

Testing grain-surface chemistry in massive hot-core regions[★]

S. E. Bisschop¹, J. K. Jørgensen², E. F. van Dishoeck¹, and E. B. M. de Wachter¹

¹ Leiden Observatory, Leiden University, PO Box 9513, 2300 RA Leiden, The Netherlands
e-mail: bisschop@strw.leidenuniv.nl

² Harvard-Smithsonian Center for Astrophysics, 60 Garden Street MS42, MA 02138, USA

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ABSTRACT

Aims. We study the chemical origin of a set of complex organic molecules thought to be produced by grain surface chemistry in high mass young stellar objects (YSOs).

Methods. A partial submillimeter line-survey was performed toward 7 high-mass YSOs aimed at detecting H₂CO, CH₃OH, CH₂CO, CH₃CHO, C₂H₅OH, HCOOH, HNCO and NH₂CHO. In addition, lines of CH₃CN, C₂H₅CN, CH₃CCH, HCOOCH₃, and CH₃OCH₃ were observed. Rotation temperatures and beam-averaged column densities are determined. To correct for beam dilution and determine abundances for hot gas, the radius and H₂ column densities of gas at temperatures >100 K are computed using 850 μm dust continuum data and source luminosity.

Results. Based on their rotation diagrams, molecules can be classified as either cold (<100 K) or hot (>100 K). This implies that complex organics are present in at least two distinct regions. Furthermore, the abundances of the hot oxygen-bearing species are correlated, as are those of HNCO and NH₂CHO. This is suggestive of chemical relationships within, but not between, those two groups of molecules.

Conclusions. The most likely explanation for the observed correlations of the various hot molecules is that they are “first generation” species that originate from solid-state chemistry. This includes H₂CO, CH₃OH, C₂H₅OH, HCOOCH₃, CH₃OCH₃, HNCO, NH₂CHO, and possibly CH₃CN, and C₂H₅CN. The correlations between sources implies very similar conditions during their formation or very similar doses of energetic processing. Cold species such as CH₂CO, CH₃CHO, and HCOOH, some of which are seen as ices along the same lines of sight, are probably formed in the solid state as well, but appear to be destroyed at higher temperatures. A low level of non-thermal desorption by cosmic rays can explain their low rotation temperatures and relatively low abundances in the gas phase compared to the solid state. The CH₃CCH abundances can be fully explained by low temperature gas phase chemistry. No cold N-containing molecules are found.

Key words. astrochemistry – line: identification – methods: observational – stars: formation – ISM: abundances – ISM: molecules

1. Introduction

A characteristic stage of high-mass star formation is the hot core phase where high abundances of complex molecules are found in the gas-phase and in the ice mantles around the dust grains in the cold extended envelope (e.g., Millar 1996; Ikeda et al. 2001; Boogert & Ehrenfreund 2004). The existence of both ices and gas-phase molecules around the same high-mass protostars suggests that the envelopes have temperatures that range from >200 K in the inner region to as low as 10 K in the cold outer part (e.g., van der Tak et al. 2000; Gibb et al. 2000b; Ikeda et al. 2001). Together the submillimeter and infrared data indicate that there is a rich chemistry which involves interactions between gas and grain-surface species. The precise origin of complex molecules is still heavily debated: are they “first generation” species formed on the grains during the cold collapse phase (e.g., Tielens & Charnley 1997), or “second generation” species produced by high-temperature gas phase reactions following ice evaporation (e.g., Millar et al. 1991; Charnley et al. 1992; Charnley 1995)? The aim of this work is to test the formation schemes of key complex organic molecules by presenting submillimeter observations of a sample of high-mass cores using the same instrument and frequency settings.

High column densities of the organic molecules CH₃OH, CH₄, HCOOH, OCN⁻ and potentially H₂CO have been detected in ice mantles in the cold outer envelopes of high-mass young stellar objects (YSOs) (Gibb et al. 2004). Abundances for some ice species such as CH₃OH and OCN⁻ are large, up to 30% with respect to H₂O ice or 3×10^{-5} with respect to H₂ (Brooke et al. 1999; Dartois et al. 1999; Pontoppidan et al. 2003; van Broekhuizen et al. 2005). Information on minor species such as CH₄, HCOOH, H₂CO and CH₃CHO is more limited, but they appear to have constant abundances at the level of a few % with respect to H₂O ice (Schutte et al. 1999; Keane et al. 2001; Gibb et al. 2004). More complex species with abundances lower than 0.1% cannot be detected by infrared observations because of their small optical depth. Moreover, features of the same functional groups such as C–H and O–H groups overlap, so that they can only be assigned to a class of molecules rather than to a specific species. Deep submillimeter surveys in the gas phase are more sensitive to detect these minor species.

The abundances of the detected ice species are so large that they cannot be produced by pure gas-phase processes but must be formed on the grains. A proposed scheme for the formation of various “first generation” molecules on grains (based on Tielens & Charnley 1997) is shown in Fig. 1. Through successive addition of H, C, O, and N to CO, many complex organic species are formed. The outcome depends on the relative gas-phase

[★] Appendices are only available in electronic form at <http://www.aanda.org>

Table 2. List of observed frequency settings and molecular lines covered at the JCMT and IRAM.

Molecule	Freq. [GHz]	E_u [K]	$\mu^2 S$ [D ²]	Transition	Additional molecules
JCMT					
HCO	346.725	41.61	7.341	4 _{0,4,5,4} –3 _{0,3,4,3}	C ₂ H ₅ CN/C ₂ H ₅ OH/H ₂ C ¹⁸ O/HCOOCH ₃
	348.778	172.36	5.313	4 _{2,3,4,4} –3 _{2,2,3,3}	HCOOCH ₃ /CH ₃ ¹³ CN/C ₂ H ₅ OH
H ₂ CO	218.222	20.96	16.2985	3 _{0,3} –2 _{0,2}	CH ₃ OH/C ₂ H ₅ CN/HCOOCH ₃
	364.275	158.42	52.1647	5 _{3,3} –4 _{3,2}	C ₂ H ₅ OH
CH ₃ OH	338.344	5.3431	69.36	7 _{3,4,0} –6 _{3,4,0}	HCOOCH ₃
CH ₂ CO	240.187	88.01	71.489	12 _{1,12} –11 _{1,11}	CH ₃ OH/H ₂ C ¹⁸ O/C ₂ H ₅ CN/CH ₃ CN/HCOOCH ₃
	220.178	76.48	65.434	11 _{1,11} –10 _{1,10}	CH ₃ ¹³ CN/ ¹³ CH ₃ OH/HCOOCH ₃
	222.298	182.26	61.0830	11 _{3,9} –10 _{3,8}	CH ₃ OCH ₃ /CH ₃ CCH/HCOOCH ₃
CH ₃ CHO	226.590	71.30	77.779	12 _{0,12} –11 _{0,11}	HCOOCH ₃
C ₂ H ₅ OH	230.991	85.53	48.953	16 _{5,11} –16 _{4,12}	CH ₃ OH/C ₂ H ₅ CN
	347.916	250.88	66.314	20 _{4,17,0} –19 _{4,16,0}	C ₂ H ₅ CN/HCOOCH ₃ /H ₂ C ¹⁸ O
	339.979	57.89	17.724	9 _{4,6} –8 _{3,5}	C ₂ H ₅ CN
HNCO	219.798	58.02	28.112	10 _{0,10,11} –9 _{0,9,10}	H ₂ ¹³ CO/C ₂ H ₅ CN
	352.898	187.25	43.387	16 _{1,15,17} –15 _{1,14,16}	C ₂ H ₅ CN/C ₂ H ₅ OH/HCOOCH ₃
HN ¹³ CO	240.881	112.53	30.4313	11 _{1,11,12} –10 _{1,10,11}	CH ₃ OCH ₃ /CH ₃ OH/HNCO
NH ₂ CHO	339.904	245.79	539.342	16 _{6,11} –15 _{6,10}	same setting as C ₂ H ₅ OH
	340.491	165.59	605.520	16 _{3,14} –15 _{3,13}	same setting as CH ₂ CO
	345.183	151.59	664.219	17 _{0,17} –16 _{0,16}	HCOOCH ₃ /C ₂ H ₅ OH/ ¹³ CH ₃ OH
HCOOH	225.238	88.02	17.734	10 _{3,8} –9 _{3,7}	H ₂ CO/HCOOCH ₃ /CH ₃ OCH ₃ / ¹³ CH ₃ OH
	338.144	180.53	27.1548	15 _{4,12} –14 _{4,11}	CH ₃ OH/C ₂ H ₅ OH
	346.719	143.92	28.726	15 _{2,13} –14 _{2,12}	same setting as HCO
	356.137	158.67	30.6650	16 _{2,15} –15 _{2,14}	NH ₂ CHO/HCOOCH ₃ /C ₂ H ₅ CN/H ₂ ¹³ CO/CH ₃ OH
IRAM					
CH ₂ CO	81.586	22.86	22.4974	4 _{1,3} –3 _{1,2}	
	140.127	39.97	41.1295	7 _{1,7} –6 _{1,6}	
	244.712	89.42	71.4841	12 _{1,11} –11 _{1,10}	
	262.619	140.63	25.3788	13 _{2,12} –12 _{2,11}	
CH ₃ CHO	98.901	16.51	31.2087	5 _{1,4} –4 _{1,3}	
	112.249	21.14	37.9261	6 _{1,6} –5 _{1,5}	
	149.507	34.59	51.1868	8 _{1,8} –7 _{1,7}	
	168.093	42.66	57.7848	9 _{1,9} –8 _{1,8}	
HCOOH	223.916	71.93	18.7011	10 _{2,9} –9 _{2,8}	
	247.514	150.71	17.0080	11 _{5,6/7} –10 _{5,5/6}	
	257.975	83.91	23.2080	12 _{1,12} –11 _{1,11}	
	262.103	82.77	23.2998	12 _{0,12} –11 _{0,11}	

2.2. Observed frequency settings

The observed frequencies are shown in Table 2 and were selected based on whether they contained at least one strong line for a target molecule as well as lines of other interesting species. Strong lines of target molecules were chosen because of their high main-beam temperatures and minimum line confusion in line surveys of Orion A by Sutton et al. (1985) and Orion KL by Schilke et al. (1997) at 230 GHz and 345 GHz respectively. The aim was to cover at least two transitions for each molecule to determine rotation temperatures: one with $E_u > 100$ K, and one with $E_u < 100$ K. Table 2 shows the additional lines present in the rest frequency side band. The molecular species detected in the image band differed, because it varied per source. Finally, for H₂CO, CH₃OH, and HNCO lines of the isotopic species were observed to determine the optical depth of the main isotopologues.

An additional number of settings have been observed for CH₂CO, CH₃CHO, and HCOOH with the Institut de Radioastronomie Millimétrique² (IRAM) 30 m telescope for

AFGL 2591, G75.78, NGC 7538 IRS1, and W 3(H₂O). The settings were chosen because of their high line strengths and low excitation temperatures. Combined with the JCMT settings, they allow for the determination of any cold component of these molecules.

2.3. JCMT observations

The frequencies given in the first part of Table 2 were observed for the 7 sources with the JCMT on Mauna Kea, Hawaii from August 2004 to January 2006. The beam size (θ_{beam}) is 20–21'' for observations in the 230 GHz band and 14'' in the 345 GHz band. The spectra were scaled from the observed antenna temperature scale, T_A^* , to main beam temperatures, T_{MB} , using main beam efficiencies η_{MB} of 0.69 and 0.63 at 230 GHz and 345 GHz, respectively. The front-ends consisted of the facility receivers A3 and B3; the back-end was the Digital Autocorrelation Spectrometer (DAS), covering 500 MHz instantaneous bandwidth with a spectral resolution of 312.5 and 625 kHz respectively. Pointing was checked every 2 hrs or whenever a new source was observed and was always within 3''. To subtract atmospheric and instrumental backgrounds beam switching with a chop throw of 180'' was used. We aimed for a $T_{\text{rms}} \sim 20$ mK on the T_A^* scale when binned to a ~ 1 km s⁻¹

² The IRAM 30 m telescope is operated by the Institut de Radioastronomie Millimétrique on behalf of the Centre National de la Recherche Scientifique of France, the Max Planck Gesellschaft of Germany, and Instituto Geográfico Nacional of Spain.

channel, which was obtained with integration times of ~ 1 hr at 230 GHz and 2 hrs at 345 GHz dependent on weather. For NGC 6334 IRS1 the confusion limit was reached after 30 minutes of integration, however, at which point the observation was terminated. The absolute calibration is accurate to better than 15% from comparison to spectral line standards.

The JCMT B3 345 GHz receiver was operated in single side band mode to minimize line confusion. The A3 230 GHz receiver does not have the option of image sideband suppression. For line-rich sources like Orion (Sutton et al. 1985), it is difficult to assign a line to the correct sideband. For our sources the 230 GHz spectra are not confusion limited except for NGC 6334 IRS1 (see Fig. 3). For NGC 6334 IRS1, pairs of spectra with 10 MHz offsets were taken. This shifts lines in the image sideband by 20 MHz with respect to those observed in the rest sideband making a unique identification possible.

