g., Gratier et al. 2016; Cernicharo et al. 2021b, 2020c; Marcelino et al. 2020, and references therein). In particular, for H_2CCN , Gratier et al. (2016) find a rotational temperature between 3 K and 5 K. We consider that the lowest value for T_r implies a large opacity for the N=2-1 line of the ortho species (see Fig. 1). The ortho species of the molecule behaves, for low rotational temperatures, as a linear rotor because only levels with $K_a=0$ are populated in cold clouds. The first $K_a=2$ level (2₂₀) is at 55.6 K above the ground level (0_{0,0}). The situation is similar for the para species because only levels with $K_a=1$ are populated. The first $K_a=3$ level (3_{3,0}) is 110.3 K above the ground level (1_{1,1}). Hence, the rotational partition function, Q_r , for both symmetry species has a nearly linear dependence on the rotational temperature ($T_r \le 30$ K).

We computed the ortho and para column densities that best fit the observed transitions shown in Figs. 1b-d and for rotational temperatures between 5 K and 10 K. In these figures the red curve represents the synthetic model spectrum computed at $T_{\rm r} = 10 \,\rm K$. The derived total column density of H₂CCN shows a very smooth variation between 5 K and 10 K. This is due to the near linear behavior of the partition function with $T_{\rm r}$ and the low energy of the involved levels. Therefore the synthetic model spectrum is practically the same for these rotational temperatures. Adopting a value for $T_r = 5 \text{ K}$ (see Appendix C), we derive $N(\text{H}_2\text{CCN}) = (1.48 \pm 0.15) \times 10^{13} \text{ cm}^{-2}$ and an ortho/para ratio of 7.7, which is much larger than the statistical ratio of 3. It seems, hence, that the para species, which is higher in energy than the ortho, has suffered interconversion processes through the interaction with XH+ species such as H₃+, HCO+, or H₃O⁺. Our derived column density is in good agreement with that derived by Ozeki et al. (2004) and Gratier et al. (2016).

The rotational partition function shows a dependence on $T_{\rm r}^{3/2}$ for HDCCN, and hence the column density shows a larger dependence on the assumed $T_{\rm r}$. Table 2 gives the derived column densities of $o\text{-H}_2\text{CCN}$, $p\text{-H}_2\text{CCN}$, and HDCCN for rotational temperatures between 5 K and 10 K. We derive $N(\text{HDCCN}) = (7.4 \pm 0.7) \times 10^{11} \, \text{cm}^{-2}$ and $N(\text{H}_2\text{CCN})/N(\text{HDCCN}) = 20 \pm 4$ for $T_{\rm r} = 5$ K. For $T_{\rm r} = 10$ K, these values are $(10.5 \pm 1.1) \times 10^{11} \, \text{cm}^{-2}$ and 14 ± 3 , respectively. The derived XH/XD abundance ratios for H₂CCN and other species are given in Table 3 (see Appendix C and Table C.1).

The H₂CCN ortho/para ratio was also derived in L1544 by Vastel et al. (2015). However, the derived values for the column density of the ortho and para species and their abundance ratio show a strong dependency on the assumed rotational temperature. The ortho/para ratio for $T_r = 5 \,\mathrm{K}$ is 0.1 and the column density for the para species is ten times higher than for the ortho species. However, for $T_r = 10 \,\mathrm{K}$, the column density of the para species decreases by two orders of magnitude and the ortho/para ratio increases to a value of 1. We re-examined their column densities using the data shown in their Fig. 1. Considering, as we have for TMC-1, that the ortho and para spin symmetry states are two different species, we derive for this cloud $N(o\text{-H}_2\text{CCN}) = 4 \times 10^{12} \,\mathrm{cm}^{-2}$ and $N(p\text{-H}_2\text{CCN}) = 5 \times 10^{11} \,\mathrm{cm}^{-2}$. Hence, the ortho/para ratio in L1544 is ~8, that is, very similar to that found in TMC-1. Similar results are obtained for $T_r = 10 \,\mathrm{K}$.

4.2. Chemical models

After the surprise raised by the detection of DCN in the Orion cloud by Jefferts et al. (1973), it was very quickly suggested that the observed deuterium enrichment is the result of chemical fractionation and not of an exceptional source of deuterium through nucleosynthetic processes. The first determinant step is

Table 2. H₂CCN and HDCCN column densities in TMC-1.

Parameter/T _{rot}	5 K	6 K	7 K	8 K	9 K	10 K
N(o-H ₂ CCN) (a)	131	121	118	120	124	128
$N(p-H_2CCN)^{(a)}$	17	17	17	17	18	19
N(H ₂ CCN) (a)	148	138	135	137	142	147
$N(o-H_2CCN)/N(p-H_2CCN)$	7.7	7.1	7.0	7.1	6.9	6.7
N(HDCCN) (a)	7.4	7.5	8.0	8.8	9.6	10.5
N(H ₂ CCN)/N(HDCCN)	20.0	18.4	16.9	15.6	14.8	14.0

Notes. ^(a)Column densities are in units of 10^{11} cm⁻². The uncertainty on the derived column densities is ~10% based on the adopted calibration accuracy (see Sect. 2). The adopted line width for all features of H₂CCN and HDCCN is $0.6 \, \mathrm{km \, s^{-1}}$.

Table 3. Deuteration enhancement in TMC-1 for selected molecules compared to our gas-phase chemical model.

Molecule	TMC-1	Model ^(a) N/O = 0.5	Model ^(a) N/O = 1	Model ^(a) N/O=1.5
H ₂ CCN/HDCCN	20 (b)	17.6	16.2	15.2
HC ₃ N/DC ₃ N (c)	62	78.8	75.3	73.0
HNCCC/DNCCC (c)	43	34.8	35.9	37.2
HCCNC/DCCNC (c)	30	71.7	68.3	65.9
HCNCC/DCNCC (d)		21.2	20.9	20.6
HC ₅ N/DC ₅ N (c)	82	27.8	28.3	29.4
CH ₃ CCH/CH ₃ CCD	49	204.6	210.	213.3
CH ₃ CCH/CH ₂ DCCH	10 ^(e)	68.2	70.1	71.1
CH ₃ CN/CH ₂ DCN	11 ^(e)	10.2	9.9	9.7
c - C_3H_2/c - C_3HD	$27^{(b)}$	45.5	45.1	45.1
H ₂ C ₄ /HDC ₄	83 ^(b)	114.4	113.6	114.4
C ₄ H/C ₄ D	118	104.3	107.1	109.5

Notes. ^(a)See text for the assumed physical conditions. $[O/H] = 8 \times 10^{-6}$, $D/H = 1.5 \times 10^{-5}$, and C/O = 0.75. ^(b)The elemental H/D abundance ratio is affected by the presence of two identical nuclei of H. ^(c)These H/D abundance ratios are from Cernicharo et al. (2020c). ^(d)This isomer of HC₃N has not yet been detected in space. ^(e)The elemental H/D abundance ratio is affected by the presence of three identical nuclei of H.

