

CH₂DCCH along the TMC-1 ridge

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Abstract. In this paper we present measurements of the CH₂DCCH/CH₃CCH ratio along the TMC-1 ridge. The level of deuterium fractionation in this molecule is found to be larger than previously thought, but more significantly, the fractionation increases as one moves from the cyanopolyne peak to the ammonia peak, as is the case with measurements of other molecules in TMC-1. This confirms the prediction of our recent chemical/dynamical model of TMC-1 and supports the hypothesis that the chemical evolution of TMC-1 has been affected by Alfvén waves.

Key words. MHD: waves – ISM: abundances – ISM: molecules – ISM: individual (Taurus Molecular Cloud) – molecular processes

1. Introduction

The small, dark interstellar cloud TMC-1 is a dense, quiescent ridge of gas which runs southeast to northwest in the Heiles Cloud 2 complex of dark interstellar dust clouds, at a distance of 140 pc. It contains an embedded protostellar source, IRAS 04381+2540 towards the northwestern end. When the ridge was mapped in HC₇N and NH₃, the distributions of these molecules were found to differ – the HC₇N emission peaks near the southeastern end of the ridge, but ammonia peaks further to the northwest (Olano et al. 1988). These regions have become known as the CP (cyanopolyne peak) and the AP (ammonia peak). A comprehensive study by Pratap et al. (1997), in which abundances are measured relative to HCO⁺, shows abundance gradients for several molecules, the most pronounced being for CS, SO, HCN, HNC, C₂H, HC₃N.

Studies of deuterated species have also shown spatial gradients to exist along the ridge, this time in the level of fractionation. Observations show that the fractionation in the molecules HCO⁺, C₃H₂, HC₃N and HNC increases from the CP to the AP (Guélin et al. 1982; Butner & Charnley 1997, 2001; Bell et al. 1988; Howe et al. 1994; Hirota & Yamamoto 1998; Hirota et al. 2001).

In Markwick et al. (2000), we presented a model which produces the molecular abundance gradients and explains the organic diversity in the CP region. The model is based on the assumption that shear Alfvén waves are

propagating in TMC-1, perhaps from the embedded protostellar source IRAS 04381+2540. In Markwick et al. (2000), Alfvén waves are shown to be capable of desorbing grain ice mantles, releasing the carbonaceous species C₂H₂, C₂H₄ and CH₄ into the gas phase, which then react in a complex chemistry to enhance the abundances of other organic compounds. After some time, the carbon chain species are destroyed and molecules like ammonia begin to be formed. If the waves propagate along the ridge from the IRAS source through AP to CP, releasing mantles as they go, the CP region will be younger than the AP in chemical terms, and the observed gradients in molecular abundance are reproduced. There is circumstantial evidence for the propagation of MHD waves in TMC-1. Firstly, they could be responsible for the observed superthermal molecular line widths (Arons & Max 1975), and furthermore, the dissipation length of the waves, which is the length over which chemistry is affected, is similar to the clump sizes observed in CCS by Peng et al. (1998).

Recently, the model was augmented to include deuterium chemistry to address the fractionation gradient issue, and to treat more completely the effect of the wave on the cloud chemistry (Markwick et al. 2001, hereafter Paper I). In particular, the small ion-neutral drift induced by the passing wave enhances the rates of ion-neutral reactions, enabling small energy barriers to be overcome (Charnley 1998). Some of the most affected reactions are fundamental to molecular deuteration, for example the reaction H₂D⁺ + H₂ → H₃⁺ + HD. This reaction is usually

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ineffective at 10 K, but becomes around 10^6 times faster as the wave passes, effectively switching off deuteration for a time. This has a devastating effect on the fractionation ratios of the major deutrating ions, thereby affecting the fractionation of all species. After the wave passes, the above reaction switches off and deuteration proceeds normally, ultimately recovering to its pre-wave value. In Paper I, the passage of the Alfvén wave was shown to be capable of causing the molecular abundance *and* the fractionation gradients.

Previously, however, the comparison of the model with observations was limited to the molecules for which spatial information was available at the time; DCO⁺, C₃HD, DC₃N and DNC. Here, we present measurements of the CH₂DCCH/CH₃CCH ratio at various positions along the ridge between the cyanopolyne peak and the ammonia peak for comparison with the model.