2.4. IRAM observations

The frequencies given in the second part of Table 2 have been observed with the IRAM 30 m telescope. The beam size (θ_{beam} in arcsec) is 29'' at 86 GHz, 17'' at 140 GHz, and 10.5'' at 235 GHz. The front-ends were the B100, C150, B230, and C270 facility receivers. The VESPA auto-correlator was used as the back-end, with a bandwidth of 120 MHz, and a channel spacing of 80 kHz. Pointing was checked every 1.5 hrs and was always within 2–3''. Atmospheric and instrumental backgrounds were subtracted through “wobbler” switching where the telescope alternately observes positions +220'' and –220'' away from the source. Integration times differed per source and depended on the atmospheric conditions, but were generally 30 min for AFGL 2591 and G75.78, and 1 hr for NGC 7538 IRS1 and W 3(H₂O). All observations were converted to the main-beam temperature scale using $F_{\text{eff}}/B_{\text{eff}}$ of 0.78 at 86 GHz to 0.46 at 260 GHz. Conveniently, all observations were performed in single-sideband mode.

3. Data analysis

3.1. Rotation diagrams

Line assignments were made by comparing the observed frequencies corrected for source velocity with the JPL³, CDMS⁴ and NIST⁵ catalogs. Possible assignments had to be within ~ 1 MHz, unless the uncertainty of the frequencies in the catalogs was larger. For NH₂CHO, for example, lines were only assigned if multiple transitions were detected in the same spectrum.

Integrated intensities for the identified lines are given in Tables B.1–B.15. These were used to derive rotation temperatures and column densities through the rotation diagram method (Goldsmith & Langer 1999) when three or more lines were detected over a sufficiently large energy range. The integrated main-beam temperatures are then related to the column density in the upper energy level by:

$$\frac{N_u}{g_u} = \frac{3k \int T_{\text{MB}} dV}{8\pi^3 \nu \mu^2 S} \quad (1)$$

³ <http://spec.jpl.nasa.gov/ftp/pub/catalog/catform.html>

⁴ <http://www.ph1.uni-koeln.de/vorhersagen/>

⁵ <http://physics.nist.gov/PhysRefData/Micro/Html/contents.html>

where N_u is the column density of the upper level, g_u is the degeneracy in the upper level, k is Boltzmann’s constant, ν is the transition frequency, μ is the dipole moment, and S is the line strength. The total *beam-averaged* column density N_T in cm⁻² can then be computed from:

$$\frac{N_u}{g_u} = \frac{N_T}{Q(T_{\text{rot}})} e^{-E_u/T_{\text{rot}}} \quad (2)$$

where $Q(T_{\text{rot}})$ is the rotational partition function, and E_u is the upper level energy in K.

The calculated rotation temperatures and column densities can be used to make predictions for the intensities of other lines for a specific species. This was used as an independent check on line-assignments. If lines predicted to have high intensities in the observed frequency ranges are not seen, a detection is considered questionable. In the analysis, a correction has been made for the different beam-sizes at different frequencies. The assumption has been made that the maximum source size is equal to the smallest observed beam-size of the JCMT for all sources, i.e. 14''. The calculations have also been performed assuming that the emission is extended with respect to the largest JCMT beam-size of 21''. In this case, the rotation temperatures are higher and have larger errors by a factor of 2. In reality, as will be discussed in Sect. 3.2, the hot core gas will be even more severely beam-diluted. Finally, the calibration uncertainties are taken into account in the rotation diagram analysis and are shown as error-bars on the rotation diagram plots.

When only very few lines were detected or only a small range in E_u was covered, no rotation temperatures or column densities could be derived. If the detection of the species was considered real, the column density was estimated assuming the rotation temperature to be the average for the same molecule in other sources. If no lines were detected, a 3σ upper limit was determined. For lines from isotopologues, rotation temperatures equal to the main species were used to estimate the column densities.

A number of assumptions are implicitly made in the rotation diagram method. In short, they are that the excitation can be characterized by a single temperature T_{rot} and that the lines are optically thin. Even though the first assumption will almost certainly be incorrect, the rotation diagram method does give the average excitation temperature of the region from which most of the molecular emission arises. If the lines are sub-thermally excited because the density is below the critical density (n_{cr}), the rotation temperature is a lower limit. This could also be partly responsible for the scatter in the rotation diagrams. Since for most molecules studied here no collisional rate coefficients are known, a more sophisticated statistical equilibrium analysis is not possible. The critical density, n_{cr} , for the CH₃OH $7_{6,K}-6_{6,K}$ transitions is $\sim 10^6$ cm⁻³ while for the H₂CO $3_{0,3}-2_{0,2}$ transition it is $\sim 5 \times 10^6$ cm⁻³. Since n_{cr} is proportional to μ^2 and ν^3 , transitions with high values for either will be most affected. Most species in this study have dipole moments μ close to the CH₃OH value of 1.7 Debye. H₂CO, CH₃CHO, HNCO, NH₂CHO, and CH₃CN have dipole moments 2–3 times larger and for these species sub-thermal effects should thus be more important. Optical depth can significantly influence the results as well. Column densities derived from optically thick lines will be severely under- and rotation temperatures overestimated. Lines of isotopic species can be used to determine the optical depth. Alternatively, it is possible to predict which lines are optically thick based on the rotation diagram results and exclude those from the fit. In this paper both approaches have been used. A final uncertainty is infrared pumping. It is a well-studied phenomenon for HNCO

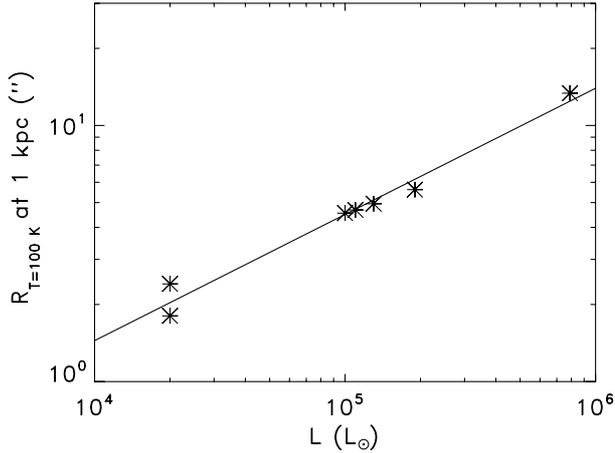


Fig. 2. Comparison of luminosity with source size where $T = 100$ K. $R_{T=100\text{ K}}$ is given in arcsec assuming all sources to have a distance of 1 kpc. The solid line represent the best fit to the data.

(Churchwell et al. 1986). Its importance for other species is unknown, although all molecules have dipole-allowed mid-infrared transitions. However, no vibrationally excited lines have been detected in this survey except for CH_3OH . With these limitations the rotation diagram method gives a useful indication of the temperature region where the abundance of a molecule peaks.

3.2. Beam dilution correction and hydrogen column density in hot gas

If the emission from the complex organic species comes from a warm region where the temperature is higher than the ice evaporation temperature of 100 K, it becomes important to properly take the amount of beam dilution into account, especially for inter-comparison of the sources. The region where the temperature is above 100 K is expected to be small ($\sim 1''$) compared to the size of the single-dish beam, and is related to the luminosity L . Also, the beam-dilution depends on the distance as d^2 . To correct for the amount of beam dilution we established a set of models for the dust envelopes of our sources based on SCUBA 850 μm data⁶. For simplification we fix the inner radii of the envelopes to 500 AU, the outer radii to 100 000 AU (≈ 0.5 pc) and adopt a power-law density profile, $n \propto r^{-1.5}$, typical of a free-falling core. r_0 is chosen to be so large that gas extending beyond this radius will not significantly contribute. We assume that the total luminosity quoted in Table 1 is provided by a single 30 000 K blackbody at the center of the envelope and fit the observed 850 μm fluxes of each source by adjusting the mass of the envelope using the DUSTY dust radiative transfer code (Ivezić et al. 1999) as described in Jørgensen et al. (2002). The results of these models provide a temperature profile for each source from which the radius where the temperature increases above 100 K ($R_{T=100\text{ K}}$), and the beam-dilution can be estimated (see Table 3). Figure 2 compares the inferred source sizes of the warm gas in arcsec with luminosity of the sources with all distances set to 1 kpc. These correlate very strongly with luminosity as expected. Our results are consistent with the independently derived relation from dust modeling of a large range of sources, $R_{T=100\text{ K}} \approx 2.3 \times 10^{14} (\sqrt{L/L_\odot})$ cm (Doty private

communication). In fact, the luminosity itself can be used to estimate the linear size of the region where the temperature is higher than 100 K. We therefore adopt these radii for the beam dilution correction for molecules that unambiguously probe warm ≥ 100 K gas estimated from the rotation diagrams. The derived beam dilution (η_{bf}),

$$\eta_{\text{bf}} = \frac{(R_{T=100\text{ K}}/d)^2}{(\theta_{\text{beam}}/2)^2 + (R_{T=100\text{ K}}/d)^2} \quad (3)$$

is then used to calculate *source-averaged* column densities ($N_{T \geq 100\text{ K}}$) of hot molecules (X) in Sect. 4.4:

$$N_{T \geq 100\text{ K}}(\text{X}) = \frac{N_{\text{T}}(\text{X})}{\eta_{\text{bf}}} \quad (4)$$

The radiative transfer models used to calibrate the $R_{T=100\text{ K}} - L$ relation can also be used to calculate the hydrogen mass in the inner warm regions and thereby provide an H_2 column density to determine molecular abundances for $T \geq 100$ K. With the derived density and temperature profiles listed in Table 3, the number of H_2 molecules, $Mol(\text{H}_2)$, within $R_{T=100\text{ K}}$, can be calculated from:

$$\begin{aligned} Mol(\text{H}_2) &= \int_{r_i}^{R_{T=100\text{ K}}} 4\pi r^2 n_i \left(\frac{r}{r_i}\right)^{-\alpha} dr \\ &= \frac{4\pi n_i r_i^\alpha}{3 - \alpha} \left(R_{T=100\text{ K}}^{3-\alpha} - r_i^{3-\alpha}\right), \end{aligned} \quad (5)$$

where n_i is the density of H_2 molecules at the inner radius, and α the power of the density profiles. The column density of H_2 in gas with $T \geq 100$ K, $N_{T \geq 100\text{ K}}(\text{H}_2)$ in cm^{-2} is then simply $Mol(\text{H}_2)$ divided by $\pi R_{T=100\text{ K}}^2$

$$N_{T \geq 100\text{ K}}(\text{H}_2) = \frac{4n_i r_i^\alpha}{(3 - \alpha)R_{T=100\text{ K}}^2} \left(R_{T=100\text{ K}}^{3-\alpha} - r_i^{3-\alpha}\right). \quad (6)$$

Since both $N_{T \geq 100\text{ K}}(\text{H}_2)$ and $N_{T \geq 100\text{ K}}(\text{X})$ refer to column densities for the same temperature region, they can be used directly to calculate abundances for a species X in the hot gas through:

$$[\text{X}] = \frac{N_{T \geq 100\text{ K}}(\text{X})}{N_{T \geq 100\text{ K}}(\text{H}_2)}. \quad (7)$$

It is clear from Table 3 that the derived values of $R_{T=100\text{ K}}$ and $N_{T \geq 100\text{ K}}(\text{H}_2)$ are rather similar for most sources with the exception of AFGL 2591 and G24.78. AFGL 2591 has a significantly smaller and G24.78 a much larger radius and H_2 column density. These are related to their low and very high luminosities. The beam-dilution for G24.78 is thus very similar to that for the other sources in our sample. This makes a good comparison between G24.78 and the other sources possible.

Some of our sources have previously been modeled in more detail (e.g., van der Tak et al. 2000; Mueller et al. 2002; Hatchell & van der Tak 2003); these models show some differences, e.g., in the slope of the density distribution from source to source and even for the same source depending on method. As an example, van der Tak et al. (2000) found that $R_{T=100\text{ K}}$ is 5.3×10^3 AU for NGC 6334 IRS1 and 2.8×10^3 AU for W 3(H_2O) with a density slope $\alpha = 2$. Both values are within 10–15% of our model result. Together with the good agreement with the models by Doty (private communication) this implies that the model-dependent uncertainty has a rather small effect on $R_{T=100\text{ K}}$ and the column densities. We therefore consider our adopted approach to provide the internally most consistent set of estimates – even though we

⁶ The SCUBA 850 μm data were retrieved through the JCMT archive as a Guest User at the Canadian Astronomy Data Center, which is operated by the Dominion Astrophysical Observatory for the National Research Council of Canada's Herzberg Institute of Astrophysics.