provided by the deuterium exchange in the $H_3^+ + HD \rightleftharpoons H_2D^+$ + H₂ reaction. This reaction proceeds essentially in the forward direction in the cold $(T \sim 10 \,\mathrm{K})$ interstellar environments, as the endothermicity of the reverse reaction is about 232 K, which results from the difference in zero point energies (ZPE) of the products and reactants. This crucial point was first mentioned by Watson (1976) and extended to the $D + H_3^+ \rightleftharpoons H + H_2D^+$ reaction by Dalgarno & Lepp (1984), which proceeds essentially in the forward direction as the endothermicity of the reverse reaction is then 632 K. The H₂D⁺ molecular ion subsequently transfers the deuteron to various stable molecules, such as CO, N_2 , H_2O , NH₃, producing DCO⁺, N₂D⁺, H₂DO⁺, NH₃D⁺, which disassociatively recombine to produce, for example, HDO and NH₂D. This scenario has been further extended to a few other stable molecular ions such as CH_3^+ and $C_2H_3^+$, invoking exchange reactions with HD as in CH_3^+ + HD \rightleftarrows CH_2D^+ + H₂ and $C_2H_3^+$ + HD \rightleftarrows $C_2H_2D^+$ + H₂ (Herbst et al. 1987; Gerlich & Schlemmer 2002). The exothermicity achieved in the forward reactions are somewhat larger than that involving H₂D⁺ (Herbst et al. 1987; Roueff et al. 2013; Nyman & Yu 2019), allowing these exchange reactions to occur in somewhat warmer environments such as clumps in the Orion Bar (Roueff et al. 2007; Parise et al. 2009). These exchange reactions can even proceed further toward doubly deuterated molecular ions like D₂H⁺ and CHD_2^+ , and up to fully deuterated ions like D_3^+ , CD_3^+ , and $C_2D_3^+$ in environments with CO depleted (Walmsley et al. 2004), where

the ion-molecule reactions with CO, H₂O, and NH₃ become negligible in comparison to the reactions with HD (see, e.g., Roberts et al. 2003; Roueff et al. 2005). Additional deuterium enhancements could result from the evaporation of the icy mantles surrounding dust grain as a remnant of cold grain chemistry. Gas-phase chemical processes are reasonably successful to interpret the observations, as reported recently by Bacmann et al. (2020) and Melosso et al. (2020) in their study of the nitrogen hydrides NHD and ND₂.

We introduced the deuteration of the carbon chains C_nH , C_nH₂, and nitriles in our gas-phase chemical network, which has been used to study the influence of the C/O ratio on the chemistry of nitriles by Le Gal et al. (2019). That study benefitted from the recent updates reported by Loison et al. (2014, 2017) for nitrogen and carbon-chain chemistry, including the CH₃CN isomers. We did not introduce an additional deuterium exchange reaction in the network because the various exchange reactions with HD that we considered (HCNH⁺, HC₃NH⁺, CH₃CNH⁺) are all hindered by the presence of a barrier of about 56 kcal mol⁻¹ (see Appendix A). Then, deuterium fractionation results directly from the enrichment of H₂D⁺ and, to a lesser extent, from CH₂D⁺ and C₂H₂D⁺. In the case of HDCCN, the main route to deuteration is provided by the reaction $H_2CCN + H_2D^+ \rightarrow CH_2DCN^+ + H_2$, followed by $CH_2DCN^+ + e^- \rightarrow HDCCN + H$. The various dissociative channels of that reaction are reported in Loison et al. (2014) for the hydrogenated ions. We assume, as in our previous studies on nitrogen deuteration (Roueff et al. 2005), that in absence of experimental information the total rate coefficient of dissociative recombination of the deuterated ion is identical to that of the hydrogenated molecular ion. We hypothesize that hydrogen ejection is twice as efficient as deuterium ejection, as found experimentally for HDO⁺ (Jensen et al. 2000). Our final gas-phase chemical network includes 288 chemical species and more than 7000 chemical reactions³. More details on our gas-phase chemical network will be described in a later publication. We ran steady-state models that are appropriate for the TMC-1 conditions, that is, $n(H_2) = 10^4 \text{ cm}^{-3}$, T = 10 K, $\zeta = 1.3 \times 10^{-17} \,\mathrm{s}^{-1}$, and deplete oxygen to get the observed fractional abundance of CO relative to molecular hydrogen of $\sim 1.2 \times 10^{-5}$ (Cernicharo & Guélin 1987). The results are shown in Table 3 and compared to the observational ratios. The deuterium ratio of H₂CCN is found to be remarkably stable, independent of the assumed N/O ratio and close to the observational value. The other deuterium ratios are within about a factor of two of the observed values, with a couple of exceptions, and are not very sensitive to the N/O ratio. The HCCNC isomer is found by the observations to be more fractionated than theoretically predicted, which may indicate specific chemical rearrangements in the concerned chemical reactions. The main disagreement is obtained for deuterated methyl acetylene, where some additional deuterium enrichment mechanism could be missing. Further studies are desirable.

5. Conclusions

We detected, and fully characterized spectroscopically, the singly deuterated isotopolog of H_2CCN from astronomical and laboratory measurements. The observed deuterium ratio for this and other species is well accounted for by a standard gas-phase model, in which deuteration is transferred through the H_2D^+ molecular ion, followed by dissociative recombination.

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³ The initial network (Le Gal et al. 2019) comprised 190 chemical species and about 3500 chemical reactions.

Appendix A: Quantum chemical calculation

Quantum chemical calculations were carried out to estimate the molecular parameters of HDCCN. Very precise values for the rotational parameters of HDCCN can be obtained using experimetal/theoretical ratios derived for H₂CCN species. This is the most common method to predict the expected experimental rotational constants for an isotopic species of a given molecule when the rotational constants for its parent species are known. The geometry optimization calculations were done using a coupled cluster with single, double, and perturbative triple excitation (CCSD(T); Raghavachari et al. 1989) methods with all electrons (valence and core) correlated and Dunning's correlation-consistent basis sets with polarized core-valence correlation quadruple-ζ (cc-pCVQZ; Woon & Dunning 1995). This calculations were carried out using the Molpro 2018.1 program (Werner et al. 2018). Table A.1 shows the results for these calculations. In addition to the rotational constants, we calculated other parameters necessary to interpret the rotational spectrum of HDCCN. In this manner, the three spin-rotation coupling constants (ε_{aa} , ε_{bb} and ε_{cc}), the magnetic hyperfine constants (a_F , T_{aa} and T_{bb}), and the nuclear electric quadrupole constants for the deuterium nucleus (χ_{aa} and χ_{bb}) were estimated following the same procedure used for the rotational constants. However, in this case the optimized molecular structure at CCSD(T)/ccpCVQZ was used. These calculations were carried out via the Becke, three-parameter, Lee-Yang-Parr (B3LYP) hybrid density functional and Dunning's correlation-consistent basis sets with correlation-consistent polarized valence quadruple-ζ (cc-pVQZ; Dunning 1989). The results for these calculations carried out using the Gaussian16 (Frisch et al. 2016) program package are shown in Table A.1.

In addition to the structural optimizations, quantum chemical calculations were carried out to examine the potential energy surface for the deuteration reactions of HCNH⁺, HC₃NH⁺, CH₃CNH⁺ with HD. All the possible stationary points were fully optimized using the coupled cluster singles and doubles correlated method (CCSD; Cížek et al. 1969) along with Dunning's correlation-consistent polarized valence triple-ζ (cc-pVTZ; Dunning 1989) basis set. The starting point, with the cationic species (HCNH⁺, HC₃NH⁺ or CH₃CNH⁺) and HD separated from each other, was assumed to be energy zero. The reactions proceed initially by the formation of a pre-reactive complex (HCNH⁺-HD, HC₃NH⁺-HD, and CH₃CNH⁺-HD), which have energies 2.23, 2.09, and 2.85 kcal mol⁻¹ lower than

Table A.1. Experimental, theoretical, and scaled values for the spectroscopic parameters of H₂CCN and HDCCN(all in MHz).