2. Observations

The observations were made in December 2000 with the Onsala 20 m Space Observatory in Sweden. The lines we observed are shown in Table 1. CH₂DCCH has been observed previously at TMC-1:CP by Gerin et al. (1992), and we chose to observe one of the same lines, namely the 6₀₆–5₀₅ line at 97.081 GHz. For CH₃CCH, we chose the lines observed by Pratap et al. (1997) in their comprehensive study of TMC-1. These are the 6₀–5₀ and 6₁–5₁ lines at 102.548 and 102.546 GHz respectively. Because of the long integration times involved to observe CH₂DCCH (around 6 hours per point), we were only able to obtain column densities for CH₂DCCH at three positions. We obtained column densities for CH₃CCH at 6 positions. The observations were made with the SIS 100 GHz receiver. The beam size is 40". The backend was the low resolution correlator (LRC) with a bandwidth of 20 MHz and 1600 channels, giving a spectral resolution of 12.5 kHz. For the CH₃CCH observations, we used frequency switching with a throw of 4 MHz and for CH₂DCCH we used beam switching, where the throw is 11'. The pointing errors were typically 2" rms on each axis, determined by observing the IK Tau SiO maser emission. The system temperature was 400 K on average, although on 2 nights the weather was too bad to observe. We had enough time to determine the fractionation of CH₃CCH in 3 positions; TMC-1:CP and two further points, the most distant from CP being about 60% of the way towards the ammonia peak.

3. Data analysis

The observed spectra for three positions are shown in Fig. 1. In the data reduction, we averaged the individual scans over position, weighting them according to system temperature. We have removed only linear baselines, and the spectra are shown unsmoothed.

Deriving column densities from these observations requires knowledge of the rotational temperature of the molecules. For this, we use the estimates given in

Pratap et al. (1997), which were derived from methyl acetylene, and which were shown to be consistent with adopting a temperature of 10 K along the ridge from CP to AP. We therefore take the rotational temperature as 10 K at all positions. In practice, the CH₂DCCH/CH₃CCH abundance ratio is not sensitive to the assumed temperature (Gerin et al. 1992).

For an optically thin line, the relationship between its integrated area $T_b\Delta v$ and the total column density N is

$$T_b\Delta v f(T) = \frac{8\pi^3}{3hQ} \mu^2 S_{ul} T g_K g_I N \exp^{-E_u/kT} \left(\exp^{h\nu/kT} - 1 \right).$$

Here, μ is the electric dipole moment, g_K and g_I are the state degeneracies and Q is the partition function. We took $\mu = 0.78D$, and obtained the quantities Q and $S_{ul}g_K g_I$ from tabulated data in the JPL catalogue (Pickett et al. 1998). The factor $f(T)$ is given by

$$f(T) = \frac{kT}{h\nu} \left[\left(e^{h\nu/kT} - 1 \right)^{-1} - \left(e^{h\nu/kT_{bg}} - 1 \right)^{-1} \right]^{-1}$$

where $T_{bg} = 2.73$ K. The values of the molecular line parameters are given in Table 1. The column densities derived for each line at each position we observed are shown in Table 2, along with the integrated intensities and other observational data.

3.1. Ortho/para ratios

Both the species we observed exist in ortho and para forms. For CH₃CCH, we have observed lines with $K = 0$ (A symmetry state) and $K = 1$ (E symmetry state), which correspond to ortho and para respectively. Column 6 in Table 2 lists the total column densities in each molecule assuming the statistical ortho/para ratio of 3:1. The fact that these numbers are different for the two lines of CH₃CCH is indicative of the actual ortho/para ratio being different from this. Therefore, we have used the observations of the two CH₃CCH lines to determine the ortho/para ratio in this species at the various positions observed. This is presented in Table 3. The ratio is found to vary between 4:3 and 2:1. For CH₂DCCH, Table 2 also shows the total column density assuming 3:1. Since we have observed only the $K = 0$ (para) line, we have assumed the ortho/para ratio for this species to be the same as that of CH₃CCH at the same position. The result of this adjustment is shown in Table 4.

