Comparative chemistry of diffuse clouds

V. Ammonia and formaldehyde

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ABSTRACT

Using the VLA and NRAO 140’ telescopes we observed the J1.3 cm (1, 1) and (2, 2) lines of NH3 in absorption and emission toward the compact extragalactic continuum sources NRAO150 (B0355+508) and 3C 111 (B0415+379). These sources are occulted by some seven local diffuse and translucent clouds showing molecular absorption in OH, CO, HCO+ and C2H: for the four features having NH3 absorption, we find rotational excitation temperatures 18–24 K, suggesting kinetic temperatures of at least 25–30 K. The abundance ratio \( \frac{N(\text{NH}_3)}{N(\text{HCO}^+)} \) is comparable to values quoted for the cyanopolyne peak in TMC-1 (i.e., 2.5) in three of four cases where \( \text{NH}_3 \) was seen. For clouds with higher column density \( N(\text{HCO}^+) \geq 10^{12} \text{ cm}^{-2} \), the \( \text{NH}_3 \) column density \( N(\text{NH}_3) \) is well correlated only with \( N(\text{CS}) \) \( \langle \frac{N(\text{NH}_3)}{N(\text{CS})} \rangle \approx 1.0 \) and \( N(\text{H}_2\text{CO}) \) \( \langle \frac{N(\text{NH}_3)}{N(\text{H}_2\text{CO})} \rangle \approx 0.4 \). \( N(\text{H}_2\text{CO}) \) is well correlated with \( N(\text{NH}_3) \) and \( N(\text{CS}) \) \( \langle \frac{N(\text{H}_2\text{CO})}{N(\text{NH}_3)} \rangle \approx 3.2 \) and the \( \text{H}_2\text{CO} \) abundance relative to other species is like TMC-1, with \( \langle \frac{N(\text{H}_2\text{CO})}{N(\text{HCO}^+)} \rangle \approx 2.3 \).

Key words. ISM: molecules

1. Introduction

Ammonia and formaldehyde are hardly thought of as common molecules in diffuse clouds but Nash (1990) detected both in absorption toward the compact extragalactic radiocontinuum source B0355+508 (NRAO150), against which Cox et al. (1988) had earlier seen C3H2 in absorption “in the diffuse interstellar medium”. Unfortunately there is no evidence in Nash’s work that she was aware of the C3H2 detection – her observations were actually taken prior to 1988 – and she characterized the absorption features in question as arising in “molecular clouds.” In retrospect, it now seems clear that many of the \( \text{H}_2\text{CO} \) features seen in the galactic plane against galactic HII regions arise in diffuse clouds and it is clear that \( \text{H}_2\text{CO} \) lines, if not the most ubiquitous spectral features, are seen along sightlines where the gas cannot by any means be characterized as very dark or dense (Liszt & Lucas 1995; Moore & Marscher 1995).

Ammonia and formaldehyde have in common the fact that both have been subject to considerable uncertainty as to their formation mechanisms in dense dark gas, being at times understood as only capable of formation on grain surfaces. It is now recognized that \( \text{NH}_3 \) may be produced in the gas phase in dark clouds via the reaction of \( \text{N}^+ \) and \( \text{H}_2 \) (Le Bourlot 1991) or \( \text{N} \) and \( \text{H}_2 \) (Scott et al. 1997) but its origin in more diffuse gas is obscure. Federman & Allen (1991) argued that \( \text{H}_2\text{CO} \) could be seen at extinctions of \( A_V = 1 \text{ mag} \) at the edges of dark clouds, having formed on grains and then diffused or convected out of the protected inner regions. However, maps of extinction toward lines of sight in which \( \text{H}_2\text{CO} \) is seen at \( A_V \leq 1 \text{ mag} \) do not generally reveal the presence of nearby dark clouds.