	$_{}$ H_2	CCN	HDCCN			
Parameter	Calc. (a)	Exp. (b)	Calc. (a)	Scaled (c)		
\overline{A}	287050.9	284981.	197241.4799	195819.1		
B	10252.6	10246.7658	9608.3332	9602.3		
C	9899.0	9876.029	9162.0187	9141.2		
$\overline{oldsymbol{arepsilon}_{aa}}$	-776.794	-661.537	-533.755	-454.558		
$arepsilon_{bb}$	-27.537	-24.1205	-25.807	-22.605		
$arepsilon_{cc}$	1.059	-2.0345	0.980	-1.883		
$\overline{a_F(\mathrm{H})}$	-54.9478	-59.63				
T_{aa} (H)	-18.2057	-15.9006				
T_{bb} (H)	17.7382	14.6				
$\overline{a_F(D)}$			-8.434	-9.153		
T_{aa} (D)			-2.219	-2.156		
T_{bb} (D)			2.147	1.972		
χ_{aa} (N)	-4.195	-4.7705				
χ_{bb} (N)	1.8327	2.0341				
χ_{aa} (D)			-0.0111	-0.0097		
χ_{bb} (D)			0.0165	0.1048		

Notes. ^(a)Rotational constants calculated at CCSD(T)/cc-pCVQZ level of theory and the rest of parameters at B3LYP/cc-pVQZ level of theory. ^(b)Saito & Yamamoto (1997) and Ozeki et al. (2004). ^(c)Scaled by the ratio Exp/Calc. of the corresponding parameter for the H₂CCN species

the separated reactants for HCNH $^+$, HC₃NH $^+$ and CH₃CNH $^+$, respectively. The barrier heights of the transition states between the pre-reactive complexes and the deuterated species (HCND $^+$, HC₃ND $^+$, CH₃CND $^+$) are 56.42, 51.87, and 52.76 kcal mol $^{-1}$, respectively.

Appendix B: Laboratory transition frequencies for HDCCN

The laboratory measurements of HDCCN described in Sect. 3.2 permitted us to measure 132 hyperfine components in the $N_{K_a,K_c} = 2_{0,2}-1_{0,1}$ and $1_{0,1}-0_{0,0}$ rotational transitions. The observed frequencies, quantum number assignments, and calculated line intensities are given in Table B.1. We uploaded a file with predictions up to N = 16 to the CDS.

 Table B.1. Laboratory observed transition frequencies for HDCCN.

	<i>K'</i> _a	<i>K'</i> _c	J'	F_1'	F_2'	F'	N''	<i>K</i> '' _a	K'' _c	J''	F'' ₁	$F_2^{\prime\prime}$	F''	ν _{obs} (MHz)	ν _{calc} (MHz)	Obs-Calc (MHz)	Int. (a)
1	0	1	0.5	1.5	1	2	0	0	0	0.5	1.5	1	2	18708.037	18708.035	0.002	0.065
1	0	1	1.5	1.5	2	1	0	0	0	0.5	1.5	2	1	18729.487	18729.484	0.003	0.068
1	0	1	1.5	1.5	2	3	0	0	0	0.5	1.5	2	3	18733.323	18733.327	-0.004	0.460
1	0	1	1.5	1.5	2	2	0	0	0	0.5	1.5	2	2	18734.829	18734.828	0.001	0.464
1 1	0 0	1 1	1.5 1.5	0.5 0.5	1 1	1 1	0	0	0	0.5 0.5	0.5 0.5	1 1	1 0	18735.210 18735.956	18735.212 18735.958	-0.002 -0.002	0.093 0.106
1	0	1	1.5	1.5	1	2	0	0	0	0.5	1.5	2	3	18736.776	18736.776	0.002	0.100
1	0	1	1.5	0.5	1	2	0	0	0	0.5	0.5	1	2	18737.107	18737.113	-0.006	0.307
1	0	1	1.5	1.5	1	0	0	0	0	0.5	0.5	1	1	18738.806	18738.809	-0.003	0.163
1	0	1	1.5	1.5	1	2	0	0	0	0.5	0.5	1	1	18739.142	18739.144	-0.002	0.220
1	0	1	0.5	1.5	2	1	0	0	0	0.5	1.5	2	1	18740.523	18740.520	0.003	0.440
1	0	1	0.5	1.5	2	2	0	0	0	0.5	1.5	2	2	18740.684	18740.681	0.003	0.669
1	0	1	1.5	0.5	1	2	0	0	0	0.5	0.5	0	1	18741.148	18741.154	-0.006	1.347
1 1	0	1 1	1.5 0.5	0.5 1.5	1 2	1	0	0	0	0.5 0.5	0.5 1.5	1 2	2	18741.432 18741.500	18741.432 18741.497	0.000 0.003	0.144 1.866
1	0	1	1.5	1.5	2	3	0	0	0	0.5	0.5	1	2	18741.910	18741.497	-0.003	1.865
1	0	1	1.5	0.5	0	1	0	0	0	0.5	0.5	1	1	18742.375	18742.376	-0.001	0.564
1	0	1	1.5	2.5	3	2	0	0	0	0.5	1.5	2	1	18743.128	18743.128	0.000	1.509
1	0	1	1.5	2.5	3	3	0	0	0	0.5	1.5	2	2	18743.343	18743.343	0.000	2.320
1	0	1	1.5	2.5	3	4	0	0	0	0.5	1.5	2	3	18743.468	18743.469	-0.001	3.000
1	0	1	1.5	1.5	2	1	0	0	0	0.5	0.5	1	0	18744.633	18744.631	0.002	0.699
1	0	1	1.5	1.5	2	2	0	0	0	0.5	0.5	1	1	18744.711	18744.713	-0.002	0.888
1	0	1 1	0.5 1.5	1.5 2.5	2 2	1 1	0	0	0	0.5 0.5	1.5 1.5	2	2	18745.042	18745.037 18745.116	0.005 0.002	0.363 0.261
1 1	0	1	1.5	1.5	1	2	0	0	0	0.5	0.5	1	2	18745.118 18745.363	18745.110	-0.002	1.079
1	0	1	1.5	0.5	1	1	0	0	0	0.5	0.5	0	1	18745.471	18745.474	-0.003	0.645
1	0	1	1.5	0.5	1	0	0	0	0	0.5	0.5	1	1	18745.751	18745.760	-0.009	0.129
1	0	1	1.5	2.5	2	1	0	0	0	0.5	1.5	1	1	18745.785	18745.783	0.002	0.500
1	0	1	1.5	2.5	2	2	0	0	0	0.5	1.5	1	1	18746.086	18746.083	0.003	1.017
1	0	1	0.5	0.5	1	1	0	0	0	0.5	1.5	1	0	18746.812	18746.811	0.001	0.480
1	0	1	1.5	2.5	2	1	0	0	0	0.5	1.5	1	2	18747.164	18747.160	0.004	0.209
1	0	1	1.5	2.5	2	3	0	0	0	0.5	1.5	1	2	18747.320	18747.318	0.002	2.316
1 1	0	1 1	0.5 1.5	0.5 2.5	1 2	0 2	0	0	0	0.5 0.5	1.5 1.5	1 1	1 2	18747.339 18747.462	18747.335 18747.460	0.004 0.002	0.308 0.624
1	0	1	1.5	1.5	1	1	0	0	0	0.5	0.5	1	1	18747.734	18747.733	0.002	0.024
1	0	1	0.5	1.5	2	2	0	0	0	0.5	1.5	2	3	18748.202	18748.199	0.003	0.554
1	0	1	1.5	0.5	0	1	0	0	0	0.5	0.5	1	2	18748.595	18748.596	-0.001	0.124
1	0	1	0.5	0.5	1	1	0	0	0	0.5	1.5	1	2	18748.856	18748.855	0.001	0.455
1	0	1	1.5	1.5	1	0	0	0	0	0.5	0.5	0	1	18749.069	18749.070	-0.001	0.164
1	0	1	1.5	1.5	1	2	0	0	0	0.5	0.5	0	1	18749.404	18749.405	-0.001	0.225
1	0	1	0.5	0.5	1	2	0	0	0	0.5	1.5	1	1	18749.583	18749.582 18750.085	0.001	0.623 0.454
1 1	0 0	1 1	0.5 1.5	1.5 1.5	2 2	3 1	0	0	0 0	0.5 0.5	0.5 0.5	1 1	2 2	18750.085 18750.111	18750.085	0.000 0.005	0.434
1	0	1	0.5	1.5	2	2	0	0	0	0.5	0.5	1	1	18750.568	18750.567	0.003	0.401
1	0	1	1.5	1.5	2	2	0	0	0	0.5	0.5	1	2	18750.923	18750.933	-0.010	0.191
1	0	1	0.5	0.5	1	2	0	0	0	0.5	1.5	1	2	18750.959	18750.959	0.000	0.935
1	0	1	1.5	0.5	0	1	0	0	0	0.5	0.5	0	1	18752.636	18752.637	-0.001	0.224
1	0	1	1.5	1.5	1	1	0	0	0	0.5	0.5	1	2	18753.952	18753.954	-0.002	0.473
1	0	1	0.5	0.5	0	1	0	0	0	0.5	1.5	1	0	18753.986	18753.983	0.003	0.234
1	0	1	0.5	0.5	0	1	0	0	0	0.5	1.5	1	1	18754.648	18754.650	-0.002	0.462 0.093
1 1	0	1 1	0.5 1.5	1.5 1.5	2 2	1 2	0	0	0	0.5 0.5	0.5 0.5	1	1 1	18754.925 18754.971	18754.923 18754.975	0.002 -0.004	0.093
1	0	1	1.5	2.5	3	2	0	0	0	0.5	1.5	2	3	18755.163	18755.163	0.004	0.085
1	0	1	0.5	1.5	2	1	0	0	0	0.5	0.5	1	0	18755.670	18755.668	0.002	0.090
1	0	1	0.5	0.5	0	1	0	0	0	0.5	1.5	1	2	18756.024	18756.027	-0.003	0.264
1	0	1	1.5	0.5	1	0	0	0	0	0.5	0.5	0	1	18756.024	18756.021	0.003	0.152
1	0	1	0.5	1.5	1	2	0	0	0	0.5	1.5	2	1	18756.391	18756.390	0.001	0.116
1	0	1	1.5	1.5	1	1	0	0	0	0.5	0.5	0	1	18757.992	18757.995	-0.003	0.071
1	0	1	0.5	1.5	1	2	0	0	0	0.5	1.5	2	2	18760.907	18760.906	0.001	0.427
1	0 0	1 1	0.5 0.5	1.5 1.5	1 1	1	0	0	0 0	0.5 0.5	1.5 1.5	2 2	1 1	18762.014 18764.270	18762.013 18764.269	0.001 0.001	0.320 0.258
1 1	0	1	0.5	1.5	1	1	0	0	0	0.5	1.5	2	2	18766.530	18766.530	0.001	0.238
1	0	1	0.5	1.5	1	2	0	0	0	0.5	1.5	2	3	18768.425	18768.424	0.000	0.489
1	0	1	0.5	1.5	1	2	0	0	0	0.5	0.5	1	1	18770.792	18770.792	0.000	0.141
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Notes. ${}^{(a)}$ Calculated line strength factor.