4. Discussion

CH₂DCCH has been observed previously at TMC-1:CP by Gerin et al. (1992), who obtained a fractionation ratio of between 0.05 and 0.06 depending on the kinetic temperature used – a relatively high ratio (compare DCO⁺ \sim 0.013). In Paper I, we presented predictions for the spatial variation of the fractionation ratio of several molecules, including methyl acetylene, which was found to rise quite steeply along the ridge compared to other

Table 1. Molecular line data.

Species	Transition	ν_{rest} (GHz)	E_u (K)	$Q(10)$	S_{uigKI}	$f(T)$
CH ₂ DCCH	6 ₀₆ –5 ₀₅	97.080695	16.3	62.56	6.000	1.467
CH ₃ CCH	6 ₀ –5 ₀	102.547983	17.2	38.01	6.001	1.477
CH ₃ CCH	6 ₁ –5 ₁	102.546023	24.4	38.01	5.834	1.477

Table 2. Observational data. The offsets are in seconds of arc relative to TMC-1:CP. N is the total column density in each molecule, assuming the statistical ortho/para ratio of 3:1.

Position	Species	T_{mb} (K)	Δv (km s ⁻¹)	$T_{\text{mb}}\Delta v$ (K km s ⁻¹)	N (10 ¹³ cm ⁻²)
(0, 0)	CH ₂ DCCH	0.20	0.41	0.087	1.61 ± 0.30
	CH ₃ CCH $K = 0$	1.35	0.44	0.640	7.39 ± 0.70
	CH ₃ CCH $K = 1$	1.25	0.45	0.600	14.6 ± 1.37
(-40, +90)	CH ₃ CCH $K = 0$	1.42	0.32	0.484	5.59 ± 0.57
	CH ₃ CCH $K = 1$	1.31	0.31	0.427	10.4 ± 1.26
(-80, +150)	CH ₂ DCCH	0.34	0.25	0.091	1.68 ± 0.30
	CH ₃ CCH $K = 0$	1.32	0.32	0.453	5.23 ± 0.60
	CH ₃ CCH $K = 1$	1.25	0.36	0.484	11.8 ± 1.19
(-120, +200)	CH ₃ CCH $K = 0$	1.71	0.26	0.474	5.47 ± 0.57
	CH ₃ CCH $K = 1$	1.73	0.25	0.456	11.1 ± 1.17
(-160, +240)	CH ₂ DCCH	0.34	0.24	0.088	1.63 ± 0.35
	CH ₃ CCH $K = 0$	1.21	0.27	0.415	4.79 ± 0.59
	CH ₃ CCH $K = 1$	1.23	0.23	0.305	7.44 ± 1.19
(-200, +310)	CH ₃ CCH $K = 0$	0.67	0.34	0.242	2.79 ± 0.39
	CH ₃ CCH $K = 1$	0.64	0.26	0.181	4.41 ± 0.80

Table 3. Ortho/para ratios, determined from the CH₃CCH data. The offsets are in seconds of arc relative to TMC-1:CP.

Offset (")	o/p ratio
(0, 0)	1.52
(-40, +90)	1.61
(-80, +150)	1.33
(-120, +200)	1.48
(-160, +240)	1.93
(-200, +310)	1.90

species (see Paper I, Table 7), a prediction confirmed by the observations presented here.

Table 4 shows the total column densities, adjusted for the derived ortho/para ratio, and the calculated CH₂DCCH/CH₃CCH abundance ratios. The ratio we obtained at TMC-1:CP is 0.11, a factor of 2 higher the ratio obtained by Gerin et al. (1992). A comparison of the derived column densities at TMC-1:CP in each case shows that for a rotational temperature of 10 K, Gerin et al. found 1.7×10^{14} and 9.2×10^{12} cm⁻² for CH₃CCH and CH₂DCCH respectively, giving a ratio of 0.054. On the other hand, we find column densities of 9.18×10^{13} and 1.01×10^{13} cm⁻² respectively, giving 0.110. The column density of CH₃CCH observed recently at TMC-1:CP by Pratap et al. (1997) is 8.1×10^{13} cm⁻². The level of

Table 4. Column density determinations and levels of fractionation observed. The offsets are in seconds of arc relative to TMC-1:CP.