Here, we discuss new observations of the J1.3 cm (23.7 GHz) (1, 1) and (2, 2) lines of \( \text{NH}_3 \) made recently with the VLA. As is commonly understood, such observations offer the possibility of determining not only the abundance of \( \text{NH}_3 \) – thereby providing interesting insights into the diffuse gas chemistry – but the kinetic temperature \( T_K \) (Danby et al. 1988) as well. Our other observations have provided some interesting insights into the local thermal pressure (Liszt & Lucas 1998), and found it to be moderate, but the density and temperature are difficult to separate in cases of such weak excitation as occurs in diffuse clouds, even in CO. We suspect that the gas has kinetic temperatures well above those of dark clouds, both because the column densities of CO are too small to consider CO to be the main carrier of carbon and because the ratio \( N(\text{^{13}CO})/N(\text{CO}) \) is commonly seen to be 2–3 times higher than the isotopic abundance ratio 1:60 (Lucas & Liszt 1998), most likely due to chemical fractionation (Watson et al. 1976).

Upon finding in the new absorption data that the abundance of \( \text{NH}_3 \) was best-correlated with that of \( \text{H}_2\text{CO} \), we reanalyzed the available data on that species, following our earlier survey.
(Liszt & Lucas 1995) which was published at a time when we had not yet made extensive studies of the various chemical families which are actually detectable in diffuse gas, such as OH and HCO$^+$ (Lucas & Liszt 1996; Liszt & Lucas 1996), CH, C$_2$H and C$_3$H$_2$ (Liszt & Lucas 2002; Lucas & Liszt 2000), CN, HCN and HNC (Liszt & Lucas 2001) and CS, SO, H$_2$S and HCS$^+$ (Lucas & Liszt 2002). Improved results for several sources are also available in the work of Moore & Marscher (1995). Section 2 describes our VLA and NRAO140' observations of NH$_3$ and discusses various elemental aspects of the molecular physics needed to interpret them. In Sect. 3 we compare the abundances of NH$_3$ and H$_2$CO with those of other species.

2. Observations and data

2.1. NH$_3$

2.1.1. Absorption

We observed the 23.7 GHz (1, 1) and (2, 2) lines of NH$_3$ at the VLA in B-configuration for three hours during the first week of 2005 March under program AL645. We used reference pointing on the background sources themselves (see Table 1). Following a bandpass calibration on J0319+415 we observed B0355+508 (NRAO150) for some 30 min, followed by almost 2 h on 3C 111 (including reference pointing). The NH$_3$ lines were observed simultaneously in different 128-channel IF’s at spectral channel spacing 12.2 kHz or 0.154 km s$^{-1}$. Having foregone on-line hanning smoothing to achieve higher resolution, the actual resolution of the spectrometer is some 1.3 times lower than the channel spacing. Source characteristics and achieved rms noise levels are given in Table 1.

Reduction of the VLA data consisted of forming the bandpass cal, applying it to the sources, then forming absorption spectra using the AIPS task UVLSD. The latter is similar to UVLIN but for each integration period it uses absorption-free channels to subtract a linear spectral baseline in the $uv$-plane, producing an absorption spectrum, and averages all such spectra together to produce a final result. The actual flux of the background sources was not determined, but, during a 19.7 GHz observing session at the GBT in 2005 January, the fluxes were about 5 Jy and 2 Jy for B0355+508 and B0415+379, respectively. The VLA spectra achieved are shown in Figs. 1 and 2 below.

2.1.2. NH$_3$ emission

We earlier observed the 23.7 GHz (1, 1) and (2, 2) lines of NH$_3$ at the NRAO 140' during the period 1996 January 24–29. Unfortunately, these observations, when compared to unpublished (1, 1) results of Moore and Marscher, or to the results of Nash (1990), seemed to have sizeable calibration errors (30%–40%), no doubt accruing from the relatively poor antenna performance. We do not refer to these data below for their absorption profiles, but they do suffice to place an interesting limit on the emission brightness and excitation temperature of the (1, 1) line near 3C 111, toward which strong absorption was seen in the earlier data (at 19.54 kHz = 0.246 km s$^{-1}$ resolution). As with our earlier work on H$_2$CO (Liszt & Lucas 1995) at the 140', we observed off-source at positions removed by 1.2 HPBW (in this case 96') in each of the cardinal directions and averaged these to form an emission spectrum: the ratio of

\begin{table}[h]
\centering
\begin{tabular}{llll}
\hline
Source & $l$ & $b$ & $\sigma_{l/c}(1,1)$ & $\sigma_{l/c}(2,2)$ \\
\hline
B0355+508 & 150.3773 & -1.6037 & 2.2 & 2.6 \\
B0415+379 & 161.6764 & -8.8186 & 2.0 & 2.2 \\
\hline
\end{tabular}
\caption{Background sources observed in NH$_3$.}
\label{tab:nh3}
\end{table}