Table B.1. continued.

<i>N'</i>	<i>K'</i> _a	K_c'	J'	F_1'	F_2'	F'	N"	<i>K</i> ′′′	K'' _c	J''	F''_1	F''_2	F"	ν _{obs} (MHz)	ν _{calc} (MHz)	Obs-Calc (MHz)	Int. (a)
1	0	1	0.5	1.5	1	2	0	0	0	0.5	0.5	1	2	18777.011	18777.012	-0.001	0.072
1	0	1	0.5	1.5	1	1	0	0	0	0.5	0.5	1	2	18782.635	18782.635	0.000	0.062
2	0	2	2.5	2.5	3	3	1	0	1	1.5	2.5	3	3	37473.508	37473.508	0.000	0.087
2	0	2	2.5	2.5	3	4	1	0	1	1.5	2.5	3	4	37477.170	37477.173	-0.003	0.298
2	0	2	2.5	2.5	2	2	1	0	1	1.5	2.5	3	2	37479.963	37479.958	0.005	0.189
2	0	2	2.5	3.5	4	3	1	0	1	1.5	2.5	3	2	37481.282	37481.280	0.002	0.055
2 2	0	2 2	2.5 2.5	2.5 1.5	2 2	3	1 1	0	1 1	1.5 1.5	2.5 1.5	3 2	3	37481.398	37481.393 37481.940	0.005	0.234 0.241
2	0	2	2.5	1.5	2	2	1	0	1	1.5	1.5	1	2	37481.934 37482.690	37481.940	-0.006 0.003	0.241
2	0	2	2.5	3.5	4	3	1	0	1	1.5	2.5	3	3	37485.589	37485.583	0.003	0.172
2	0	2	2.5	1.5	2	1	1	0	1	1.5	1.5	1	2	37486.615	37486.607	0.008	0.054
2	0	2	2.5	1.5	2	2	1	0	1	1.5	0.5	1	1	37486.620	37486.619	0.001	1.285
2	0	2	2.5	2.5	3	2	1	0	1	1.5	1.5	2	1	37486.636	37486.630	0.006	0.224
2	0	2	2.5	1.5	2	3	1	0	1	1.5	0.5	1	2	37486.748	37486.742	0.006	2.547
2	0	2	2.5	1.5	2	1	1	0	1	1.5	1.5	1	0	37486.940	37486.942	-0.002	0.503
2	0	2	2.5	2.5	3	4	1	0	1	1.5	1.5	2	3	37487.310	37487.315	-0.005	3.286
2	0	2	2.5	1.5	1	2	1	0	1	1.5	1.5	2	2	37487.342	37487.343	-0.001	0.209
2	0	2	1.5	2.5	3	3	1	0	1	1.5	2.5	3	2	37487.579	37487.574	0.005	2.372
2	0	2	2.5	2.5	3	3	1	0	1	1.5	1.5	1	2	37487.597	37487.592	0.005	1.833
2	0	2	2.5	3.5	4	5	1	0	1	1.5	2.5	3	4	37487.598	37487.598	0.000	4.400
2	0	2	2.5	1.5	1	2	1	0	1	1.5	1.5	2	1	37488.174	37488.170	0.004	1.211
2	0	2	1.5	2.5	2	2	1	0	1	0.5	1.5	1	2	37488.564	37488.556	0.008	0.325
2 2	0	2 2	2.5	2.5 2.5	2 2	3	1	0	1 1	1.5	1.5 1.5	2	2	37489.902 37490.414	37489.907	-0.005 0.000	2.209 1.400
2	0	2	1.5 2.5	1.5	2	3 1	1 1	0	1	0.5 1.5	0.5	1 1	2	37490.414	37490.414 37490.539	0.000	0.348
2	0	2	2.5	1.5	2	2	1	0	1	1.5	0.5	1	2	37490.339	37490.339	-0.002	0.348
2	0	2	2.5	2.5	3	3	1	0	1	1.5	1.5	2	3	37490.930	37490.938	0.002	0.622
2	0	2	2.5	1.5	1	1	1	0	1	1.5	1.5	2	1	37491.506	37491.500	0.006	0.059
2	0	2	2.5	3.5	3	2	1	0	1	1.5	2.5	2	2	37492.147	37492.152	-0.005	0.473
2	0	2	2.5	3.5	3	2	1	0	1	1.5	2.5	2	1	37492.449	37492.452	-0.003	1.394
2	0	2	1.5	0.5	1	1	1	0	1	0.5	0.5	0	1	37492.480	37492.487	-0.007	0.462
2	0	2	2.5	2.5	2	2	1	0	1	1.5	1.5	2	2	37492.774	37492.776	-0.002	0.170
2	0	2	2.5	3.5	4	3	1	0	1	1.5	2.5	3	4	37492.978	37492.974	0.004	0.114
2	0	2	2.5	2.5	2	1	1	0	1	1.5	0.5	1	0	37493.001	37493.007	-0.006	0.132
2	0	2	2.5	3.5	3	3	1	0	1	1.5	2.5	2	2	37493.031	37493.034	-0.003	2.285
2	0	2	2.5	3.5	3	3	1	0	1	1.5	2.5	2	3	37493.173	37493.176	-0.003	0.468
2	0	2	2.5	2.5	2	2	1	0	1	1.5	1.5	2	1	37493.607	37493.603	0.004	0.156
2	0	2	1.5	2.5	3	2	1	0	1	0.5	1.5	2	2	37493.802	37493.804	-0.002	0.785
2	0	2	2.5	1.5	1	0	1	0	1	1.5	1.5	2	1	37493.916	37493.909	0.007	0.065
2 2	0	2 2	1.5 1.5	2.5 1.5	3 2	4 1	1 1	0	1 1	1.5 1.5	2.5 2.5	2 2	3 2	37494.060 37494.093	37494.067 37494.087	-0.007 0.006	3.562 0.102
2	0	2	1.5	1.5	2	1	1	0	1	1.5	2.5	2	1	37494.093	37494.087	-0.001	0.102
2	0	2	1.5	1.5	1	0	1	0	1	0.5	1.5	1	1	37494.952	37494.962	-0.001	0.443
2	0	2	1.5	1.5	2	2	1	0	1	1.5	2.5	2	2	37495.683	37495.686	-0.003	0.261
2	0	2	1.5	1.5	2	2	1	0	1	1.5	2.5	2	3	37495.826	37495.828	-0.002	0.073
2	0	2	2.5	2.5	2	1	1	0	1	1.5	0.5	0	1	37496.386	37496.391	-0.005	0.229
2	0	2	1.5	1.5	2	3	1	0	1	1.5	2.5	2	2	37497.641	37497.645	-0.004	0.116
2	0	2	1.5	1.5	2	3	1	0	1	1.5	2.5	2	3	37497.787	37497.787	0.000	0.441
2	0	2	2.5	2.5	2	2	1	0	1	1.5	1.5	1	2	37498.348	37498.345	0.003	0.273
2	0	2	2.5	2.5	2	3	1	0	1	1.5	1.5	2	3	37498.931	37498.926	0.005	0.200
2	0	2	1.5	1.5	1	1	1	0	1	0.5	1.5	1	2	37499.444	37499.439	0.005	0.444
2	0	2	1.5	2.5	3	2	1	0	1	1.5	1.5	2	2	37499.651	37499.657	-0.006	0.052
2	0	2	2.5	1.5	1	1	1	0	1	1.5	0.5	1	1	37500.173	37500.174	-0.001	0.116
2	0	2	1.5	2.5	3	2	1	0	1	0.5	1.5	2	3	37500.509	37500.506	0.003	0.145
2	0	2	2.5	1.5	1	2	1	0	1	1.5	0.5	1	2	37501.169	37501.163	0.006	0.140
2	0	2	1.5	1.5	1	2	1	0	1	0.5	1.5	1	2	37501.191	37501.185	0.006	0.771
2	0	2	1.5	2.5	3	3	1	0	1	0.5	1.5	2	3	37501.238	37501.240	-0.002	0.231
2 2	0	2 2	2.5 1.5	2.5 2.5	2 2	2 2	1 1	0	1 1	1.5 1.5	1.5 2.5	2 3	3 2	37501.794 37501.804	37501.795 37501.817	-0.001 -0.013	0.074 0.272
2	0	2	1.5	2.5	2	3	1	0	1	1.5	2.5	3	3	37501.804	37501.817	-0.013 -0.007	0.272
2	0	2	1.5	2.5	2	1	1	0	1	0.5	1.5	2	2	37508.394	37508.394	0.000	0.254
2	0	2	1.5	0.5	0	1	1	0	1	0.5	1.5	1	2	37508.720	37508.720	0.000	0.209
2	0	2	1.5	2.5	2	2	1	0	1	0.5	1.5	2	2	37508.780	37508.720	-0.001	0.069
2	0	2	1.5	2.5	2	3	1	0	1	0.5	1.5	2	2	37510.641	37510.639	0.002	0.404
2	0	2	1.5	1.5	1	2	1	0	1	1.5	2.5	3	3	37518.740	37518.749	-0.009	0.055
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Appendix C: Observed deuteration in other molecular species