Offset (")	R	$N_{\text{CH}_2\text{DCCH}}$ (10 ¹³ cm ⁻²)	$N_{\text{CH}_3\text{CCH}}$ (10 ¹³ cm ⁻²)
(0, 0)	0.11 ± 0.03	1.01 ± 0.19	9.18 ± 0.87
(-40, +90)	-	-	6.80 ± 0.69
(-80, +150)	0.14 ± 0.05	0.98 ± 0.17	6.87 ± 0.79
(-120, +200)	-	-	6.87 ± 0.72
(-160, +240)	0.22 ± 0.04	1.19 ± 0.26	5.45 ± 0.67
(-200, +310)	-	-	3.19 ± 0.45

fractionation we find in CH₂DCCH is the largest observed at TMC-1:CP.

Gerin et al. also placed an upper limit on the column density of CH₃CCD. We chose not to observe this molecule, due to the long integration times we estimated based on this previous search. However, in light of the results presented here, it appears that another search for this molecule is necessary.

4.1. Chemistry of CH₂DCCH

Theoretical models of deuterium fractionation form CH₂DCCH through the CH₂D⁺ ion, by dissociative recombination of CH₂DC₂H₂⁺ or CH₂DC₃H₂⁺ ions, which

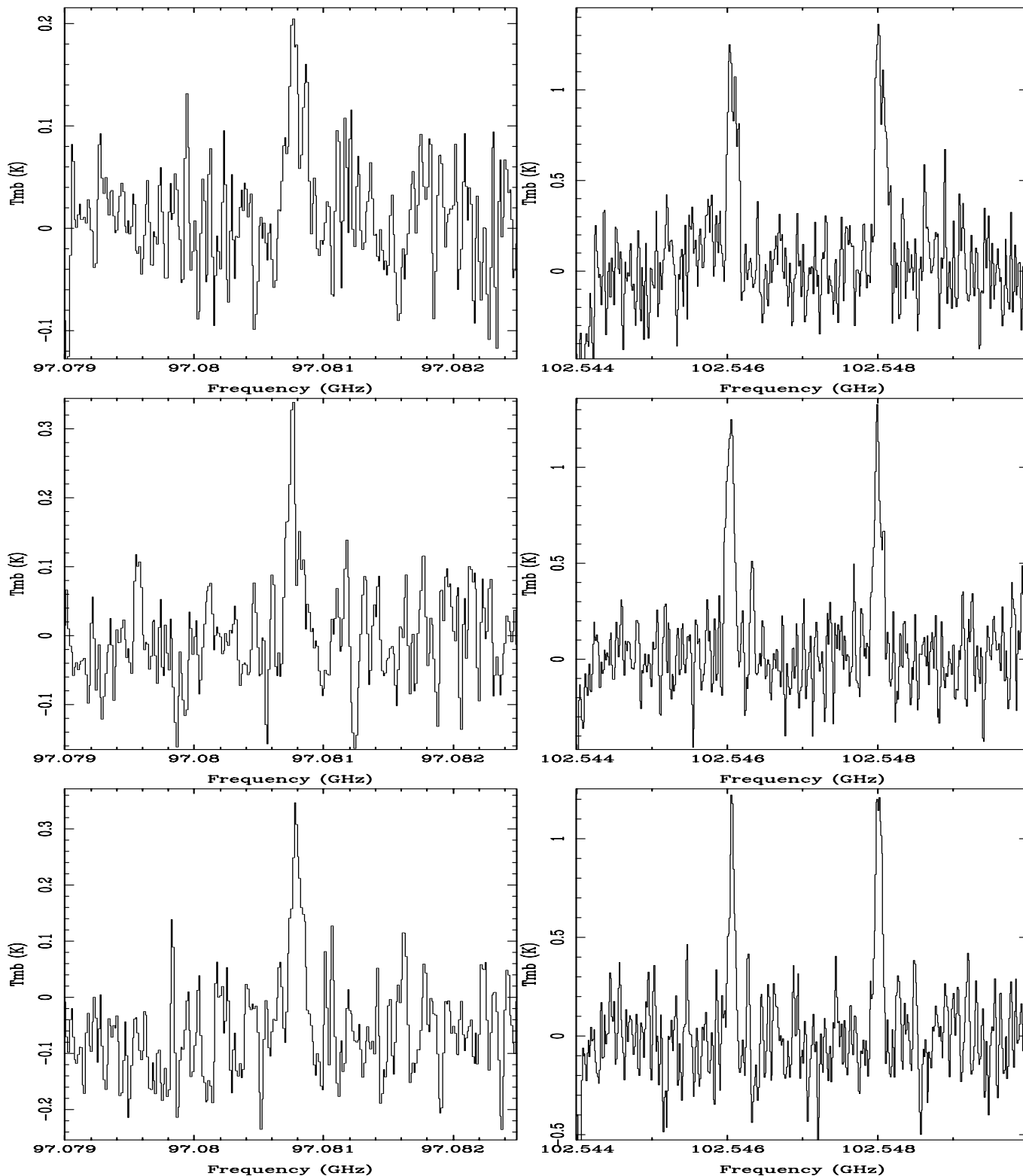


Fig. 1. Observed spectra at offsets (0, 0), (−80, +150) and (−160, +240) from TMC-1:CP (top to bottom). The left panels show CH₂DCCH 6₀₆–5₀₅ and the right panels CH₃CCH 6₀–5₀ and 6₁–5₁. The spectra are unsmoothed, and linear baselines have been removed.

are themselves formed from CH₂D⁺ by ion-neutral reactions with smaller hydrocarbon species like methane or acetylene. CH₂DCCH is destroyed by atomic and molecular ions, primarily H₃⁺ and He⁺. Since experimental data