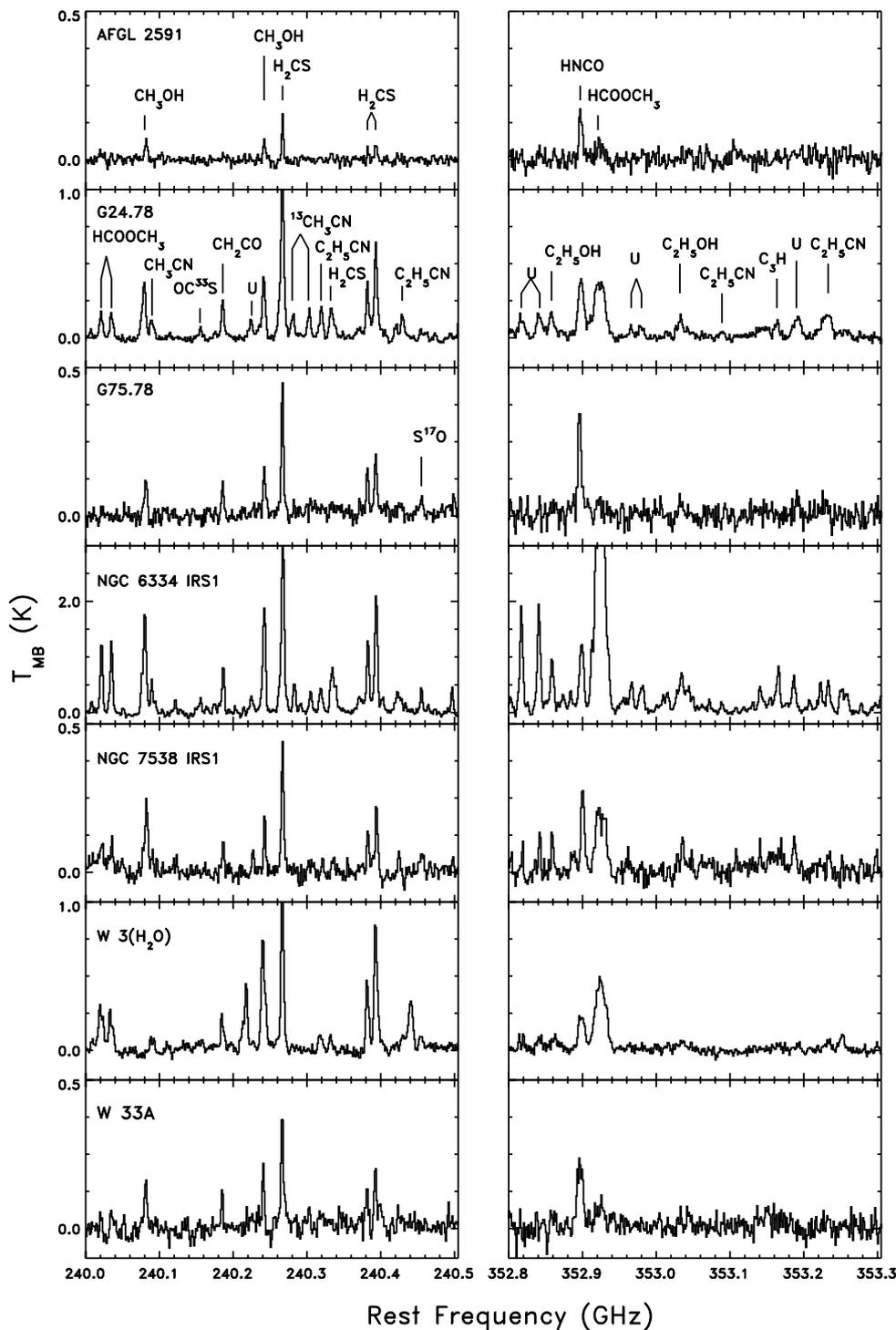


Fig. 3. Spectra centered at 240.25 and 353.05 GHz for all sources. Note that the T_{MB} scales are different for individual sources.

caution that a full self-consistent set of models (explaining, e.g., both single-dish and interferometric continuum data) is required to estimate abundances to better than a factor 2–3.

Finally, interferometric observations have shown that even on scales as small as a few thousand AU, there may be strong chemical differentiation, e.g., oxygen-bearing molecules peaking at a different position than nitrogen-bearing species (Liu 2005; Blake et al. 1996, 1987; Wyrowski et al. 1999; Sutton et al. 1995). From single-dish observations of a single source, it is not possible to determine whether the emission of various species arises from different regions. A comparison of the abundances

of the species studied in this paper for a number of sources does, however, show whether the emission likely comes from the same or a different region (Sect. 5).

4. Results

4.1. General comparison between sources

Figure 3 shows the spectra at 240.25 and 353.05 GHz toward all sources and Tables B.1–B.15 contain all measured line intensities for the molecules discussed in this paper. There is a clear

Table 3. Parameters of temperature and density profiles, and beam-filling factors derived from 850 μm SCUBA data and source luminosities.

Sources	n_i^a [cm^{-3}]	$R_{T=100\text{ K}}$ [AU]	τ^b	$N_{T \geq 100\text{ K}}(\text{H}_2)$ [cm^{-2}]	η_{bf}^c
AFGL 2591	8.5(6)	1.8(3)	0.40	7.6(22)	6.2(-2)
G24.78	1.0(8)	1.3(4)	4.8	4.0(23)	5.8(-2)
G75.78	2.1(7)	5.6(3)	0.98	1.2(23)	3.7(-2)
NGC 6334 IRS1	3.8(7)	4.7(3)	1.8	2.4(23)	1.3(-1)
NGC 7538 IRS1	3.4(7)	4.9(3)	1.6	2.1(23)	6.0(-2)
W 3(H ₂ O)	6.2(7)	2.4(3)	2.9	1.8(23)	2.9(-2)
W 33A	4.1(7)	4.5(3)	1.9	2.6(23)	2.6(-2)

^a Density at inner radius of 500 AU. ^b The optical depth τ is given for the dust at 100 μm . ^c η_{bf} is the beam-filling factor.

difference in the “richness” of the sources in molecular lines. The intensities and thus also column densities of “line-poorer” sources are lower for most organic species for sources such as AFGL 2591, G75.78, NGC 7538 IRS1, and W 33A. Some lines, such as the HCOOCH₃ 31_{4,27}–30_{4,26} at 352.922 GHz, are weak relative to the HNCO 16_{1,15}–15_{1,14} at 352.898 GHz for the “line-poorer” sources, but strong for the “line-richer” sources, i.e. G24.78, NGC 6334 IRS1, NGC 7538 IRS1, and W 3(H₂O). There therefore seems to be an intrinsic chemical difference between the “line-rich” and “line-poor” sources.

The two settings shown in Fig. 3 were aimed at detecting the CH₂CO 12_{1,12}–11_{1,11} transition at 240.187 GHz and the C₂H₅OH 21_{1,20,2}–20_{2,19,2} line at 352.858 GHz. Both transitions were detected in most sources, indicating that the selected settings are suitable for obtaining detections or otherwise limits. Many lines of other target molecules have been detected in these settings as well.

4.2. Optical depth determinations

Some species such as CH₃OH, H₂CO, and CH₃CN have previously been found to be optically thick in their main lines (see e.g., Helmich & van Dishoeck 1997). Lines for selected isotopic species were used to determine whether the lines of the main species are optically thick and to derive column densities for the main isotopologues (see Tables B.1, B.2, B.3, B.8, and B.10). For H₂CO and HNCO, the same transition could be used for different isotopologues. For CH₃CN, the optical depth was determined by comparing the column density derived from the rotation diagram analysis for the main species and that derived from the detected transition for the isotopologues. The ratios for the beam-averaged column densities are given in Table 4, and the corrected source-averaged column densities in Table 5. In the remainder of the paper a ¹²C/¹³C ratio value of 53 for the 4 kpc molecular ring has been used (Wilson & Rood 1994). Note that this value is uncertain, however, and known to depend on the distance to the galactic center. However, for simplicity we have assumed the same ratio for all sources.

The rotation diagram results by van der Tak et al. (2000) were not corrected for optical depth, because no isotopologues were observed. In our data, however, lines for H₂¹³CO and H₂C¹⁸O are observed. The optical depth of H₂CO has been determined from the 3_{1,2}–2_{1,1} transition. The observed H₂CO/H₂¹³CO ratio ranges from 6.5 for NGC 6334 IRS1 to 37 for AFGL 2591 and NGC 7538 IRS1, lower than the expected ratio, which implies that the main isotopologue is optically thick in many of its transitions. The results for W 3(H₂O) are

consistent with the optical depth found by Helmich & van Dishoeck (1997). The optical depth of H₂¹³CO was estimated based on the 5_{2,3}–4_{2,2} transition observed for both H₂¹³CO and H₂C¹⁸O. The H₂¹³CO/H₂C¹⁸O ratio is lower than the expected value of 5–8 for most sources, but it is unclear whether this difference is significant. We assume therefore for simplicity that H₂¹³CO is optically thin for all sources.

No direct comparison of the same transition is possible for CH₃OH with ¹³CH₃OH. Many detected lines have strengths that are lower by more than an order of magnitude compared to the strongest transitions. It is therefore possible to determine the column density of the main isotopic species by excluding lines predicted to be optically thick for the *source-averaged* column density from the rotation diagram. This also circumvents the additional error introduced by the uncertain isotopic ratios. As seen in Fig. A.2, this excludes mainly lines lying below the fit to the optically thin lines and results in a much better determined rotation temperature.

The limits on the HNCO/HN¹³CO ratios are smaller than ~ 53 , and HNCO could thus be optically thick for all sources. Note that the optical depth determination of HNCO is potentially complicated by the close spacing of the lines of the main species and its isotopologue of only ~ 5 MHz. However, since most of our sources have line widths of only 3–4 km s⁻¹, the upper limits for HN¹³CO are significant. Since HN¹³CO has not been detected for any of our sources and most sources studied by Zinchenko et al. (2000), HNCO is assumed to be optically thin throughout the remainder of this paper.

Since it was not one of the target molecules, CH₃CN was not observed for all sources. Its lines and those of CH₃¹³CN were coincidentally present in the image sidebands for some of the sources. For G24.78, NGC 6334 IRS1, and W 33A, both the main species and its isotopologue have been detected and CH₃CN is found to be optically thick in the 12–11 and 13–12 transitions with ratios ranging from 5.8 for G24.78 to 24 for NGC 6334 IRS1. For G24.78, this result confirms the findings by Beltrán et al. (2005).

Since HCOOCH₃ and CH₃OCH₃ have relatively high column densities, predictions were made for which lines were optically thick. All lines detected for both molecules were found to be optically thin, if we assume that the emission does not come from a radius smaller than $R_{T=100\text{ K}}$.

4.3. Rotation temperatures

In the following subsection we discuss each molecule individually and compare it with literature; the main results are summarized at the end of this subsection. The rotation temperatures were derived from single temperature fits to rotation diagrams (see Sect. 3.1). Calibration uncertainties of 15% are shown as error bars on the rotation diagrams. All rotation diagrams are shown in Appendix 7, and resulting rotation temperatures in Table 6.

H₂CO: For H₂CO, the data are taken from van der Tak et al. (2000) except for G24.78 and G75.78 (see Fig. A.1). The rotation temperatures for all sources except NGC 6334 IRS1 and W 3(H₂O) are ~ 80 K. NGC 6334 IRS1 and W 3(H₂O) have temperatures of about 100 K higher. However, as the optical depths are similar for most sources, this temperature difference between the sources is real.

CH₃OH: For CH₃OH no clear temperature trend is seen, but the temperature is generally higher than 100 K. This implies that the emission is coming from warm gas. The CH₃OH rotation

Table 4. Column density or line intensity ratios for $\text{H}_2\text{CO}/\text{H}_2^{13}\text{CO}$, $\text{H}_2^{13}\text{CO}/\text{H}_2\text{C}^{18}\text{O}$, $\text{HNCO}/\text{HN}^{13}\text{CO}$, and $\text{CH}_3\text{CN}/\text{CH}_3^{13}\text{CN}$ before correction for optical depth.

Species	Sources						
	AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W 3(H_2O)	W 33A
$\text{H}_2\text{CO}/\text{H}_2^{13}\text{CO}^a$	37	9.3	22	6.5	37	31	14
$\text{H}_2^{13}\text{CO}/\text{H}_2\text{C}^{18}\text{O}^b$	<2.4	2.8	>1.7	2.4	<1.6	>3.0	>3.5
$\text{HNCO}/\text{HN}^{13}\text{CO}^c$	>2.4	>11	>6.5	>8.6	>2.0	>3.6	>7.3
$\text{CH}_3\text{CN}/\text{CH}_3^{13}\text{CN}^d$	–	5.8	>5.9	25	7	>9.5	7.6

^a Derived from the $3_{1,2}-2_{1,1}$ transition. ^b Derived from the $5_{2,3}-4_{2,2}$ transition. ^c Derived from the $10_{0,10}-9_{0,9}$ transition. ^d Derived from the overall column density determined from all detected transitions.

Table 5. Column densities N for hot core molecules (source-averaged) and cold molecules (beam-averaged) ^a [cm^{-2}].