In order to compare the derived deuteration enhancement in $\rm H_2CCN$ with that of other species, we searched in our data for the lines of the main and singly deuterated isotopologs of $\rm CH_3CN$, $\rm CH_3CCH$, $c\text{-}\rm C_3H_2$, $\rm C_4H$, and $\rm H_2C_4$. The line parameters for the observed rotational transitions of these species are given in Table C.1. For most of these species, only one or two low-J lines were observed in the line survey. We assume a rotational temperature of 10 K for these species. We would like to stress that for most of the species analyzed the variation in column density between $T_r = 5 - 10 \, \rm K$ is rather small as was the case for HDCCN (see Sect. 4.1). We observe several rotational transitions for $\rm C_6H$ and a rotational diagram was performed. The derived XH/XD abundances are given in Table 3. We now comment on each molecule.

In our survey CH₃CCH only has a rotational transition for each symmetry species A and E. The data have previously been shown to compare their A/E abundance ratio with that of CH₃CO⁺ (Cernicharo et al. 2021a). The derived line parameters are given in Table C.1. The total column density (A+E) of CH₃CCH for $T_r = 10 \text{ K}$ is $(1.3 \pm 0.1) \times 10^{14} \text{ cm}^{-2}$. Adopting a $T_r = 5 \text{ K}$, we derive practically the same column density. Taking into account the low dipole moment of this species, we could expect that its lowest energy levels are thermalized at the kinetic temperature of the cloud. Two singledeuterated species are observed. We obtain a column density for $T_r = 10$ of $(2.8 \pm 0.3) \times 10^{12}$ cm⁻² from the line parameters of CH₃CCD given in Table C.1. The H/D abundance ratio for this deuterated isotologue is ~48. The other single-deuterated substitute, CH₂DCCH, is an asymmetric rotor and several lines are observed in our line survey. We derive a column density for $T_r =$ $10 \text{ K of } (1.4 \pm 0.1) \times 10^{13} \text{ cm}^{-2} \text{ from the derived line parameters}$ (see Table C.1). This value decreases by 15% if a rotational temperature of 5 K is adopted. The H/D abundance ratio for this deuterated isotopolog is ~10. To compare with the value derived for CH₃CCD, we have to take into account a factor of 3 in the abundance ratio coming from the three identical possible substitutions of H by D ina the methyl group. It seems that deuteration in the methyl group is slightly favored (H/D = 30) with respect to deuteration in the acetylenic group of the molecule (H/D = 48). The three ¹³C isotopologs were detected. Using the integrated lines intensity, we derive an averaged ${}^{12}\text{C}/{}^{13}\text{C}$ ratio of 75 ± 10 . This abundance ratio is in very good agreement with the value derived by Cernicharo et al. (2020c) from the ¹³C isotopologs of HC_5N .