$\sigma_{l/c}$ = rms error in the line/continuum ratio.
We assume that all the molecules are in the lowest states of the various $K$-ladders ($J = K$), because intraladder transitions are connected by the usual fast $\Delta J = \pm 1$ electric dipole transitions with $\mu = 1.47$ Debye. Within one ($J, K$) level, the inversion transitions observed here have dipole moments $\mu_J^2 = \mu^2 K^2 / (J + 1)$. The intraladder excitation temperature of the $K = 1$ ladder is shown below to be very small compared to the derived interladder rotation temperature.

All parameters mentioned here are discussed in Townes and Schawlow (1954) but the factor two degeneracy for inversion used in their formulae 46 must be replaced by $\beta = 1 + \exp(-h\nu/(kT_{\text{CMB}})) = 1.66$ since the 23.7 GHz or 1.14 K inversion splitting is not negligible compared to the excitation temperature (see the next subsection) of 2.73 K. The nuclear statistical weights of the various $K$ ladders are 6, 3$\beta$, 3$\beta$, and 6$\beta$ for $K = 0, 1, 2, 3$ respectively. The statistical weights of the various $J$-levels are $2J + 1$ as usual.

### 2.2. NH$_3$ excitation and rotation temperatures

We know from observations of other molecules (Lucas & Liszt 1996) that rotational excitation of non-metastable levels will be small. Using the single-channel (0.25 km s$^{-1}$ resolution) rms in the average 140$^\circ$ emission spectrum around 3C 111, 0.0019 K, the optical depth at line center 0.022 and the nominal beam efficiency 0.4, it follows that a 2-$\sigma$ upper limit on the excitation temperature within the (1, 1) level is $T_{\text{EXC}}(1, 1) < 3.2$ K. Integrating over the whole line profile, one achieves a somewhat lower limit, $T_{\text{EXC}}(1, 1) < 3.0$ K. Of course, such limits on $T_{\text{EXC}}$ are relaxed if the singledish beam is not filled, but, having detected absorption against a background point source, the intervening molecular medium cannot be too porous. We saw earlier that the excitation temperatures of CO derived in absorption and from comparison of absorption and emission are consistent with a beam-filling, largely uniform gas (Liszt & Lucas 1998). Interferometer maps of molecular emission around B0355+508, now in progress, will clarify this point.

The interladder rotational temperature between the (1, 1) and (2, 2) levels is found from the ratio of column densities, i.e. $T_{\text{ROT}} = -41.5 K/\ln(3N(2, 2)/5N(1, 1))$. The relationship between $T_{\text{ROT}}$ and the kinetic temperature $T_K$ is discussed here in Sect. 3 following the calibration of the NH$_3$ temperature scale by Danby et al. (1988).

### 2.2.2. NH$_3$ column densities

Assuming that the excitation temperature within one ($J, K$) level is 2.73 K as indicated by the discussion in Sect. 2.2.1, it follows that $N(1, 1) = 3.62 \times 10^{13}$ cm$^{-2}$ $\int \tau(1, 1) dv$ and $N(2, 2) = 2.71 \times 10^{13}$ cm$^{-2}$ $\int \tau(2, 2) dv$ where the integrals (in units of km s$^{-1}$) are over all hyperfine subcomponents and the column density totals include both inversion levels. Such are tabulated in the Appendix.

Where detected (see Figs. 1 and 2), the observed absorption profiles were decomposed using a simultaneous gaussian fitting assuming LTE intensity ratios and identical linewidths for all hyperfine components within the passband. The integrated

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**Fig. 2.** As in Fig. 1, but for B0415+379 (3C 111).