Species	Sources						
	AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W 3(H_2O)	W 33A
Hot molecules							
H_2CO^b	1.3(16)	6.4(16)	2.2(16)	1.3(17)	2.5(16)	1.8(17)	5.4(16)
H_2^{13}CO	2.4(14)	1.2(15)	4.0(14)	2.5(15)	4.8(14)	3.3(15)	1.0(15)
$\text{H}_2\text{C}^{18}\text{O}$	7.4(13)	4.3(14)	<2.5(14)	1.1(15)	1.3(14)	<1.7(15)	<3.0(14)
CH_3OH^c	4.7(16)	2.8(17)	1.1(17)	9.7(17)	1.2(17)	1.0(18)	2.0(17)
$^{13}\text{CH}_3\text{OH}$	<1.8(15)	2.9(16)	4.3(15)	8.5(16)	6.7(15)	3.2(16)	6.5(15)
$\text{C}_2\text{H}_5\text{OH}$	<1.0(15)	7.1(15)	<2.4(15)	1.9(16)	5.7(15)	8.4(15)	4.7(15)
HNCO	8.3(14)	5.0(15)	3.8(15)	4.3(15)	2.3(15)	4.9(15)	6.6(15)
HN^{13}CO	<4.7(14)	<9.3(14)	<1.2(15)	<3.8(14)	<1.2(14)	<2.7(15)	<1.2(15)
NH_2CHO	<1.3(15)	7.2(14)	2.0(14)	7.4(14)	5.7(14)	1.3(15)	2.1(15)
CH_3CN	<3.5(15)	5.9(16) ^d	1.8(15)	2.9(16) ^d	<8.2(15)	7.0(15)	2.7(16)
$\text{CH}_3^{13}\text{CN}$	<5.3(14)	1.1(15)	<3.0(14)	5.6(14)	<1.2(15)	<7.2(14)	<5.0(14)
$\text{C}_2\text{H}_5\text{CN}$	<7.5(14)	4.0(15)	<1.2(15)	5.1(15)	<9.2(14)	4.5(15)	<2.1(15)
HCOOCH_3	<2.4(16)	3.1(16)	7.1(15)	1.2(17)	1.4(16)	5.2(16)	2.5(16)
CH_3OCH_3	<7.7(15)	1.2(17)	2.3(16)	5.8(17)	<1.6(16)	1.5(17)	2.7(16)
Cold molecules							
CH_2CO	1.6(13)	2.1(14)	6.4(13)	7.2(14)	9.7(13)	1.1(14)	6.3(13)
CH_3CHO	3.1(12)	<1.5(13)	<2.1(13)	<1.2(14)	2.8(13)	3.5(13)	<3.0(13)
HCOOH	3.4(13)	2.1(14)	7.9(13)	4.9(14)	9.8(13)	1.5(14)	1.3(14)
CH_3CCH	7.9(14)	2.3(15)	8.6(14)	5.2(15)	8.4(14)	1.5(15)	1.3(15)

^a The column densities for the hot core molecules are corrected for beam-dilution using the values of η_{bf} listed in Table 2. The cold molecules are given as a beam-averaged column density for a beam of $14''$. ^b From H_2^{13}CO multiplied by 53. ^c From rotation diagram excluding optically thick lines. ^d From $\text{CH}_3^{13}\text{CN}$ multiplied by 53.

diagram for some sources like W 33A appears to consist of two temperature components as seen previously by van der Tak et al. (2000) (Fig. A.2). However, deducing the presence of two components from a rotation diagram is non-trivial as this behavior could also result from sub-thermal excitation and optical depth effects. Indeed, excluding lines of high optical depth leads to the disappearance of the two-component structure for most sources other than W 33A. The fits in Fig. A.2 are especially sensitive to lines with high excitation temperatures and are therefore thought to probe the hot component of the CH_3OH gas.

CH_2CO : CH_2CO has temperatures that are very low, generally only around 40 K. Slightly higher temperatures are found for NGC 6334 IRS1 and W 3(H_2O). These temperatures suggest that this molecule is not present in hot regions, but rather in the cold envelope. The IRAM and JCMT detections for NGC 7538 IRS1 and W 3(H_2O) are consistent. MacDonald et al. (1996) find very high values of 273 K for G34.3 based on three lines with $E_u > 100$ K. However the integrated intensity for the CH_2CO $17_{1,16}-16_{1,15}$ line at 346.6 GHz with $E_u > 150$ K in this survey, also detected by MacDonald et al. (1996), is consistent with the low rotation temperature found in our survey.

CH_3CHO : Due to unfavorable weather conditions, IRAM observations of CH_3CHO have only been performed for AFGL 2591, G75.78, NGC 7538 IRS1, and W 3(H_2O).

Almost all targeted lines were detected toward the sources NGC 7538 IRS1 and W 3(H_2O). Both sources have rotation temperatures <20 K, consistent with the results found by Ikeda et al. (2001, 2002). Although its relatively high dipole moment of 2.69 D implies that sub-thermal excitation can play a role, it is unlikely that the temperature is above 100 K, especially since molecules with similarly high dipole moments such as HNCO and CH_3CN have much higher rotation temperatures.

$\text{C}_2\text{H}_5\text{OH}$: $\text{C}_2\text{H}_5\text{OH}$ consistently has temperatures of more than 100 K. This is somewhat higher than was found by Ikeda et al. (2001), which could be due to the detection of more high excitation lines in this study. For the lower energy transitions, a steeper slope is observed, which indicates a lower temperature.

HCOOH : HCOOH has relatively low temperatures ranging from ~ 40 K for G24.78, and W 33A to ~ 70 K for NGC 6334 IRS1 and NGC 7538 IRS1. For NGC 7538 IRS1 and W 3(H_2O), it was possible to combine results from both IRAM and the JCMT. For NGC 7538 IRS1, low rotation temperatures are confirmed, whereas for W 3(H_2O) a scatter of more than one order of magnitude is present. As the W 3(OH) region is close to W 3(H_2O) it is possible that emission from this region is picked up in one of the settings. Alternatively, HCOOH could be present in both hot and cold gas with only the cold component present

Table 6. Derived rotation temperatures ^a in K from rotation diagram analysis (see Appendix 7).

Species	Sources						
	AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W 3(H ₂ O)	W 33A
H ₂ CO	89 ^b	83 ± 9	69 ± 6	193 ^b	87 ^b	181 ^b	88 ^b
CH ₃ OH	147 ± 11	211 ± 13	113 ± 7	178 ± 10	156 ± 10	139 ± 8	259 ± 16
C ₂ H ₅ OH	[139]	104 ± 7	[139]	166 ± 15	164 ± 17	129 ± 16	122 ± 16
HNCO	64 ± 8	96 ± 11	108 ± 13	105 ± 12	278 ± 88	147 ± 24	85 ± 8
NH ₂ CHO	[119]	170 ± 20	78 ± 22	166 ± 29	164 ± 28	71 ± 7	40 ± 13
CH ₃ CN	[218]	261 ± 20	189 ± 25	170 ± 13	[218]	196 ± 14	278 ± 44
C ₂ H ₅ CN	[96]	99 ± 5	[96]	92 ± 3	[96]	94 ± 16	[96]
HCOOCH ₃	[119]	121 ± 6	87 ± 13	144 ± 7	134 ± 8	109 ± 7	112 ± 15
CH ₃ OCH ₃	[130]	134 ± 12	[130]	241 ± 35	[130]	94 ± 5	43 ± 7
CH ₂ CO	[47]	39 ± 6	40 ± 6	67 ± 20	37 ± 4	54 ± 8	45 ± 7
CH ₃ CHO	[20]	[20]	[37.5]	[37.5]	18 ± 5	16 ± 3	[37.5]
HCOOH	[55]	42 ± 7	[55]	75 ± 15	73 ± 12	189 ± 108	38 ± 6
CH ₃ CCH	45 ± 12	59 ± 6	59 ± 14	63 ± 12	58 ± 10	82 ± 15	40 ± 6

^a The bracket notation indicates assumed rotation temperature, when no temperature could be derived from a rotation diagram. ^b H₂CO data from van der Tak et al. (2000).

for NGC 7538 IRS1. Due to the larger beam size of the JCMT, the 230 GHz observations are more sensitive to extended emission, whereas the IRAM 270 GHz observations probe a more compact region. Since most sources have low rotation temperatures, we conclude that it is mostly located in a cold environment. Low rotation temperatures have previously been inferred for HCOOH in other high mass star forming regions (Ikeda et al. 2001).

HNCO: The rotation temperature of HNCO varies between the different sources. This is not unexpected since the HNCO excitation is known to be dominated by radiation and not collisions (Churchwell et al. 1986). It is therefore not likely to provide much information about the actual kinetic temperature of the species. However it does need to be close to an infrared source to be excited, and thus must be present in hot gas. Finally, it is important to note that excluding the line with $E_u > 300$ K (see Fig. A.7) strongly reduces the inferred rotation temperature for NGC 6334 IRS1 from 180 K to 106 K.

NH₂CHO: NH₂CHO has a variety of rotation temperatures. The scatter in the derived rotation temperatures appear to be relatively large, with especially NGC 7538 IRS1 being an outlier (Fig. A.8). There is also some indication that there may be two temperature components – most clearly seen for the sources where many lines with $E_u > 200$ K are detected. Since NH₂CHO has lines very close to the detection limit, this component may also be present for the other sources, but simply below the detection limit.

CH₃CN: Rotation temperatures of CH₃CN range from 170–280 K. It is thus always present in hot gas (see Fig. A.9). The particularly high temperatures for G24.78 and W 33A are upper limits due to high optical depths. Pankonin et al. (2001) find lower rotation temperatures of 89 K for G24.78 which confirm this picture. For NGC 6334 IRS1 only three lines with $E_u \geq 500$ K have been detected, making the resulting rotation temperature uncertain. However, it seems clear that the rotation temperature for CH₃CN is high, and the species must be present in hot gas.

C₂H₅CN: The other cyanide, C₂H₅CN, only has enough detected lines for three of the sources to allow a determination of the rotation temperature (see Fig. A.10). For all three sources temperatures of ~90 K are found. Fewer lines have been detected for W 3(H₂O), but the fact that the results for this source

are rather similar gives additional credibility to the derived temperatures.

CH₃CCH: CH₃CCH generally has rotation temperatures below ~80 K – except for G24.78, NGC 6334 IRS1, and W 3(H₂O). The higher temperatures for these three sources are, however, questionable. Removing the point with $E_u > 300$ K gives significantly lower temperatures for G24.78 and NGC 6334 IRS1 (see Fig. A.11). The W 3(H₂O) generally has a higher uncertainty and scatter on the data, which may in fact cause the higher temperature found for this source.

HCOOCH₃: HCOOCH₃ is detected in almost all of the sources, but with a relatively large scatter on the data. The temperature generally seems to be between 100–150 K, with somewhat lower values when less lines have been detected. For G75.78 the results are inconclusive as the upper limits on high energy transitions are not significant. Somewhat lower temperatures were found by Ikeda et al. (2001), which could be due to a difference in the number of detected lines.

CH₃OCH₃: CH₃OCH₃ also appears to have a relatively large spread in rotation temperatures (see Fig. A.13). For W 33A, only lines with $E_u < 100$ K were detected, which result in low rotation temperatures. The upper limits on lines with $E_u > 150$ K are not significant, however, and do not exclude temperatures above 100 K. Rotation temperatures above 100 K are commonly seen for CH₃OCH₃ (Ikeda et al. 2001).

In summary, the temperatures for most molecules are rather consistent between all sources. The molecules can clearly be divided into “hot” ($T > 100$ K) and “cold” ($T < 100$ K) where H₂CO, CH₃OH, C₂H₅OH, HNCO, NH₂CHO, CH₃CN, C₂H₅CN, HCOOCH₃, and CH₃OCH₃ belong to the first and CH₂CO, CH₃CHO, HCOOH and CH₃CCH to the second category.

4.4. Column densities

Table 5 and Fig. 4 present the *source-averaged* column densities, i.e., corrected for beam dilution (Sect. 3.2), for the hot molecules per source. Absolute column densities are compared to determine the variability between different sources. Since we have assumed the beam dilution to be the same for all molecules, the trends for the *beam-averaged* column densities are similar. Molecules with rotation temperatures below 100 K are present in a different, cold region, presumably the envelope surrounding

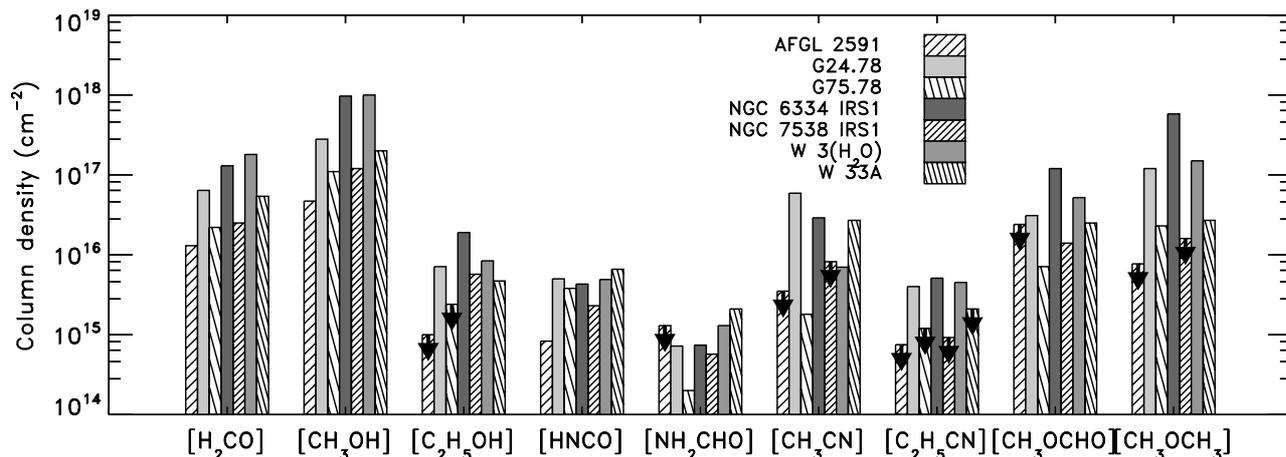


Fig. 4. Histogram of molecular column densities of hot molecules corrected for beam dilution for all sources for a source size defined by $T > 100$ K.

the hot region, and beam-averaged column densities are therefore given for those. Thus, the column densities for the hot and cold molecules in Figs. 4 and 5 cannot be compared directly.