We also show the data previously in the same context as for CH₃CCH (Cernicharo et al. 2021a) for CH₃CN. The line parameters for all hyperfine components of this molecule are given in Table C.1. We derive N(A-CH₃CN) = $(2.5 \pm 0.2) \times 10^{12}$ cm⁻² and N(E-CH₃CN) = $(2.2 \pm 0.2) \times 10^{12}$ cm⁻² ($T_r = 10$ K). For CH₂DCN, we detect three rotational transitions exhibiting hyperfine splitting. To the best of our knowledge, this hyperfine structure has only been observed in the early years of microwave spectroscopy (Thomas et al. 1955). The reported frequencies, often blends of several hyperfine components, are shifted by 50–100 kHz with respect to the observed frequencies

in TMC-1. From these new rest frequencies we derive $\chi_{aa} = -4.194\pm0.012$ MHz. We find that χ_{bb} was not necessary to reproduce the observed transitions within our frequency accuracy $(10 \, \text{kHz})$. The column density derived for this deuterated species is $(4.5\pm0.6)\times10^{11}$ cm⁻² and the N(CH₃CN)/N(CH₂DCN) abundance ratio is ~11, that is, very similar to the value found for the N(CH₃CCH)/N(CH₂DCCH) abundance ratio.

We observed the $3_{2,1}$ – $3_{1,2}$ ortho line at 44104.777 MHz for c-C₃H₂. This line appears in absorption against the cosmic background radiation. This line was not detected in the line survey of TMC-1 performed with the Nobeyama telescope (Kaifu et al. 2004). In addition, we observed in three *para* lines in our data (see Table C.1). Of these three lines, only the strongest one was reported by Kaifu et al. (2004). One of the para lines (4_{40} – 3₃₁) also appears in our data in absorption against the cosmic background radiation. Hence, it is not possible to adopt a uniform rotational temperature for the lines of both ortho and para species. Nevertheless, collisional rates are available for the system c-C₃H₂/He (Avery & Green 1989), which allows us to perform excitation models using the large velocity gradient approximation. For the typical volume density of TMC-1 of 4×10^4 cm⁻³ and $T_K = 10 \,\mathrm{K}$, the three para lines are very well reproduced, including the one in absorption that has a column density for this species of $N(p-c-C_3H_2) = (3.2 \pm 0.3) \times 10^{13} \text{ cm}^{-2}$. The absorption produced by the ortho line shows a very strong dependence on the density for a large range of column densities. The line is predicted in absorption for $n(H_2) \le 3 \times 10^{-6}$ 10^4 cm⁻³ and N(o-c- C_3H_2) $\geq 5 \times 10^{13}$ cm⁻². The line is predicted to be in emission for larger densities and/or lower column densities. We found that $n(H_2) = (2-3) \times 10^4 \text{ cm}^{-3}$ and N(o-c- C_3H_2) = $(9 \pm 1.0) \times 10^{13}$ cm⁻² reproduce the observed absorption of this species. Hence, the total column density of c-C₃H₂ is $(1.2 \pm 0.1) \times 10^{14}$ cm⁻², that is, very similar to that of CH₃CCH. The derived *ortho/para* ratio is 2.8 ± 0.6 . The column density and rotational temperature of c-C₃HD can be derived from the four observed lines to be $(4.5 \pm 0.5) \times 10^{12}$ cm⁻² and $T_r = 6 \pm 1$ K, respectively. Therefore, the H/D abundance ratio in this species is 27 ± 5 .

Six lines of $\rm H_2C_4$ and two of HDC₄were found in our survey. The line parameters are given in Table C.1. We explored the column densities and rotational temperatures that reproduce at the best the observed line intensities. We find that for $T_r = 10 \, \rm K \, N(o\text{-}H_2C_4) = (3.0 \pm 0.3) \times 10^{12} \, \rm cm^{-2}$ and $N(o\text{-}H_2C_4) = (1.0 \pm 0.1) \times 10^{12} \, \rm cm^{-2}$. Similar results are obtained for $T_r = 7 \, \rm K$. For $T_r = 5 \, \rm K$, the column densities have to be increased by a 25%. The *ortho/para* abundance ratio is 3 ± 0.3 for this species. We obtain $N(\text{HDC}_4) = (4.8 \pm 0.5) \times 10^{10} \, \rm cm^{-2}$ for $T_r = 10 \, \rm K$. Consequently, the H/D abundance ratio for H_2C_4 is 83 ± 15 .

For C₄H, we modeled the emission for all the lines in our survey, including the weak hyperfine lines that are typically a factor 30–50 weaker than the strongest components. The best fit is obtained for $T_r = 7 \, \text{K}$ and $N(C_4H) = (1.3 \pm 0.1) \times 10^{14} \, \text{cm}^{-2}$, although a model with the same column density and $T_r = 10 \, \text{K}$ provides practically the same result. The column density increases by a 20% for $T_r = 5 \, \text{K}$. Adopting $T_r = 7 \, \text{K}$ for C₄D, we derive $N(C_4D) = (1.1 \pm 0.1) \times 10^{11} \, \text{cm}^{-2}$. Hence, the H/D abundance ratio for C₄H is 118 ± 20 .

Table C.1. Observed line parameters for selected molecules and their deuterated isotopologs in TMC-1.