**Table 2.** As in Table 1, but for B0415+379 (3C 111).
optical depths tabulated in Tables A.1 and A.2 of the Appendix for the detected lines are sums over all hyperfine components, most but not all of which were observed and fit; the tabulated sums are extrapolated to include the others and the peak optical depth is for the strongest hyperfine component. To derive upper limits for unseen NH3 features toward B0355+508, we first subtracted the fits corresponding to the detected features. We then calculated the rms integrated intensity over those regions of the NH3 passband where the strongest hyperfine components of the "other" NH3 absorption lines – those unseen but corresponding to detected HCO+ features – would have appeared. These limits were extrapolated to limits on the total integrated optical depth assuming LTE line ratios, then converted to column density (see also the following paragraph) and doubled to form the $2\sigma$ upper limits presented in the tables.

The total column density of NH3 shown here was derived assuming that the ortho and para ladders are populated in proportion to their statistical weights: the column density was also calculated under the assumption that all of the rotational levels are populated at the rotational temperature determined from the (1, 1) and (2, 2) levels (perhaps by spin exchange in collisions with protons in diffuse gas, as occurs in H2) and found to be the same within 10%. For features toward B0355+508 having only upper limits on N(1, 1), we formed the ratio of total column density to N(1, 1) for the detected features and applied this average value to all of the limits.

2.3. H2CO

No new observations of H2CO were taken for this work; rather, we blend data from our single dish survey (Liszt & Lucas 1995, their Table 1) and the contemporaneous (1994 May) VLA work of Moore & Marscher (1995, their Table 2). We use the latter's data for B0212, B0415, and B2200 because (being interferometer data) it is potentially better calibrated and not subject to any uncertainty induced by anomalous absorption (excitation below the cosmic microwave background which is present in the extended single dish beam and must be compensated, see Liszt & Lucas 1995). For B0355 there is good agreement between the two datasets and our work covered more kinematic components.

We relate the total column density of H2CO to the optical depth measured at $\lambda$6 cm by assuming that the ortho:para ratio is 3:1 and that all the molecules are in the lowest J-levels of their respective (ortho and para) rotational ladders. The ortho:para ratio was actually determined to be approximately three in the work of Liszt & Lucas (1995) using a combination of cm-wave and mm-wave observations and emission was not detected at quite low levels. It follows that $N$(H2CO)/$\tau$$_{\text{opt}}$ = 3.68 x $10^{13}$ $\int \tau\,d\nu$ using Eq. (2) in Liszt & Lucas (1995): basically one has simply that $N$(H2CO)/$\tau$$_{\text{opt}}$ = 1.3333 $N$(111)/0.46 where the factors 1.333 and 0.46 account (respectively) for the ortho/para ratio and the distribution over ortho-states when $\tau$$_{\text{obs}}$ = $\tau$$_{\text{CMB}}$. $\tau$$_{\text{exc}}$ is the excitation temperature between the levels participating in the 06 cm transition (111 and 110), a typical value for which was found to be 1.7–2.7 K (Liszt & Lucas 1995).

### Table 2. NH3 column densities and $T$$_{\text{rot}}$.

<table>
<thead>
<tr>
<th>Source</th>
<th>$v$ [km s$^{-1}$]</th>
<th>$T$$_{\text{rot}}$ [K]</th>
<th>N(NH3) [10$^{-12}$ cm$^{-2}$]</th>
<th>X(NH3) [$10^{-9}$]</th>
</tr>
</thead>
<tbody>
<tr>
<td>B0355</td>
<td>17.1</td>
<td>17.6</td>
<td>2.69</td>
<td>4.67</td>
</tr>
<tr>
<td></td>
<td>(15, 12)</td>
<td>&gt;0.53</td>
<td>&gt;0.75</td>
<td></td>
</tr>
<tr>
<td></td>
<td>(10.5</td>
<td>20.9</td>
<td>2.54</td>
<td>4.25</td>
</tr>
<tr>
<td></td>
<td>(9.4, 6.9)</td>
<td>&gt;0.46</td>
<td>&gt;0.84</td>
<td></td>
</tr>
<tr>
<td></td>
<td>(6.9, 1.5)</td>
<td>&gt;0.65</td>
<td>&lt;1.13</td>
<td></td>
</tr>
<tr>
<td>B0415</td>
<td>−2.5</td>
<td>21.8</td>
<td>1.68</td>
<td>0.77</td>
</tr>
<tr>
<td></td>
<td>−0.9</td>
<td>23.5</td>
<td>10.53</td>
<td>2.83</td>
</tr>
</tbody>
</table>