Clearly, most molecules have column densities that vary by about an order of magnitude in the hot gas between the different sources. The column density variations do not scale with the luminosity and thus total mass in the hot region. CH_3OH , H_2CO , but also HCOOCH_3 , and CH_3OCH_3 – if detected – have column densities that are orders of magnitude higher than those of $\text{C}_2\text{H}_5\text{OH}$, HNCO , and NH_2CHO . As will be discussed in Sect. 5 abundances show a similar trend. The advantage of comparing column densities is that they come directly from the observations and no assumption on the H_2 column in the hot gas enters the analysis.

Most hot molecules have source-averaged column densities that are highest for the line-rich sources NGC 6334 IRS1, G24.78, and W 3(H_2O). For the oxygen-bearing molecules $\text{C}_2\text{H}_5\text{OH}$, CH_3OCH_3 , HCOOCH_3 , G24.78 has a higher column density than W 3(H_2O). W 3(H_2O) has a higher column density of CH_3OH and H_2CO , however. Thus H_2CO and CH_3OH have variable column densities compared to the other oxygen-bearing molecules. Nonetheless the column density trends between the oxygen-bearing species are similar.

The HNCO and NH_2CHO column densities peak for W 33A and G24.78, and have very similar column density trends with respect to each other. W 33A has the highest and G75.78 the lowest column densities. NGC 7538 IRS1 has a relatively high column density for these species compared with the oxygen-bearing species. The column densities of the nitrogen-bearing species do, therefore, behave different from the oxygen-bearing species.

CH_3CN has a relatively high column density for G24.78 and W 33A, but the uncertainties make it more difficult to compare its column density trends with other molecules (see Sect. 4.3). The general CH_3CN column density trends appear more similar to those of HNCO and NH_2CHO than to the O-bearing species. However, if we consider only the two sources for which both the main species and its isotopologue $\text{CH}_3^{13}\text{CN}$ were detected with $E_u < 500$ K, the relative column densities appear more similar to those of the oxygen-bearing species.

Figure 5 shows the *beam-averaged* column densities for the cold molecules, CH_2CO , HCOOH , and CH_3CCH . Clearly all three have very similar trends. NGC 6334 IRS1 has the highest column density and AFGL 2591 the lowest. For CH_3CHO , the number of sources with detections is limited and it was thus

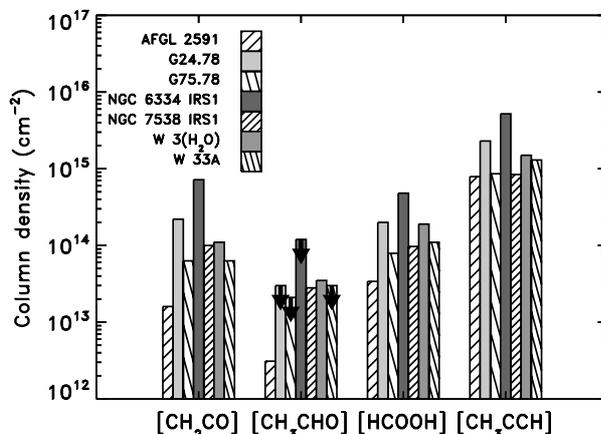


Fig. 5. Histogram of molecular beam-averaged column densities of the cold molecules for all sources.

not possible to compare its column density trends with the other species.

In summary, the molecular abundances vary about an order of magnitude between the sources. Furthermore, all oxygen-bearing species have similar abundance trends, but the abundances of HNCO and NH_2CHO behave differently.

5. Column density and abundance correlations

Molecular column densities and abundances can be compared between sources in different ways, two of which will be presented here. Abundances can be estimated relative to column densities of (i) H_2 calculated on the basis of CO , (ii) H_2 derived from dust continuum observations (see Sect. 3.2) or (iii) one of the observed molecules, typically CH_3OH . The disadvantage of method (i) is that CO gas is likely extended over much larger scales than the hot molecules due its lower desorption temperature. Observations of high excitation lines of CO isotopologues which probe only hot gas, will make this method possible in the future. In this paper, we use (ii) and (iii), and adopt (i) only for the cold molecules.

Abundances with respect to H_2 column densities from modeling of dust continuum observations (Sect. 3.2) and to CH_3OH column densities (Sect. 5.2) have the advantage that they allow comparisons to “only material at high temperatures” or “species probing the regions solely of interest”. Both approaches also

Table 7. Molecular abundances ($N(X)/N(H_2)$).

	AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W 3(H ₂ O)	W 33A	Average	σ^a
Hot molecules ^b									
H ₂ CO	1.7(-7)	1.6(-7)	1.8(-7)	5.4(-7)	1.2(-7)	1.0(-6)	2.1(-7)	3.4(-7)	3.2
CH ₃ OH	7.0(-7)	6.5(-7)	9.2(-7)	4.0(-6)	5.7(-7)	5.4(-6)	7.3(-7)	1.9(-6)	2.0
C ₂ H ₅ OH	<1.4(-8)	1.9(-8)	2.0(-8)	7.5(-8)	2.5(-8)	4.4(-8)	1.8(-8)	3.6(-8)	2.3
HNCO	7.0(-9)	1.3(-8)	3.2(-9)	1.8(-8)	1.1(-8)	2.7(-8)	2.5(-8)	1.6(-8)	0.91
NH ₂ CHO	<1.7(-8)	1.8(-9)	1.7(-9)	3.1(-9)	2.6(-9)	5.6(-9)	8.5(-9)	3.9(-9)	2.7
CH ₃ CN	<4.6(-8)	1.5(-7)	1.5(-8)	1.2(-7)	<3.9(-8)	3.9(-8)	1.0(-7)	8.5(-8)	5.6
C ₂ H ₅ CN	<9.9(-9)	1.0(-8)	<1.0(-8)	2.1(-8)	<4.4(-9)	2.5(-8)	<8.1(-9)	1.7(-8)	0.77
HCOOCH ₃	<3.2(-7)	7.3(-8)	5.9(-8)	4.6(-7)	6.7(-8)	2.9(-7)	9.6(-8)	1.7(-7)	1.6
CH ₃ OCH ₃	<1.0(-7)	3.0(-7)	1.9(-7)	2.4(-6)	<7.6(-8)	8.3(-7)	1.0(-7)	7.7(-7)	9.6
Cold molecules ^c									
CH ₂ CO	3.1(-11)	8.8(-10)	8.5(-11)	2.8(-10)	4.1(-10)	1.3(-10)	4.1(-10)	3.2(-10)	2.9
CH ₃ CHO	6.1(-12)	<6.3(-11)	<2.8(-11)	<4.7(-11)	1.2(-10)	4.1(-11)	<1.9(-10)	5.6(-11)	5.8
HCOOH	6.7(-11)	8.8(-10)	1.0(-10)	1.9(-10)	4.2(-10)	1.8(-10)	8.4(-10)	3.8(-10)	3.4
CH ₃ CCH	1.5(-9)	9.6(-9)	1.1(-9)	2.1(-9)	3.6(-9)	1.8(-9)	8.4(-9)	4.0(-9)	3.5

^a σ refers to the same order of magnitude as the average. ^b The abundances are derived with respect to the column density of H₂ within $R_{T=100\text{ K}}$ in Sect. 3.2 and Table 3. ^c The abundances are derived using the CO column densities from van der Tak et al. (2000), except G24.78 and G75.78 for which the data comes from Cesaroni et al. (2003) and Hatchell et al. (1998) respectively. ^c σ refers to the same order of magnitude as the average.

Table 8. Abundance correlations with respect to H₂ for the abundances within $R_{T=100\text{ K}}$ for hot core molecules.

	H ₂ CO	CH ₃ OH	C ₂ H ₅ OH	HNCO	NH ₂ CHO	CH ₃ CN	HCOOCH ₃	CH ₃ OCH ₃	HCOOH
H ₂ CO	1.00	0.98	0.76	0.46	0.53	-0.07	0.88	0.75	0.76
CH ₃ OH	0.98	1.00	0.86	0.38	0.39	-0.07	0.92	0.86	0.70
C ₂ H ₅ OH	0.76	0.86	1.00	0.02	0.22	0.12	0.92	0.95	0.30
HNCO	0.46	0.38	0.02	1.00	0.38	0.81	0.16	-0.27	0.98
NH ₂ CHO	0.53	0.39	0.22	0.38	1.00	0.17	0.42	-0.02	0.65
CH ₃ CN	-0.07	-0.07	0.12	-0.81	0.17	1.00	0.29	0.20	-0.51
HCOOCH ₃	0.88	0.92	0.92	0.16	0.42	0.29	1.00	0.90	0.49
CH ₃ OCH ₃	0.75	0.86	0.95	-0.27	-0.02	0.20	0.90	1.00	0.09
HCOOH ^a	0.76	0.70	0.30	0.98	0.65	-0.51	0.49	0.09	1.00

^a Assuming that the HCOOH emission comes from within $R_{T=100\text{ K}}$.

have disadvantages, however. Since H₂ is not directly observed, one has to rely on other tracers, in our case dust. Extrapolations of density and temperature profiles are made to scales smaller than actually probed by the observations. CH₃OH is directly observed, but this method implicitly assumes that the abundance of CH₃OH is constant from source to source. If this is not the case incorrect correlations of abundances with respect to CH₃OH can be found, as we will return to in Sect. 5.2.

5.1. Abundances in $T \geq 100\text{ K}$ gas

Table 7 lists the molecular abundances within the gas with $T \geq 100\text{ K}$ for each source in our sample calculated through Eq. (7). The abundances have the same trends between the different sources as the column densities. Most molecules have abundances that vary one order of magnitude from source to source, as was also seen for CH₃OH by van der Tak et al. (2000). The standard deviation on the average abundances is larger than our estimated uncertainty of a factor 2–3 (Sect. 3). Thus the observed variations are real. Furthermore, the overall abundances are relatively high compared to values in the literature. This is because we have determined the abundances with respect to the H₂ column density within $R_{T=100\text{ K}}$, whereas in e.g., Ikeda et al. (2001) they are determined based on CO transitions, which trace gas that is cooler and extended with respect to the hot core.

The abundances of the species detected in our sources with $T \geq 100\text{ K}$ were compared by estimating the Pearson correlation coefficients between $\log(X/H_2)$ vs. $\log(Y/H_2)$ (see Table 8).

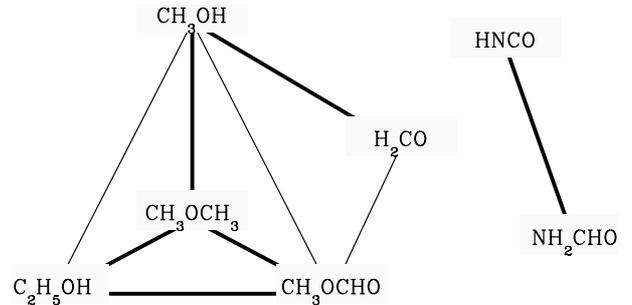


Fig. 6. Empirical correlations between the different molecules based on the Pearson correlation coefficients. All solid lines indicate correlations coefficients > 0.8 and the thick lines > 0.9 . When two species correlate, it does not mean that they are directly chemically related. Two species can also correlate if they are both formed from the same mother species.

C₂H₅CN is not compared with the other species due to the low number of sources with detections. Although the number of sources for the other molecules are limited and correlations therefore only suggestive, a number of interesting trends are seen. These result from groups of molecules with very similar abundance trends.

First, the abundances of the nitrogen-bearing species do not correlate with those of oxygen-bearing species. Strong correlations are, however, found between the oxygen-bearing species H₂CO, CH₃OH, C₂H₅OH, HCOOCH₃, and CH₃OCH₃. Figure 7 shows the abundances of all hot oxygen-bearing species with

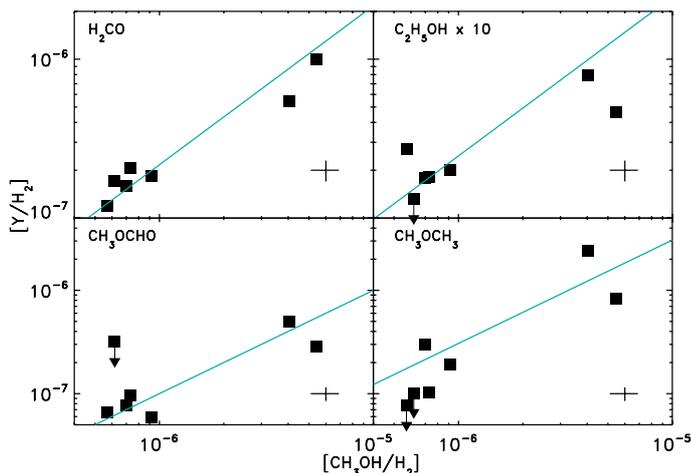


Fig. 7. Comparison of molecular abundances with respect to H_2 within $R_{T=100\text{ K}}$ for the oxygen-bearing molecules. The gray lines indicate the linear fit to the data. An error-bar representing the calibration uncertainties is depicted in the lower right corner.