is scarce for deuterium reactions of importance in the ISM, a deuterium chemistry is usually constructed with statistical analogues. There are two assumptions in this method which are important for the abundance of CH₂DCCH.

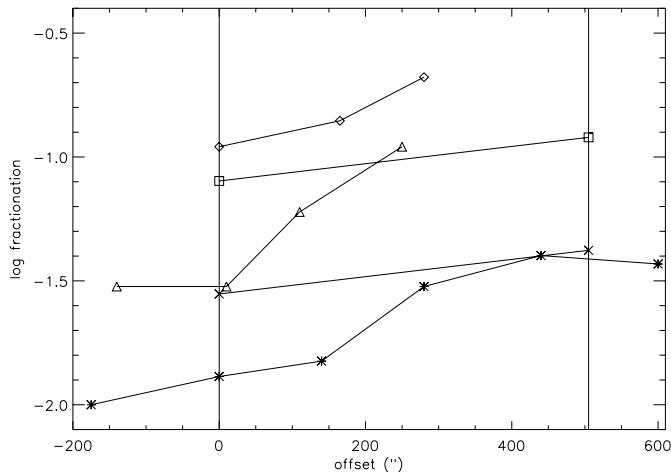
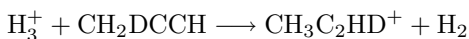
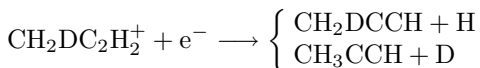


Fig. 2. Observed spatial variations of the fractionation in DC₃N (triangles; Howe et al. 1994), C₃HD (squares; Bell et al. 1988), DCO⁺ (stars; Butner & Charnley 1997, 2001), DNC (crosses; Hirota et al. 2001) and CH₂DCCCH (diamonds; this work) along the TMC-1 ridge. The vertical lines mark the positions of the CP and AP.

First, a deuteron cannot change site in the same molecule in the same reaction, e.g. the reaction



is not allowed. Second, any branching ratios are statistical, that is, the reactions



will have rates $\frac{3}{5}k$ and $\frac{2}{5}k$ respectively, where the analogue reaction rate is k .

If either of these assumptions are wrong, the amount of CH₂DCCCH produced in the models will change. There is already some evidence from storage ring experiments that dissociative recombination reactions preferentially form deuterated species (Jensen et al. 2000). Where experimental data exists, it is put into the reaction schemes, but there are still many reactions (like those above) which are assumed to have statistical branching ratios and as such are possible sources of error. In this case, the ratio CH₂DCCCH/CH₃CCH will be enhanced if dissociative recombination prefers to form CH₂DCCCH.