1 Assuming X(HCO+) = N(HCO+)/N(H2) = 2 x $10^{-9}$.

2.4. Comparison with results of Nash (1990)

Nash (1990) observed H2CO and NH3 toward B0355+508. As we noted previously (Liszt & Lucas 1995), her H2CO results from the Bonn 100 m are in excellent agreement with our results from the NRAO 43 m, so that we should be able to derive NH3 column densities from the quanta in her tables with confidence. Such derivation is mandated by the fact that she gave only upper limits on N(NH3) (despite having clearly detected several features), and the limits quoted (as multiples of $10^{15}$ cm$^{-2}$) are suspect because they are some three orders of magnitude too generous. Most likely the table units should have been given as $10^{12}$ cm$^{-2}$ and her limits were actually intended to have values which are slightly above the levels of our detections.

Nash (1990) very obviously detected (1, 1) absorption from the main NH3 hyperfine components of the −17 km s$^{-1}$ and −10 km s$^{-1}$ features, and detected (2, 2) absorption from the latter. The rotational excitation temperature we derive from her results at −10 km s$^{-1}$ is 20.9 K, corresponding to $T$$_{K}$ ≥ 24 K, in agreement with Nash (1990) and our data (see Sect. 3.1 and Table 2); the absence of the (2, 2) feature at −17 km s$^{-1}$ in Nash’s data is puzzling but must remain a mystery.

3. Systematics

3.1. NH3

Our NH3 profiles are shown in Figs. 1 and 2 at top, where the profile decompositions are also illustrated. At bottom in each figure are shown representative profiles of other chemical families. NH3 column densities and rotational temperatures appear in Table 2.

The rotational temperatures fall in the relatively narrow range 18–24 K and are uncorrelated with the column density, NH3/HCO+ ratio, etc. According to the most recent re-calibration of the NH3 thermometer (Danby et al. 1988) the observed $T$$_{\text{rot}}$ are at the upper end of the range where the kinetic temperature can reliably be inferred from observations of the (1, 1) and (2, 2) levels alone; we infer lower limits for the kinetic temperature in the range 24–30 K. For the features in which NH3 was detected, $^{13}$CO is enhanced, presumably through chemical fractionation, by factors of order 3 which seems consistent with chemistry in a medium having N(C+)/N(CO) ≫ 1 and $T$$_{K}$ ≈ 30 K (Liszt & Lucas 1998).
As is apparent in Figs. 1 and 2, NH$_3$ is one of the least frequently observed species in our sample of clouds and background sources. Its abundance ratio with respect to HCO$^+$, and, therefore, most likely its abundance with respect to H$_2$, \( X(\text{NH}_3) = \frac{N(\text{NH}_3)}{N(\text{H}_2)} \), varies widely. Some aspects of the systematics are illustrated in Fig. 3 which compares column densities of NH$_3$ and those of other chemical families. The only reliable predictors of \( \text{N} (\text{NH}_3) \) are \( \text{N} (\text{CS}) \) and \( \text{N} (\text{H}_2\text{CO}) \); the NH$_3$ abundance seems especially poorly-correlated with those of the hydrocarbons, represented by C$_2$H (see Lucas & Liszt 2000), but the range in \( \text{N} (\text{C}_2\text{H}) \) is somewhat limited.