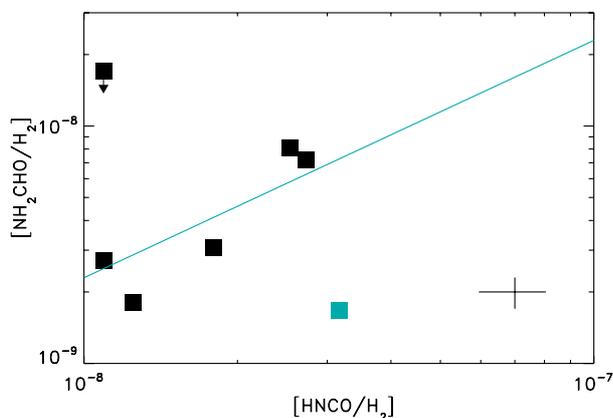


Fig. 8. Comparison of molecular abundances with respect to H_2 for HNCO with NH_2CHO within $R_{T=100\text{ K}}$. The source indicated in gray is G75.78 and the limit is AFGL 2591. An error-bar representing the calibration uncertainties is depicted in the lower right corner.

respect to H_2 compared to CH_3OH/H_2 . The nitrogen-bearing species HNCO and NH_2CHO also appear to be well correlated, which follows from the column density trends (Sect. 4 and shown in Fig. 8). The NH_2CHO abundance for G75.78 is uncertain, as only few lines were detected. The correlation coefficient is 0.38 with this source included and 0.92 without.

If the HCOOH emission is assumed to arise in hot instead of cold gas, its abundance correlates with very few molecular species. Its strongest correlation is with HNCO with a coefficient of 0.98. This result is surprising and due to the low rotation temperatures for HCOOH questionable.

The abundances of the hot molecules presented in Table 7 and Fig. 7 assume the same beam dilution factor corresponding to $R_{T=100\text{ K}}$. As Table 6 shows, the rotation diagram temperatures of several species are significantly higher than 100 K. The $T = 200\text{ K}$ radius is typically a factor of ~ 3 smaller than that for 100 K so that the beam dilution and abundances can be a factor ~ 9 larger. If all emission of hot molecules would arise from a smaller region, the correlations would remain the same. If some species are present in 100 K gas and others in 200 K gas, the correlations would disappear, however. The differences in rotation temperatures in Table 6 are assumed to arise largely from

other effects such as optical depth, sub-thermal excitation, or the detection of very high or low lying energy levels affecting the overall fit. Future interferometric observations of multiple high mass sources could determine whether the assumption of a single source size for all molecules is correct.

For the molecules that have abundances that do correlate, the abundance ratios were calculated with respect to either CH_3OH/H_2 or $HNCO/H_2$ and are shown in Table 9. Figure 6 shows the empirical relations for molecules with Pearson coefficients of 0.8 and higher. Per set of species the ratios vary no more than a factor of three, the previously found uncertainty in the abundances. H_2CO , $HCOOCH_3$, and CH_3OCH_3 have 5–10 times lower abundances than CH_3OH . NH_2CHO is a factor 3 times lower than HNCO. Possibly, coefficients between 0.8–0.9 are present between species that are not directly chemically related, but that are both strongly correlated with a third species.

5.2. Abundances with respect to CH_3OH

The Pearson correlation coefficients of $\log(X/CH_3OH)$ vs. $\log(Y/CH_3OH)$ differ significantly from the correlations in Sect. 5.1. First, strong correlations are not present between the same species. This is due to the variability of the CH_3OH abundance between the different sources, and to the dependence of the correlations on three (X , Y , and CH_3OH) rather than two (X and Y) species. The abundances of two species with respect to CH_3OH will correlate when their abundances with respect to H_2 correlate but also when the abundance of one of the species correlates with that of CH_3OH . This makes it difficult to use correlations of abundance ratios with respect to CH_3OH to chemically link molecules. Instead it is better to compare X/CH_3OH with the CH_3OH column density or abundance, i.e. $N(CH_3OH)$ or CH_3OH/H_2 as has been done by e.g., Requena-Torres et al. (2006).

5.3. Comparison to other well known sources

The abundance ratios with respect to CH_3OH are compared in Table 9 to other well-studied high mass star forming regions, specifically the Orion Compact Ridge (Sutton et al. 1995), G34.3 (MacDonald et al. 1996), Sgr B2(N) (Nummelin et al. 2000), and G327.3 (Gibb et al. 2000a). The ratios for the Orion Compact Ridge all agree within the error bars for H_2CO and $HCOOCH_3$. CH_3OCH_3 has a relatively low abundance with respect to CH_3OH . Only three lines have been detected for this species in the Compact Ridge and it has a rather low excitation temperature of 75 K implying that it is perhaps present in a colder region.

In contrast, the relative abundances for G34.3 deviate more from those in our work. The H_2CO/CH_3OH ratio by MacDonald et al. (1996) is probably a lower limit as only one line was detected for G34.3 and the column density of $H_2^{13}CO$ was calculated to be higher than the main H_2CO isotopologue based on their respective line intensities. Both C_2H_5OH and $HCOOCH_3$ have relatively high abundances with respect to CH_3OH . This could potentially be due to high optical depth in the lines of CH_3OH . MacDonald et al. (1996) assumed the emission to be optically thin, and thus underestimate the CH_3OH column density. The relative abundances of C_2H_5OH and $HCOOCH_3$ are similar between G34.3 and the sources studied in this paper.

The abundance ratios with respect to CH_3OH for Sgr B2(N) from Nummelin et al. (2000) appear to be significantly lower than for the sources in our sample. This is due to the CH_3OH

Table 9. Average abundance ratio for our sources and other well-known high mass YSOs with respect to CH₃OH for the oxygen-bearing species and with respect to HNCO for NH₂CHO. σ is the standard deviation between the sources in our sample.

Molecule	abundance ratio(σ)	Orion Compact Ridge ^a	G34.3 ^b	Sgr B2(N) ^c	G327.3 ^d	GC clouds ^e
with respect to CH ₃ OH						
H ₂ CO	0.22 ± 0.05	0.10	0.009	>0.0025	–	~0.01
C ₂ H ₅ OH	0.025 ± 0.013	0.004	0.19	0.005	0.05	~0.04
HCOOCH ₃	0.10 ± 0.03	0.08	0.86	0.005	0.96	~0.04
CH ₃ OCH ₃	0.31 ± 0.20	0.05	–	0.015	7	~0.04
with respect to HNCO						
NH ₂ CHO	0.36 ± 0.08	–	–	0.23	0.07	–

^a Sutton et al. (1995). ^b MacDonald et al. (1996). ^c Nummelin et al. (2000). ^d Gibb et al. (2000a). ^e Requena-Torres et al. (2006).

emission being decomposed into a core and a halo component for Sgr B2(N), whereas that of the other molecules is not. For this source it is more relevant to compare the relative ratios between the different molecules. The C₂H₅OH abundance is high compared to HCOOCH₃ and CH₃OCH₃. However, the HCOOCH₃/CH₃OCH₃ and the NH₂CHO/HNCO ratios of 0.33 are similar to those for our sources.

G327.3 in the study of Gibb et al. (2000a) has a C₂H₅OH/CH₃OH abundance ratio a factor of 2 higher compared to the average for our sample. Small deviations could be caused by their relatively low rotation temperature for C₂H₅OH of 66 K. HCOOCH₃ and CH₃OCH₃ have relatively high abundances with respect to CH₃OH, but HCOOCH₃/CH₃OCH₃ is within a factor two of the ratio derived for our sample. Again NH₂CHO/HNCO is the same within error to our sample.

Molecular abundance ratios can also be compared to those found in shocked gas in the galactic center by Requena-Torres et al. (2006). Their sources have relatively constant ratios of complex molecules with respect to CH₃OH of 0.04. The H₂CO/CH₃OH and CH₃OCH₃/CH₃OH ratios are significantly higher for our hot cores. Small differences may be due to the fact that Requena-Torres et al. (2006) observed rotational transitions with $E_u < 40$ –50 K, whereas lines with excitation energies that range from 20–900 K have been observed in this study.

In summary, the overall emerging picture is that of relatively constant abundance ratios between various molecules comparable to the sources for our sample. Distinct differences for individual species can often be ascribed to specific observation or analysis problems.

5.4. Comparison with ice abundances

It is interesting to relate the gas-phase abundances derived above to the ice abundances from mid-infrared observations. Of the sources in our sample, AFGL 2591, NGC 7538 IRS1 and W 33A were also observed at mid-infrared wavelengths with ISO (Gibb et al. 2004). Table 10 lists the abundances of the ice species from that paper assuming an H₂O abundance of 10⁻⁴ with respect to H₂. W 33A is particularly rich in organics in the solid state whereas the other two sources mainly have upper limits. The solid state abundances of CH₃OH are at least an order of magnitude higher than the comparable source-averaged gas phase abundances. The H₂CO abundance of $\sim 2 \times 10^{-6}$, although more uncertain, is also larger by an order of magnitude (Keane et al. 2001). For W 33A, however, the solid H₂CO/CH₃OH abundance ratio is consistent with the gas phase abundance ratio.

Where detected toward high mass YSOs, solid HCOOH and CH₃CHO have abundances of 10⁻⁶–10⁻⁵ and $\sim 10^{-5}$ respectively (Schutte et al. 1999; Keane et al. 2001; Gibb et al. 2000b, 2004). Similar abundances for solid HCOOH are found toward

background stars by Knez et al. (2005). These absolute solid state abundances are four to five orders of magnitude higher than the gas phase abundances for the same sources.

The XCN band is detected in many objects and usually assigned to OCN⁻ (Grim & Greenberg 1987; Demyk et al. 1998; Pendleton et al. 1999; Gibb et al. 2000b; van Broekhuizen et al. 2005). OCN⁻ is thought to convert to and evaporate off grains as HNCO (Sect. 6.1). Abundances of solid OCN⁻ vary and range from 1×10^{-7} – 4×10^{-6} (van Broekhuizen et al. 2005; Thi et al. 2006). Gas phase abundances of HNCO are factors of 100–1000 times lower.

Many of the species are expected to have abundances in the ice below the detection limit. Boudin et al. (1998) determined upper limits for the column densities of a number of more complex organic molecules, including C₂H₆ and C₂H₅OH toward NGC 7538 IRS9. The upper limit on the C₂H₅OH abundance is 1.2×10^{-6} with C₂H₅OH/CH₃OH < 0.28. This is consistent with the C₂H₅OH/CH₃OH abundance ratio of 0.009 in the gas phase for our sample of sources.

In summary, the absolute ice abundances of observed “first generation” species, CH₃OH and OCN⁻, are 10–10² times higher compared to the gas phase values, and as much as a factor of 10⁴–10⁵ for HCOOH and CH₃CHO.

6. Discussion and comparison to astrochemical models

6.1. General model considerations

Pure gas phase chemistry models such as presented in e.g., Lee et al. (1996), are unable to explain the high abundances of many complex organics, especially of highly saturated molecules in hot gas phase environments. CH₃OH, for example, is estimated to have abundances of $\sim 10^{-12}$ through pure gas phase chemistry, whereas abundances of 10⁻⁶ are found. For CH₃CN abundances of $\sim 10^{-10}$ are modeled (Lee et al. 1996) whereas abundances of $\sim 10^{-7}$ are detected. Furthermore, they are not able to explain the presence of complex organics in ices by simple freeze-out of gas phase species (see Herbst 2005, for a review). For highly saturated molecules, solid state chemistry provides a likely alternative. Here one can distinguish basic grain surface chemistry and chemistry induced by energetic processing (UV irradiation, energetic particle bombardment and/or thermal heating) which can proceed also inside the ice.

Surface chemistry models have been developed by several groups (e.g., Hasegawa & Herbst 1993; Charnley 2001; Keane 2001) (see Fig. 1). The results are generally very sensitive to the atomic C, O, and N abundances in the gas. Moreover, the dust temperature (T_{dust}) plays a crucial role in the mobility of the atoms and radicals as well as the availability of CO as a reaction

Table 10. Solid state abundances with respect to H₂, assuming H₂O ice to have an abundance of 10⁻⁴.

Species	W 33A ^a 10 ⁻⁵	NGC 7538 IRS1 ^b 10 ⁻⁵	AFGL 2591 ^b 10 ⁻⁵	HH 46 ^c (low mass) 10 ⁻⁵	CK 2 ^c (field star) 10 ⁻⁵	gas phase ^d
H ₂ CO	0.65	–	–	–	–	3.2(–7)
CH ₃ OH	1.4–1.7	<0.4	1.4	1.2	<0.11	2.0(–6)
HCOOH	0.37–0.71	2.2	<1.7	<1.5	0.11	3.8(–10)
CH ₃ CHO	0.98	1.2	<0.66	–	–	5.6(–11)
OCN [–]	0.63	<0.05	–	≤0.12	–	1.6(–8) ^e

^a From Gibb et al. (2000b); Dartois et al. (1999); Schutte et al. (1999); Whittet et al. (2001); Demyk et al. (1998). ^b From Gibb et al. (2004). ^c From Knez et al. (2005) and references therein. ^d Average abundances from this paper. ^e Assuming that OCN[–] evaporates as HNCO.

partner on the grain, which evaporates at $T_{\text{dust}} > 20$ K (Tielens & Hagen 1982; Caselli et al. 1993). However, if CO is located in an ice environment that is dominated by H₂O it will be available for reactions in the ice up to much higher temperatures (Sandford & Allamandola 1993; Collings et al. 2003).

If grain surface chemistry is the explanation for our observed correlations, the constant abundance ratios for different sources must imply very similar conditions during ice formation, at least of the ice layer that contains the “hot” molecules. This is also confirmed by the similarity between the composition of ices observed toward background stars compared to those toward low and high mass star forming regions (Knez et al. 2005). Perhaps the simplest explanation is that the bulk of the ice forms at similarly low temperature conditions in which the bulk of the oxygen and nitrogen are in atomic form and C/CO is at a fairly constant ratio, around 0.01 as computed for dense clouds (Gredel et al. 1989). When these “first generation” species evaporate together and have similar destruction rates, a constant abundance ratio would result even in regions where the temperatures of the gas and UV-flux are very different.