Molecule	Quantum numbers (a)	$v_{\mathrm{rest}}^{(b)}$	$\int T_A^* dv^{(c)}$	$v_{\rm LSR}^{~(d)}$	$\Delta v^{(e)}$	T_A^{*} (f)
		(MHz)	$(mK km s^{-1})$	$(km s^{-1})$	$(km s^{-1})$	(mK)
CH ₃ CCH	2 0-1 0	34183.4139 ± 0.0001	338.0 ± 0.5	5.79 ± 0.01	0.76 ± 0.01	274.4 ± 0.8
CH ₃ CCH	2 1–1 1	34182.7603 ± 0.0001	262.5 ± 0.5	5.79 ± 0.01	0.75 ± 0.01	209.3 ± 0.8
CH ₃ CCD	2 0-1 0	31152.6032 ± 0.0007	5.4 ± 1.5	5.87 ± 0.05	0.83 ± 0.12	6.1 ± 0.8
CH_3CCD	2 1–1 1	31152.0321 ± 0.0007	4.2 ± 1.5	5.73 ± 0.09	0.99 ± 0.17	4.0 ± 0.8
CH_3CCD	3 0-2 0	46728.7668 ± 0.0010	9.2 ± 1.1	5.79 ± 0.02	0.58 ± 0.04	14.8 ± 0.8
CH ₃ CCD	3 1–2 1	46727.9102 ± 0.0010	7.6 ± 1.0	5.82 ± 0.02	0.53 ± 0.04	13.5 ± 0.8
CH ₂ DCCH	2 1 2–1 1 1	32231.5022 ± 0.0019	7.4 ± 0.9	5.80 ± 0.02	0.86 ± 0.05	8.1 ± 0.5
CH ₂ DCCH	2 0 2-1 0 1	32362.1261 ± 0.0020	20.0 ± 0.6	5.77 ± 0.01	0.73 ± 0.01	25.7 ± 0.5
CH ₂ DCCH	2 1 1–1 1 0	32491.9172 ± 0.0019	8.6 ± 0.9	5.80 ± 0.02	0.89 ± 0.05	9.1 ± 0.5
CH ₂ DCCH	3 1 3–2 1 2	48347.0251 ± 0.0028	18.0 ± 1.5	5.79 ± 0.01	0.62 ± 0.03	27.4 ± 1.0
CH ₂ DCCH	3 2 2-2 2 1	48540.0261 ± 0.0027	3.0 ± 0.6	5.95 ± 0.06	0.60 ± 0.12	4.8 ± 1.1
CH ₂ DCCH	3 2 1–2 2 0	48540.4725 ± 0.0027	22.0		0.60 0.02	≤3.3
CH ₂ DCCH	3 0 3-2 0 2	48542.7476 ± 0.0029	33.8 ± 2.3	5.79 ± 0.01	0.60 ± 0.03	53.2 ± 1.1
CH ₂ DCCH	3 1 2–2 1 1	48737.6403 ± 0.0029	15.5 ± 1.5	5.80 ± 0.01	0.59 ± 0.03	24.7 ± 1.1
¹³ CH ₃ CCH	2 0–1 0	33252.9070 ± 0.0002	2.3 ± 0.6	5.77 ± 0.04	0.65 ± 0.10	3.3 ± 0.4
¹³ CH ₃ CCH	2 1–1 1	33252.2861 ± 0.0002	2.6 ± 0.6	5.93 ± 0.05	0.76 ± 0.10	3.2 ± 0.4
¹³ CH ₃ CCH	3 0–2 0	49879.1918 ± 0.0002	5.2 ± 2.1	5.83 ± 0.06	0.50 ± 0.12	9.6 ± 2.1
¹³ CH ₃ CCH	3 1–2 1	49878.2606 ± 0.0002	6.2 ± 2.1	5.93 ± 0.04	0.49 ± 0.08	11.9 ± 2.1
CH ₃ ¹³ CCH	2 0–1 0	34169.2361 ± 0.0001	3.2 ± 0.6	5.75 ± 0.03	0.76 ± 0.07	3.9 ± 0.4
CH ₃ ¹³ CCH	2 1–1 1	34168.5850 ± 0.0002	2.4 ± 0.8	5.84 ± 0.06	0.95 ± 0.15	2.3 ± 0.4
CH ₃ C ¹³ CH	2 0–1 0	33160.8989 ± 0.0002	3.1 ± 0.7	5.81 ± 0.04	0.82 ± 0.09	3.6 ± 0.4
$CH_3C^{13}CH$	2 1–1 1	33160.2761 ± 0.0002	1.8 ± 0.7	5.82 ± 0.04	0.54 ± 0.14	3.2 ± 0.4
$CH_3C^{13}CH$	3 0–2 0	49741.1817 ± 0.0003	2.7 ± 1.4	5.82 ± 0.09	0.52 ± 0.07	4.9 ± 1.9
$CH_3C^{13}CH$	3 1–2 1	49740.2489 ± 0.0003	3.7 ± 1.7	5.91 ± 0.07	0.53 ± 0.09	6.5 ± 1.9
CH ₃ CN	2 1-1 1 2-1	36793.7090 ± 0.0003	39.0 ± 0.9	5.74 ± 0.01	0.68 ± 0.01	53.6 ± 0.4
CH_3CN	2 0-1 0 2-2	36794.2044 ± 0.0003	19.9 ± 0.6	5.75 ± 0.01	0.68 ± 0.01	27.6 ± 0.4
CH_3CN	2 1–1 1 2–2	36794.3402 ± 0.0002	20.0 ± 0.8	5.61 ± 0.05	0.84 ± 0.02	22.5 ± 0.4
CH_3CN	2 0-1 0 1-0	36794.4173 ± 0.0002	19.6 ± 0.6	5.69 ± 0.05	0.60 ± 0.01	30.6 ± 0.4
CH ₃ CN	2 1–1 1 1–1	36794.7618 ± 0.0002	12.3 ± 0.7	5.74 ± 0.01	0.66 ± 0.02	17.6 ± 0.4
CH ₃ CN	2 1–1 1 3–2	36795.0234 ± 0.0001	69.5 ± 1.3	5.73 ± 0.01	0.67 ± 0.01	98.2 ± 0.4
CH ₃ CN	2 0-1 0 2-1	36795.4747 ± 0.0001	51.6 ± 1.1	5.79 ± 0.01	0.63 ± 0.01	77.2 ± 0.4
CH ₃ CN	2 0-1 0 3-2	36795.5669 ± 0.0001	113.6 ± 1.9	5.77 ± 0.01	0.72 ± 0.01	147.8 ± 0.4
CH ₃ CN	2 1–1 1 1–0	36796.3466 ± 0.0005	21.8 ± 0.9	5.74 ± 0.01	0.76 ± 0.02	27.1 ± 0.4
CH ₃ CN	2 0-1 0 1-1	36797.5829 ± 0.0006	20.9 ± 0.9	5.74 ± 0.01	0.69 ± 0.02	28.4 ± 0.4
CH ₂ DCN	2 ₁₂ -1 ₁₁ 2-1	34583.1020 ± 0.0100	2.6 ± 0.7	5.83 ± 0.00	0.84 ± 0.14	2.9 ± 0.4
CH DCN	2_{12} - 1_{11} 2 - 2	34583.1020 ± 0.0100	21:00	5.83 ± 0.00	0.61 + 0.10	22.04
CH ₂ DCN	2_{12} – 1_{11} 3– 2	34584.4110 ± 0.0100	2.1 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.61 ± 0.10	3.2 ± 0.4
CH ₂ DCN CH ₂ DCN	$\begin{array}{c} 2_{02} - 1_{01} \ 2 - 2 \\ 2_{02} - 1_{01} \ 1 - 0 \end{array}$	34733.9580 ± 0.0100 34734.1710 ± 0.0100	1.6 ± 0.0 1.3 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.51 ± 0.08 0.35 ± 0.08	2.9 ± 0.4 3.8 ± 0.4
CH ₂ DCN CH ₂ DCN	$2_{02}-1_{01}$ 1-0 $2_{02}-1_{01}$ 2-1	34735.2280 ± 0.0100	1.3 ± 0.0 5.8 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.33 ± 0.08 0.75 ± 0.07	3.8 ± 0.4 7.2 ± 0.4
CH ₂ DCN CH ₂ DCN	$2_{02}-1_{01}$ $2-1$ $2_{02}-1_{01}$ $3-2$	34735.2280 ± 0.0100 34735.3190 ± 0.0100	3.8 ± 0.0 10.6 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.73 ± 0.07 0.72 ± 0.04	13.9 ± 0.4
CH ₂ DCN CH ₂ DCN	$2_{02}-1_{01}$ 3-2 $2_{02}-1_{01}$ 1-1	34737.3130 ± 0.0100 34737.3130 ± 0.0100	1.7 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.72 ± 0.04 0.70 ± 0.09	13.9 ± 0.4 2.4 ± 0.4
CH ₂ DCN	$2_{02} - 1_{01} 1 - 1$ $2_{11} - 1_{10} 2 - 1$	3484.3920 ± 0.0100	1.7 ± 0.0 1.6 ± 0.0	5.83 ± 0.00 5.83 ± 0.00	0.70 ± 0.09 0.88 ± 0.18	1.8 ± 0.3
CH ₂ DCN	$2_{11}-1_{10}$ $2-1$ $2_{11}-1_{10}$ $1-1$	34884.3920 ± 0.0100 34884.3920 ± 0.0100	1.0 ± 0.0	5.05 ± 0.00	5.00 ± 0.10	1.0 ± 0.5
CH ₂ DCN	2_{11} -1_{10} 1 -1_{10} 2_{11} -1_{10} 3 -2	34885.7090 ± 0.0100	2.2 ± 0.0	5.83 ± 0.00	0.62 ± 0.08	3.3 ± 0.3
CH ₂ DCN	2_{11} 1_{10} 3 2 2_{11} -1_{10} 2 -2	34885.7090 ± 0.0100 34885.7090 ± 0.0100	2.2 ± 0.0	2.02 ± 0.00	5.02 ± 0.00	5.5 ± 0.5
	-11 -10	2 1002.7 07 0 2 0.0100				

Notes. ^(a)Quantum numbers are JK for symmetric rotors, J_{K_a,K_c} for asymmetric rotors. If the molecule contains an atom of nitrogen, then the additional quantum number corresponds to F, the total angular momentum of the molecule. ^(b)Rest frequencies from MADEX (Cernicharo 2012) and/or the CMDS (Müller et al. 2005). If the uncertainty on the velocity is zero, then the observed frequency is given fixing the velocity of the source to 5.83 km s⁻¹. ^(c)Integrated line intensity in mK km s⁻¹. ^(d)Local standard of rest velocity of the line in mK km s⁻¹. If the uncertainty is equal to zero, then the velocity is fixed to 5.83 km s⁻¹. ^(e)Line width at half intensity derived by fitting a Gaussian line profile to the observed transitions (in km s⁻¹). ^(f)Antenna temperature and 1σ uncertainty in mK. For upper limits we provide a 3σ value.