### 3.2. H$_2$CO

In Liszt & Lucas (1995) we compared \( \text{N} (\text{H}_2\text{CO}) \) with \( \text{N}(\text{HCO}^+) \) and \( \text{N}(^{13}\text{CO}) \), finding that H$_2$CO, like many species, has a rapid increase in abundance around \( \text{N}(\text{HCO}^+) \approx 1 \times 10^{12} \text{ cm}^{-2} \), but not as rapid as that of \(^{13}\text{CO} \). We know now (Lisz$^t$ & Lucas 1998) that \( \text{N}(^{13}\text{CO}) \) is strongly affected by fractionation, so that the H$_2$CO–$^{13}$CO comparison, while perhaps interesting, is not directly of use in deciphering the general chemistry. In Fig. 4 we show the variation of \( \text{N}(\text{H}_2\text{CO}) \) with \( \text{N}(\text{HCO}^+) \), \( \text{N} (\text{C}_2\text{H}) \), \( \text{N}(\text{HCN}) \) and \( \text{N}(\text{CS}) \). The variation of \( \text{N}(\text{H}_2\text{CO}) \) with \( \text{N}(\text{C}_2\text{H}) \) resembles the run of \( \text{N}(\text{HCN}) \) with \( \text{N}(\text{C}_2\text{H}) \) in (Lucas & Liszt 2000) so that it is not surprising that \( \text{N}(\text{H}_2\text{CO}) \) and \( \text{N}(\text{HCN}) \) are decently correlated. H$_2$CO is best correlated with \( \text{N}(\text{CS}) \) in Fig. 4 and it may be worth noting that the datum with the highest \( \text{N}(\text{CS}) \) is that for 3C 111 at \(-0.9 \text{ km s}^{-1} \), where we have preferred the \( \text{N}(\text{H}_2\text{CO}) \) value of Moore & Marscher (1995), which is a full factor two higher than in our own earlier work. Had we taken our own value, the correlation between CS and H$_2$CO would be more nearly perfect, and that between H$_2$CO and NH$_3$ in Fig. 3 somewhat less so.

### 4. Discussion

#### 4.1. Overview of molecular abundances

Table 3 presents a very reductive summary of the measurements carried out in this series, updating earlier versions; the entries for “our work” in the second column are normalized to \( \text{N}(\text{HCO}^+)/\text{N}(\text{H}_2) = 2 \times 10^{-9} \) and are intended to represent conditions of somewhat higher column density \( \text{N}(\text{HCO}^+) \approx 10^{12} \text{ cm}^{-2} \), \( \text{N}(\text{H}) \approx 10^{21} \text{ cm}^{-2} \) where the relative abundances are more stable. Species placed in the lower group in the table are rarer than those above, all of which are typically (though not invariably) found along a given line of sight. For comparison, the table also summarizes observations toward the archetypal line of sight toward $\zeta$ Oph (from various sources as given in the footnotes there) and toward the dark cloud TMC-1 from Ohishi et al. (1992). The rightmost column summarizes results from the calculations from
Our Work on OH and HCO⁺ despite its high abundance is one of the least-commonly observed features in dark clouds, as seen in OH and HCO⁺ (N(H₂CO) ≈ 10¹² cm⁻²), but the lower velocity component toward 3C 111 is deficient in NH₃ compared with those of HCO⁺. CO is the outstanding exception, typifying the dominance of C⁺ in diffuse gas. Much smaller but clear exceptions are the larger CN/HCN and HCN/HNC ratios in diffuse clouds; larger HCN/HNC ratios are typical of warmer gas, as discussed in Liszt & Lucas (2001).

### 4.2. Summary

We present new observations of absorption from the (1, 1) and (2, 2) levels of NH₃ toward two of the background sources studied in the previous papers in this series and our other related work. The line of sight toward B0355+508 (NRAO150) harbors some five diffuse clouds, each having an OH column density comparable to that seen toward ζ Oph (N(OH) = 5 × 10¹³) and NH₃ was detected from those two which show the greatest overall variety of species. The line of sight toward 3C111 has two clouds of rather higher molecular column density (3–5× larger N(OH) and N(HCO⁺)) and NH₃ was detected from both of these, albeit only very weakly in one.

NH₃ is one of the least-commonly observed features in our sample of clouds and background sources, along with the sulfur-bearing species (CS, SO, H₂S) and H₂CO: N(NH₃) and N(H₂CO) are quite well correlated when both species are detected. HCN is somewhat more easily detected than CS, SO, H₂CO, NH₃, etc. but the column densities of CS and HCN, in particular, are quite well correlated. All these species tend to be found only in clouds of higher N(H₂), as seen in OH and HCO⁺ (N(HCO⁺) ≳ 10¹² cm⁻²), but the lower velocity component toward 3C 111 is deficient in NH₃ despite its high N(HCO⁺).