In contrast, “second generation” species are expected to peak at different times dependent on their formation and destruction processes. While this argument has been used to explain the anti-coincidences between the nitrogen and oxygen-bearing species (Caselli et al. 1993; Rodgers & Charnley 2003), it is inconsistent with our observed correlations of supposedly “second generation” species HCOOCH₃ and CH₃OCH₃ with other “first generation” oxygen-bearing molecules. Indeed, in models by Garrod & Herbst (2006) where HCOOCH₃ is formed in the solid state, many of the oxygen-bearing species have similar gas phase abundance variations throughout the chemical evolution and have constant abundance ratios. Solid state formation is therefore more likely for the supposedly “second generation” oxygen-bearing species. However, the abundances of oxygen-bearing and nitrogen-bearing species do not correlate. This can be due to a different time-dependence of formation and destruction or a variable N/C abundance ratio.

Laboratory experiments show that beyond a certain minimum dose of irradiation the reaction products often do reach a constant plateau (Gerakines et al. 1996; Moore & Hudson 2000; van Broekhuizen et al. 2004; Hagen et al. 1979). Oxygen-bearing species, such as H₂CO, and nitrogen-bearing species, e.g., OCN[–] and NH₂CHO, can be formed this way (Grim et al. 1989). However, there is not enough information in the literature on the formation of the more complex molecules observed here to make a quantitative test of this scheme. It is plausible, though, that an equilibrium between formation and destruction is reached in interstellar ices and could be responsible for the constant abundance ratios between the oxygen-bearing species or HNCO/NH₂CHO.

Recent work by Garrod & Herbst (2006) combines both grain surface chemistry and UV-induced chemistry with thermal evolution of the ices during the protostellar phase. In particular, radicals created in the ices by ultraviolet radiation become mobile at the higher dust temperatures during the warm-up phase and lead to the formation of many complex organic species. This suggests that the ice composition changes with temperature. Potentially only the ices on grains that are close to the evaporation temperature have similar relative abundances to those observed in the gas phase. Since the detected ice features are observed in a column, they will contain both cold and warmer ices, which complicates the comparison of the ice abundances with the gas phase, even within one source.

Species such as HCOOH and CH₃CHO are detected with much lower column densities in the gas phase than in the ice (see Sect. 5.4) and have low rotation temperatures. Possibly, these species are formed in the solid state at low temperatures and destroyed due to reactions at high temperatures. Since their desorption temperatures are typically ~100 K (Viti et al. 2004), thermal desorption cannot account for their low temperatures. However, non-thermal mechanisms induced by, for example, cosmic rays (Léger et al. 1985), can explain the low gas phase abundances if they cause a small fraction (10⁻⁴) of the ice to evaporate (see Sect. 6.2). Alternatively, cold gas phase ion-molecule reactions are able to reproduce the abundances of e.g., CH₃CHO and CH₂CO (Lee et al. 1996, see also Sect. 6.2).

Ices are expected to be layered because different molecules such as H₂O, CO, and N₂ form or condense at different densities. Since nitrogen is transformed into N₂ deepest in the cloud it is expected to freeze-out last. To form species such as HNCO or NH₂CHO a more reactive form of nitrogen, i.e. N-atoms are needed. N-atoms could potentially be present in a non-hydrogen bonding layer dominated by CO or a layer dominated by H₂O. If they were present in the non-hydrogen bonding layer a small fraction of non-thermal desorption would result in cold gas phase N-bearing species. The lack of these cold N-bearing molecules and high rotation temperatures of HNCO and NH₂CHO suggests that they desorb at higher temperatures and must thus be present in a different, more tightly bound ice layer.

In summary, pure gas-phase reactions cannot reproduce the abundances of most species depicted in Fig. 1, and second generation gas phase chemistry is inconsistent with the observed correlations for the oxygen-bearing species. Constant abundance ratios after ice evaporation can be explained either by similar initial conditions for ice formation or an equilibrium between formation and destruction due to energetic processes. The low abundances of cold gas phase species could be due to gas phase ion-molecule reactions or non-thermal desorption of a small fraction of the ices e.g., by cosmic rays.

6.2. Individual molecules

This subsection elaborates on the general conclusions stated in Sect. 6.1 for individual molecules, grouped according to the different branches of Fig. 1.

H₂CO and CH₃OH: CH₃OH and H₂CO are molecules commonly expected to originate from grain-surface chemistry. They are formed through successive hydrogenation of CO on grains, a scheme that has been tested experimentally by Hiraoka et al. (2002), Watanabe et al. (2004), and Fuchs et al. (in prep.). Both H₂CO and CH₃OH have been detected in these experiments with different abundance ratios dependent on the exact conditions. H₂CO can also be formed by UV-irradiation of CH₃OH containing ices (Allamandola et al. 1988; Bernstein et al. 1995). However, the high ice abundances of CH₃OH (e.g., Gibb et al. 2004) and the constant abundance ratio between H₂CO/CH₃OH in this paper are inconsistent with this CH₃OH destruction mechanism. Our data imply that hydrogenation reactions are more feasible formation mechanisms.

Furthermore as discussed in Sect. 5.4, the solid state abundance of CH₃OH appears to be at least an order of magnitude higher than the comparable source-averaged gas phase abundances. This could be due to a number of effects: i) CH₃OH is partly destroyed upon evaporation, ii) intrinsic CH₃OH ice variations from source to source (Brooke et al. 1999; Dartois et al. 1999; Pontoppidan et al. 2004), iii) an underestimate of the beam-filling factor of the CH₃OH emitting gas. Possibly, all three factors contribute to the relatively low gas phase abundance.

CH₂CO, CH₃CHO, and C₂H₅OH: in Fig. 1 CH₂CO, CH₃CHO, and C₂H₅OH are thought to be interrelated through successive hydrogenation as well as to H₂CO and CH₃OH by successive H- and C-atom addition. All three species are expected to evaporate from the ice together with H₂O (Viti et al. 2004) and have similar rotation temperatures. This is not the case, however. Indeed, interferometric observations of CH₃CHO in high mass star forming regions show that CH₃CHO emission is extended with respect to hot cores species such as HCOOCH₃ and CH₃OCH₃ (Liu 2005). This suggest that CH₂CO, CH₃CHO, and C₂H₅OH gas are not present in the same region. Possibly, H-atom addition to CH₂CO and CH₃CHO on grains is very efficient and completely converts them to C₂H₅OH at high temperatures, so that only C₂H₅OH evaporates from the ice and is detected in the warm inner regions. Non-thermal desorption of a small fraction due to e.g., cosmic ray spot heating of CH₂CO and CH₃CHO ice in cold regions could explain the low gas phase abundances and rotation temperatures, as well as the high ice abundances of CH₃CHO. Alternatively, C₂H₅OH could be formed through another surface reaction related to CH₃OH and H₂CO formation, explaining its constant abundance ratio.

For CH₂CO and CH₃CHO there are also potential gas phase formation mechanisms. Millar & Nejad (1985) argue that CH₂CO could result from radiative associative reactions between CH₃⁺ and CO. Assuming that CH₃CO⁺ dissociative recombination results in CH₂CO for 50% of the cases, this would allow for CH₂CO abundances of $\sim 10^{-10}$, comparable to our observed abundances and also to those by Turner et al. (1999). Gas phase models by Lee et al. (1996) predict fractional abundances of 5×10^{-12} for CH₃CHO and its isomers. This is lower than the abundances of Ikeda et al. (2001) and Charnley (2004) but close to the abundances for some of our sources. Herbst & Leung (1989) proposed that H₃O⁺ and C₂H₂ react to form CH₃CHO after dissociative recombination, which results in higher abundances of up to $\sim 10^{-10}$ in dark clouds. Many of these

conclusions depend on high branching ratios toward these products in dissociative recombination. If three body break-up prevails as found for some species, pure gas phase production becomes untenable (Geppert et al. 2005).

HCOOH: High abundances of HCOOH in the ice, low gas phase rotation temperatures, and orders of magnitude lower abundances in the gas phase compared to the solid state can be due to two reasons: i) HCOOH is quickly destroyed in the gas phase upon evaporation, and ii) HCOOH is abundant in low temperature ice, but destroyed at higher temperatures due to reactions in the ice. Gas phase destruction is a possibility, but it is unclear why the destruction rate of HCOOH would be orders of magnitude larger compared to other species. Even if this would be the case, most high mass star-forming regions are expected to have an evaporation front that moves outward through the envelope as the luminosity increases. HCOOH should thus continuously freshly evaporate from icy grains. The second possibility of HCOOH destruction in the ice at high temperatures is much more likely and can be tested in the laboratory. The lower gas phase abundance can then be explained by a small fraction of the ice ($\sim 10^{-4}$) that evaporates due to non-thermal desorption mechanisms. The low rotation temperatures of ~ 40 – 70 K for most sources in our sample are also consistent with this picture. Finally, it can explain the high HCOOH abundances in shocked regions by e.g. Requena-Torres et al. (2006), where the ice is warmed up and evaporated over much shorter timescales.

The precise formation of HCOOH through surface chemistry still remains a puzzle, however. In Fig. 1 the formation mechanisms in the solid state is either through CO or CO₂. Laboratory experiments indicate that it only evaporates at very high temperatures (~ 100 K) similar to H₂O (Collings et al. 2004). This is even the case for HCOOH diluted in CO or CO₂ ices (Bisschop et al., in prep.). Keane (2001) predicts the formation of HCOOH through successive atom addition to CO and finds that the HCOOH ice concentration is proportional to the O/O₂ ratio. Alternatively, Allen & Robinson (1977) proposed the reaction HCO + OH in the solid state leading to HCOOH. New models by Garrod & Herbst (2006) have HCOOH gas phase abundances that are much higher than found for our sources with most of the HCOOH actually formed in the gas. If this occurs at intermediate temperatures, a fraction of the HCOOH may freeze out on grains. Further laboratory experiments are needed to elucidate the exact formation mechanism for HCOOH.

HNCO and NH₂CHO: HNCO formation has been explained by surface chemistry formation and gas phase reactions (Turner et al. 1999; Keane 2001). Both Turner et al. (1999) and Keane (2001) argue that NH₂CHO is formed on grain-surfaces due to its high level of hydrogen saturation. The strong correlation between the two species in Sect. 5.1 suggests they are related. Alternatively, both OCN⁻ and NH₂CHO can form in ices due to energetic processes such as UV-irradiation (van Broekhuizen et al. 2004; Muñoz Caro & Schutte 2003; Allamandola et al. 1999). Van Broekhuizen et al. (2004) conclude that thermal formation of OCN⁻ can, however, also not be ruled out. Concluding, solid state processes appear to be the most likely formation mechanism for both species, whether they involve energetic processing or not.

HCOOCH₃ and CH₃OCH₃: HCOOCH₃ and CH₃OCH₃ were previously thought to form from protonated methanol in the gas phase (Millar et al. 1991). Horn et al. (2004) recently found the barrier for the gas phase formation of HCOOCH₃ to be higher than previously anticipated. Moreover, the gas phase routes assume that H₂COOCH₃⁺ and (CH₃)₂OH⁺ recombine

to HCOOCH_3 and CH_3OCH_3 respectively. Measurements by Geppert et al. (2004, 2005) have shown that dissociative recombination of CH_3OH_2^+ mostly results in a three-body break-up instead of its deprotonated counterpart. If this would also be the case for species such as HCOOCH_3 and CH_3OCH_3 , they are more likely to be at least partially formed on grain surfaces. The constant abundance ratios between both species and CH_3OH also confirm this scenario. Indeed Garrod & Herbst (2006) explain surface formation of HCOOCH_3 and CH_3OCH_3 through reactions of CH_3O and HCO radicals, first proposed by Allen & Robinson (1977).

CH_3CN , $\text{C}_2\text{H}_5\text{CN}$, and CH_3CCH : CH_3CN can be formed from the fast association reaction $\text{CH}_3^+ + \text{HCN}$ and the radiative association reaction of $\text{CH}_3 + \text{CN}$ (Charnley et al. 1992). The abundances in the gas phase for Orion are, however, $\sim 10^{-10}$ (Charnley et al. 1992) compared to 10^{-8} in our sources. The relatively high abundances plus the high degree of hydrogenation for $\text{C}_2\text{H}_5\text{CN}$ and CH_3CN imply that they are possibly formed on grain-surfaces. CH_3CCH usually has very low rotation temperatures, and its formation is explained in dark and translucent clouds through gas phase ion-molecule or neutral-neutral reactions.

In summary, most species, such as H_2CO , CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, HCOOCH_3 , and CH_3OCH_3 probably result from “first generation” chemistry in ices, likely dominated by surface chemistry. This is possibly also true for CH_3CN and $\text{C}_2\text{H}_5\text{CN}$. CH_2CO , CH_3CHO , HCOOH and CH_3CCH have all been detected in the gas with low rotation temperatures and, for some species, abundances that are orders of magnitude lower than those in ices. Non-thermal desorption of a small fraction of the ice ($\sim 10^{-4}$) is sufficient to explain the abundances in the cold gas phase. For CH_2CO , CH_3CHO , and HCOOH formation plus destruction in the ice at higher temperature are plausible, but gas phase chemistry cannot be excluded. Models of gas phase reactions are able to accurately reproduce the observed CH_3CCH abundances.