Current gas-phase deuterium chemistries place the ratio CH₂DCCCH/CH₃CCH at 0.070 (Markwick et al. 2001). Roberts & Millar (2000) get 0.099, but their model excludes CH₃CCD as a separate species, so this is an overestimate of the true ratio. These numbers are in reasonably good agreement with the observed ratio of 0.11 at the CP, but are in worse agreement as we move away from this point. There are now a few species whose observed fractionation ratios are higher than gas-phase models can predict, the others being C₃HD and DC₃N. While there has been considerable interest recently in the detection of doubly deuterated species and their production in chemical models, it is worth noting that there remain species for

which the level of single deuteration remains unexplained by these models.

4.2. Fractionation gradients

Since in this paper we are primarily interested in the spatial variations of the fractionation, we plot in Fig. 2 the fractionation ratios of a number of molecules as a function of position along the ridge. In Paper I, we noted that while we could explain the spatial increase in fractionation along the ridge between the CP and AP, we generally underestimated the size of each increase. Again, we find this to be the case for CH₃CCH. The observations presented here give us a value of 0.28 for the increase in log fractionation along the ridge. The value predicted in Paper I is 0.27, which is almost the same as the observed value, but this prediction is for the whole CP-AP distance, whereas we have observed to a point roughly 60% of the way (see Fig. 2). If the fractionation keeps increasing towards the AP, as it does for the other observed molecules, then our predicted value is an underestimate of the true gradient. We found this to be the case for the previously observed species as well, and therefore it appears that our model is neglecting a process or suppressing a mechanism which could increase the fractionation further. In Paper I we speculated that this could be the accretion of gas phase species back onto grain surfaces. As the timescale for accretion is comparable to our assumed age difference between the CP and AP ($\sim 2 \times 10^5$ yr), we might expect to see the effects of accretion between the peaks. Another more interesting possibility that we have not considered yet is periodic MHD waves sweeping the gas in TMC-1. It is not obvious what effect this will have on the deuterium chemistry. These topics will therefore be the subject of a future paper.

4.3. Why this model?

In Paper I we noted that there are different ways to increase the level of fractionation in a chemical model – differences in the density of the gas or differential accretion, for example. Why is this model to be preferred over the other, simpler solutions? To answer this question, consider Fig. 2, which shows the spatial position along the TMC-1 ridge of measurements of deuterated molecular species. The curious thing here is that between TMC-1:CP and AP, the level of fractionation increases *monotonically*. Taking DC₃N as an example, there are two points away from TMC-1:CP where the fractionation is greater. For DCO⁺ there are 3. In this work, we show that for CH₂DCCCH there are 2. For the other species observed (C₃HD, DNC), there is 1. With two exceptions, none of these observations are at the same position. On the basis of the available data, one is forced to conclude that deuterium fractionation increases monotonically from TMC-1:CP to AP.

Now if we wanted to use a different argument to explain the existence of monotonic fractionation gradients, say, differential depletion due to density differences, we would then have to explain why the *density* varies monotonically along the ridge. Alternatively, if we assume that we actually have a distribution of densities and have just happened to observe monotonic gradients by chance, we find that there is a less than one percent chance of getting Fig. 2.

The monotonic fractionation gradients produced by our model are a natural consequence of the Alfvén wave traversing the ridge and resetting the clock, making TMC-1:CP less advanced chemically than AP. Without this feature, it is difficult to see how the spatial variations could be systematic.

We note also that there are other recent observations of TMC-1 which are being interpreted as due to Alfvén wave activity (Dickens et al. 2000).

5. Conclusion

We have determined the level of deuterium fractionation in CH₃CCH at three positions along the TMC-1 ridge between the cyanopolyne peak and ammonia peak. We find that the fractionation ratio is large, from 0.11 to 0.22 in the three positions observed. Gas-phase models predict a CH₂DCCH/CH₃CCH ratio of between 0.07 and 0.10 at TMC-1:CP, in reasonable agreement with our observation. These models will, however, fail to produce the higher ratios we find away from the CP, a discrepancy which could possibly be due to the assumption of statistical branching ratios in deuterium fractionation chemical modelling. The high level of fractionation in CH₂DCCH also suggests that another search for CH₃CCD would be successful.

The observations show that there is a positive gradient in CH₂DCCH/CH₃CCH along the TMC-1 ridge from CP to AP, as predicted in Paper I, and therefore support the hypothesis that the chemical evolution of molecular gas in TMC-1 has been affected by Alfvén waves.

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