Such occasional anomalies aside, when species are seen, the column density ratios among them are like those usually quoted for the dark cloud TMC-1.

The rotational temperatures found for all features with detected NH₃ fall in a rather narrow range 18–24 K, indicating an origin in cool but not very cold gas, consistent with the relatively small CO column densities and high $^{13}$CO/¹²CO ratios noted earlier as arising in gas where carbon is mostly in the form of C⁺.
Finding that $N$(NH$_3$) was best correlated with $N$(H$_2$CO) we analyzed available measurements of the latter in diffuse clouds. The column density of H$_2$CO is well-correlated with $N$(CS) as well and where detected – in a much larger survey than has been performed for NH$_3$– the ratio $N$(H$_2$CO)/$N$(HCO$^+$) takes on values typical of TMC-1.

The origin of NH$_3$ in dense dark gas is now understood as arising in the gas phase through pathways originating with either N$^+$ or N, while H$_2$CO is presumed to be formed on dust. Why these species should be so well correlated with each other in di

4.3. Future work

Detecting unexpectedly high abundances of complex molecules along lines of sight lacking dark gas has raised many unanswered questions about interstellar chemistry and the structure of the diffuse interstellar medium. In the next and probably final paper in this series we will discuss line of sight properties and statistics based on an extensive new survey of atoms, molecules and extinction along the lowest-latitude (and hence longest) lines of sight available for study at the present time.

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Appendix A: Products of Gaussian fitting and profile integrals

<table>
<thead>
<tr>
<th>Source</th>
<th>$\nu$</th>
<th>$\tau$</th>
<th>$\sigma_\nu$</th>
<th>$\int \tau(1,1)d\nu$</th>
<th>$N(1,1)$</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>km s$^{-1}$</td>
<td>0.001</td>
<td>km s$^{-1}$</td>
<td>0.01 km s$^{-1}$</td>
<td>10$^{12}$ cm$^{-2}$</td>
</tr>
<tr>
<td>B0355+508</td>
<td>$-17.2(0.07)$</td>
<td>5.31(0.16)</td>
<td>1.21(0.17)</td>
<td>2.94(0.06)</td>
<td>1.06(0.02)</td>
</tr>
<tr>
<td></td>
<td>$(-15, -12)$</td>
<td></td>
<td></td>
<td>&lt;0.54</td>
<td>&lt;0.20</td>
</tr>
<tr>
<td></td>
<td>$-10.5(0.02)$</td>
<td>14.80(0.50)</td>
<td>0.42(0.06)</td>
<td>2.87(0.06)</td>
<td>1.04(0.02)</td>
</tr>
<tr>
<td></td>
<td>$(-9.4, -6.9)$</td>
<td></td>
<td></td>
<td>&lt;0.47</td>
<td>&lt;0.17</td>
</tr>
<tr>
<td></td>
<td>$(-6.9, -1.5)$</td>
<td></td>
<td></td>
<td>&lt;0.67</td>
<td>&lt;0.24</td>
</tr>
<tr>
<td>B0415+379</td>
<td>$-2.55(0.06)$</td>
<td>4.62(0.23)</td>
<td>0.79(0.19)</td>
<td>1.68(0.06)</td>
<td>0.61(0.02)</td>
</tr>
<tr>
<td></td>
<td>$-0.91(0.08)$</td>
<td>35.83(0.33)</td>
<td>0.63(0.03)</td>
<td>10.28(0.07)</td>
<td>3.72(0.03)</td>
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<th>Source</th>
<th>$\nu$</th>
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<th>$\sigma_\nu$</th>
<th>$\int \tau(2,2)d\nu$</th>
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<td>km s$^{-1}$</td>
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<td>23.40(0.45)</td>
<td>0.65(0.05)</td>
<td>3.90(0.08)</td>
<td>10.57(0.22)</td>
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References