Table C.1. continued.

Molecule	Quantum numbers (a)	$\nu_{\mathrm{rest}}^{(b)}$	$\int T_A^* dv^{(c)}$	v _{LSR} ^(d)	$\Delta v^{(e)}$	T_A^* (f)
		(MHz)	$(mK km s^{-1})$	$(km s^{-1})$	$(km s^{-1})$	(mK)
c-C ₃ H ₂	3 ₂₁ –3 ₁₂	44104.7766 ± 0.0014	-29.5 ±	5.66 ± 0.01	0.53 ± 0.01	-52.4 ± 0.7
c-C ₃ H ₂	4 ₄₀ -3 ₃₁	35360.9275 ± 0.0018	-2.0 ± 0.0	5.88 ± 0.05	0.62 ± 0.08	-3.0 ± 0.4
c-C ₃ H ₂	4 ₃₁ –4 ₂₂	42231.2517 ± 0.0012	3.5 ± 0.7	5.73 ± 0.03	0.63 ± 0.07	5.1 ± 0.5
c-C ₃ H ₂	$2_{11}-2_{02}$	46755.6096 ± 0.0013	514.1 ± 8.8	5.75 ± 0.02	0.62 ± 0.01	780.7 ± 0.7
c-C ₃ HD	3 ₂₁ -3 ₁₂	35600.4822 ± 0.0030	4.8 ± 0.6	5.85 ± 0.02	0.69 ± 0.05	6.5 ± 0.4
c-C ₃ HD	2_{11} – 2_{02}	38224.4450 ± 0.0030	54.1 ± 1.0	5.83 ± 0.01	0.76 ± 0.01	66.8 ± 0.4
c-C ₃ HD	3_{13} – 2_{20}	39605.9580 ± 0.0080				≤1.5
c-C ₃ HD	1_{01} – 0_{00}	42064.1467 ± 0.0020	29.8 ± 1.1	5.72 ± 0.05	0.79 ± 0.02	35.3 ± 0.4
c-C ₃ HD	4 ₃₁ –4 ₂₂	44828.8655 ± 0.0040				≤2.4
c-C ₃ HD	3 ₃₀ –3 ₂₁	47788.7950 ± 0.0060				≤4.5
c-C ₃ HD	1_{11} – 0_{00}	49615.8560 ± 0.0026	142.0 ± 3.7	5.78 ± 0.01	0.65 ± 0.01	205.9 ± 2.2
H_2C_4	4 ₁₄ –3 ₁₃	35577.0079 ± 0.0015	143.0 ± 0.0	5.71 ± 0.01	0.68 ± 0.01	197.8 ± 0.4
H_2C_4	4_{04} – 3_{03}	35727.3789 ± 0.0011	92.3 ± 0.0	5.56 ± 0.01	0.65 ± 0.01	132.7 ± 0.5
H_2C_4	4_{13} – 3_{12}	35875.7746 ± 0.0016	143.2 ± 0.0	5.70 ± 0.01	0.65 ± 0.01	206.2 ± 0.4
H_2C_4	5 ₁₅ –4 ₁₄	44471.1375 ± 0.0018	123.2 ± 0.0	5.72 ± 0.01	0.48 ± 0.01	243.3 ± 0.7
H_2C_4	5 ₀₅ –4 ₀₄	35727.3789 ± 0.0011	75.2 ± 0.0	5.57 ± 0.01	0.48 ± 0.01	146.3 ± 0.8
H_2C_4	5 ₁₄ –4 ₁₃	44844.5902 ± 0.0020	123.2 ± 0.0	5.71 ± 0.01	0.53 ± 0.01	217.0 ± 0.8
HDC ₄	4_{04} – 3_{03}	33902.1130 ± 0.0030	4.6 ± 0.0	5.72 ± 0.02	0.73 ± 0.05	6.0 ± 0.4
HDC_4	5 ₀₅ -4 ₀₄	42377.2757 ± 0.0080	4.5 ± 0.0	5.73 ± 0.03	0.56 ± 0.07	7.6 ± 0.6
CCCCH	4-3 9/2-7/2 4-3	38049.6160 ± 0.0010	550.6 ± 9.4	5.78 ± 0.01	0.60 ± 0.01	862.1 ± 0.4
CCCCH	4-3 9/2-7/2 5-4	38049.6910 ± 0.0010	641.9 ± 9.8	5.78 ± 0.01	0.60 ± 0.01	1005.0 ± 0.4
CCCCH	4-3 9/2-7/2 4-4	38059.4310 ± 0.0020	24.6 ± 0.6	5.74 ± 0.01	0.65 ± 0.01	35.8 ± 0.4
CCCCH	4-3 7/2-5/2 3-3	38078.9300 ± 0.0020	25.1 ± 0.6	5.74 ± 0.01	0.65 ± 0.01	36.0 ± 0.4
CCCCH	$4-3 \ 7/2-5/2 \ 4-3 + 3-2$	38088.4600 ± 0.0010	913.3 ± 9.9	5.64 ± 0.01	0.77 ± 0.01	1120.8 ± 0.5
CCCCH	4-3 7/2-7/2 3-3	38212.6370 ± 0.0040	21.6 ± 0.7	5.75 ± 0.01	0.58 ± 0.01	34.8 ± 0.5
CCCCH	4-3 7/2-7/2 3-4	38231.9620 ± 0.0040	23.9 ± 0.7	5.74 ± 0.01	0.60 ± 0.01	37.5 ± 0.5
CCCCH	5-4 11/2-9/2 5-4 + 6-5	47566.7920 ± 0.0020	953.1 ± 9.9	5.64 ± 0.01	0.69 ± 0.01	1297.1 ± 1.2
CCCCH	5-4 11/2-9/2 5-5	47576.5100 ± 0.0020	14.0 ± 1.5	5.72 ± 0.01	0.51 ± 0.03	25.8 ± 1.2
CCCCH	5-4 9/2-7/2 4-4	47595.9910 ± 0.0020	15.1 ± 1.5	5.69 ± 0.01	0.52 ± 0.03	27.2 ± 1.2
CCCCH	5-4 9/2-7/2 5-4 + 4-3	47605.4950 ± 0.0020	761.4 ± 9.9	5.78 ± 0.01	0.58 ± 0.01	1224.6 ± 1.2
CCCCH	5-4 9/2-9/2 4-4	47768.5220 ± 0.0040	15.5 ± 1.4	5.76 ± 0.01	0.55 ± 0.02	26.5 ± 1.4
ССССН	5-4 9/2-9/2 5-5	47787.7600 ± 0.0040	15.0 ± 1.0	5.68 ± 0.01	0.49 ± 0.01	28.8 ± 1.4
CCCCD	4–3 9/2–7/2	35313.3131 ± 0.0020	17.3 ± 0.0	5.72 ± 0.01	0.89 ± 0.02	18.3 ± 0.4
CCCCD	4–3 7/2–5/2	35349.4296 ± 0.0020	13.6 ± 0.0	5.78 ± 0.01	0.89 ± 0.03	14.4 ± 0.4
CCCCD	5-4 11/2-9/2	44146.0436 ± 0.0102	18.5 ± 0.0	5.73 ± 0.01	0.63 ± 0.02	27.6 ± 0.8
CCCCD	5-4 9/2-7/2	44182.1087 ± 0.0102	15.4 ± 0.0	5.77 ± 0.01	0.65 ± 0.02	22.4 ± 0.8