7. Summary and conclusions

We have performed a survey of selected frequency settings for 7 high mass YSOs. The survey was targeted at detecting species thought to be produced by the grain surface chemistry depicted in Fig. 1. Other species such as CH_3CN , $\text{C}_2\text{H}_5\text{CN}$, CH_3CCH , HCOOCH_3 and CH_3OCH_3 have also been detected. To compare abundances between sources, the source luminosity and $850 \mu\text{m}$ dust emission was used to constrain the radius within which $T \geq 100 \text{ K}$ and the column density out to this radius, $N_{T \geq 100 \text{ K}}$. The main conclusion derived in this work are:

- Most detected molecules have similar rotation temperatures for all sources. The following molecules can be classified as cold ($T < 100 \text{ K}$): CH_2CO , CH_3CHO , HCOOH , and CH_3CCH ; and the following as hot ($T \geq 100 \text{ K}$): H_2CO , CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, HNCO , NH_2CHO , CH_3CN , $\text{C}_2\text{H}_5\text{CN}$, HCOOCH_3 , and CH_3OCH_3 .
- Groups of molecules have their highest column densities for the same sources. This is the case for the oxygen-bearing species H_2CO , CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, HCOOCH_3 and CH_3OCH_3 . HNCO and NH_2CHO have their highest column in other sources than the oxygen-bearing species, but both are highest for the same sources. Due to optical depth effects and a limited number of detections it is difficult to establish the trends for CH_3CN and $\text{C}_2\text{H}_5\text{CN}$.
- Absolute abundances in the hot regions vary by an order of magnitude. Relative abundances of the oxygen-bearing species and the two nitrogen-bearing species are, however, very constant. The absolute abundances are much higher than can be produced by gas phase chemistry. The constant abundance ratios plus the similar rotation temperatures imply that the oxygen-bearing species, H_2CO , CH_3OH , $\text{C}_2\text{H}_5\text{OH}$, HCOOCH_3 and CH_3OCH_3 , as well as the two nitrogen-bearing species, HNCO and NH_2CHO , share a common solid state formation scheme. This can be the result of very similar conditions during ice formation or a balance between formation and destruction in the ice. The oxygen-bearing species and the two nitrogen-bearing species are not related, however. The abundances of two cyanides, CH_3CN and $\text{C}_2\text{H}_5\text{CN}$, are likely due to “first generation” ice chemistry as well.
- CH_2CO , CH_3CHO , CH_3CCH and HCOOH have very low rotation temperatures. The formation of CH_2CO , CH_3CHO , and HCOOH can be explained by solid state or gas phase reactions. For HCOOH and CH_3CHO gas phase abundances are a factor of 10^4 lower than those observed in the solid state. Solid state formation for these species is thus more likely. A low fraction of non-thermal desorption could be responsible for the low gas phase abundances. Models of ion-molecule reactions in the cold gas phase can, however, also reproduce the abundances of CH_2CO and CH_3CHO . CH_3CCH on the other hand is probably created by cold gas phase chemistry, as its abundance is similar to that found in dark clouds. Interestingly, no cold N-bearing molecules are found.

The abundance correlations and branching ratios determined in this paper are key to understanding chemical relationships between these complex organic molecules and addressing their origins. Future spatially resolved observations with interferometers such as ALMA will make it possible to confirm molecular relations and address the chemical variations within each source. Additional laboratory experiments can explore the pathways to the formation of the most abundant species given these observational constraints of constant abundance ratios. Taken together such studies will be important to fully understand the chemistry of complex organic molecules in star forming regions.

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References

- Allamandola, L. J., Sandford, S. A., & Valero, G. J. 1988, *Icarus*, 76, 225
 Allamandola, L. J., Bernstein, M. P., Sandford, S. A., & Walker, R. L. 1999, *Space Sci. Rev.*, 90, 219
 Allen, M., & Robinson, G. W. 1977, *ApJ*, 212, 396
 Beltrán, M. T., Cesaroni, R., Neri, R., et al. 2005, *A&A*, 435, 901
 Bernstein, M. P., Sandford, S. A., Allamandola, L. J., Chang, S., & Scharberg, M. A. 1995, *ApJ*, 454, 327
 Blake, G. A., Sutton, E. C., Masson, C. R., & Phillips, T. G. 1987, *ApJ*, 315, 621
 Blake, G. A., Mundy, L. G., Carlstrom, J. E., et al. 1996, *ApJ*, 472, L49
 Boogert, A. C. A., & Ehrenfreund, P. 2004, in *Astrophysics of Dust*, ed. A. N. Witt, G. C. Clayton, & B. T. Draine, ASP Conf. Ser. 309, 547
 Boonman, A. M. S., Stark, R., van der Tak, F. F. S., et al. 2001, *ApJ*, 553, L63
 Boudin, N., Schutte, W. A., & Greenberg, J. M. 1998, *A&A*, 331, 749

- Brooke, T. Y., Sellgren, K., & Geballe, T. R. 1999, *ApJ*, 517, 883
- Caselli, P., Hasegawa, T. I., & Herbst, E. 1993, *ApJ*, 408, 548
- Cesaroni, R., Codella, C., Furuya, R. S., & Testi, L. 2003, *A&A*, 401, 227
- Charnley, S. B. 1995, *Ap&SS*, 224, 251
- Charnley, S. B. 2001, *ApJ*, 562, L99
- Charnley, S. B. 2004, *Adv. Space Res.*, 33, 23
- Charnley, S. B., Tielens, A. G. G. M., & Millar, T. J. 1992, *ApJ*, 399, L71
- Churchwell, E., Wood, D., Myers, P. C., & Myers, R. V. 1986, *ApJ*, 305, 405
- Churchwell, E., Walmsley, C. M., & Cesaroni, R. 1990, *A&AS*, 83, 119
- Collings, M. P., Dever, J. W., Fraser, H. J., & McCoustra, M. R. S. 2003, *Ap&SS*, 285, 633
- Collings, M. P., Anderson, M. A., Chen, R., et al. 2004, *MNRAS*, 354, 1133
- Dartois, E., Schutte, W., Geballe, T. R., et al. 1999, *A&A*, 342, L32
- Demyk, K., Dartois, E., D'Hendecourt, L., et al. 1998, *A&A*, 339, 553
- Forster, J. R., & Caswell, J. L. 1989, *A&A*, 213, 339
- Garrod, R. T., & Herbst, E. 2006, *A&A*, 457, 927
- Geppert, W. D., Thomas, R., Semaniak, J., et al. 2004, *ApJ*, 609, 459
- Geppert, W. D., Hellberg, F., Österdahl, J., et al. 2005, in ed. D. C. Lis, G. A. Blake, & E. Herbst, *IAU Symp.* 231, 117
- Gerakines, P. A., Schutte, W. A., & Ehrenfreund, P. 1996, *A&A*, 312, 289
- Gibb, E., Nummelin, A., Irvine, W. M., Whittet, D. C. B., & Bergman, P. 2000a, *ApJ*, 545, 309
- Gibb, E. L., Whittet, D. C. B., Schutte, W. A., et al. 2000b, *ApJ*, 536, 347
- Gibb, E. L., Whittet, D. C. B., Boogert, A. C. A., & Tielens, A. G. G. M. 2004, *ApJS*, 151, 35
- Goldsmith, P. F., & Langer, W. D. 1999, *ApJ*, 517, 209
- Gredel, R., Lepp, S., Dalgarno, A., & Herbst, E. 1989, *ApJ*, 347, 289
- Grim, R. J. A., & Greenberg, J. M. 1987, *ApJ*, 321, L91
- Grim, R. J. A., Greenberg, J. M., de Groot, M. S., et al. 1989, *A&AS*, 78, 161
- Hagen, W., Allamandola, L. J., & Greenberg, J. M. 1979, *Ap&SS*, 65, 215
- Hasegawa, T. I., & Herbst, E. 1993, *MNRAS*, 263, 589
- Hatchell, J., & van der Tak, F. F. S. 2003, *A&A*, 409, 589
- Hatchell, J., Thompson, M. A., Millar, T. J., & MacDonald, G. H. 1998, *A&AS*, 133, 29
- Helmich, F. P., & van Dishoeck, E. F. 1997, *A&AS*, 124, 205
- Herbst, E. 2005, in *The Dusty and Molecular Universe: A Prelude to Herschel and ALMA*, ed. A. Wilson, 205
- Herbst, E., & Leung, C. M. 1989, *ApJS*, 69, 271
- Hiraoka, K., Sato, T., Sato, S., et al. 2002, *ApJ*, 577, 265
- Horn, A., Møllendal, H., Sekiguchi, O., et al. 2004, *ApJ*, 611, 605
- Ikeda, M., Ohishi, M., Nummelin, A., et al. 2001, *ApJ*, 560, 792
- Ikeda, M., Ohishi, M., Nummelin, A., et al. 2002, *ApJ*, 571, 560
- Ivezić, Ž., Nenkova, M., & Elitzur, M. 1999, *User Manual for DUSTY*, University of Kentucky Internal Report
- Jørgensen, J. K., Schöier, F. L., & van Dishoeck, E. F. 2002, *A&A*, 389, 908
- Keane, J. V. 2001, Ph.D. Thesis, Rijks Universiteit Groningen
- Keane, J. V., Tielens, A. G. G. M., Boogert, A. C. A., Schutte, W. A., & Whittet, D. C. B. 2001, *A&A*, 376, 254
- Knez, C., Boogert, A. C. A., Pontoppidan, K. M., et al. 2005, *ApJ*, 635, L145
- Lee, H.-H., Herbst, E., Pineau des Forets, G., Roueff, E., & Le Bourlot, J. 1996, *A&A*, 311, 690
- Léger, A., Jura, M., & Omont, A. 1985, *A&A*, 144, 147
- Liu, S.-Y. 2005, in *IAU Symp.* 231, ed. D. C. Lis, G. A. Blake, & E. Herbst, 217
- MacDonald, G. H., Gibb, A. G., Habing, R. J., & Millar, T. J. 1996, *A&AS*, 119, 333
- Millar, T. J. 1996, in *Molecules in Astrophysics: Probes, & Processes*, ed. E. F. van Dishoeck, *IAU Symp.* 178, 75
- Millar, T. J., & Nejad, L. A. M. 1985, *MNRAS*, 217, 507
- Millar, T. J., Herbst, E., & Charnley, S. B. 1991, *ApJ*, 369, 147
- Moore, M. H., & Hudson, R. L. 2000, *Icarus*, 145, 282
- Muñoz Caro, G. M., & Schutte, W. A. 2003, *A&A*, 412, 121
- Mueller, K. E., Shirley, Y. L., Evans, N. J., & Jacobson, H. R. 2002, *ApJS*, 143, 469
- Nummelin, A., Bergman, P., Hjalmanson, Å., et al. 2000, *ApJS*, 128, 213
- Pankonin, V., Churchwell, E., Watson, C., & Bieging, J. H. 2001, *ApJ*, 558, 194
- Pendleton, Y. J., Tielens, A. G. G. M., Tokunaga, A. T., & Bernstein, M. P. 1999, *ApJ*, 513, 294
- Pontoppidan, K. M., Fraser, H. J., Dartois, E., et al. 2003, *A&A*, 408, 981
- Pontoppidan, K. M., van Dishoeck, E. F., & Dartois, E. 2004, *A&A*, 426, 925
- Requena-Torres, M. A., Martín-Pintado, J., Rodríguez-Franco, A., et al. 2006, *A&A*, 455, 971
- Rodgers, S. D., & Charnley, S. B. 2003, *ApJ*, 585, 355
- Sandford, S. A., & Allamandola, L. J. 1993, *ApJ*, 417, 815
- Schilke, P., Benford, D. J., Hunter, T. R., Lis, D. C., & Phillips, T. G. 2001, *ApJS*, 132, 281
- Schilke, P., Groesbeck, T. D., Blake, G. A., & Phillips, T. G. 1997, *ApJS*, 108, 301
- Schutte, W. A., Boogert, A. C. A., Tielens, A. G. G. M., et al. 1999, *A&A*, 343, 966
- Sutton, E. C., Blake, G. A., Masson, C. R., & Phillips, T. G. 1985, *ApJS*, 58, 341
- Sutton, E. C., Peng, R., Danchi, W. C., et al. 1995, *ApJS*, 97, 455
- Thi, W.-F., van Dishoeck, E. F., Dartois, E., et al. 2006, *A&A*, 449, 251
- Tielens, A. G. G. M., & Charnley, S. B. 1997, *Origins Life Evol. B.*, 27, 23
- Tielens, A. G. G. M., & Hagen, W. 1982, *A&A*, 114, 245
- Turner, B. E., Terzieva, R., & Herbst, E. 1999, *ApJ*, 518, 699
- van Broekhuizen, F. A., Keane, J. V., & Schutte, W. A. 2004, *A&A*, 415, 425
- van Broekhuizen, F. A., Pontoppidan, K. M., Fraser, H. J., & van Dishoeck, E. F. 2005, *A&A*, 441, 249
- van der Tak, F. F. S., van Dishoeck, E. F., Evans, N. J., & Blake, G. A. 2000, *ApJ*, 537, 283
- Viti, S., Collings, M. P., Dever, J. W., McCoustra, M. R. S., & Williams, D. A. 2004, *MNRAS*, 354, 1141
- Watanabe, N., Nagaoka, A., Shiraki, T., & Kouchi, A. 2004, *ApJ*, 616, 638
- Whittet, D. C. B., Pendleton, Y. J., Gibb, E. L., et al. 2001, *ApJ*, 550, 793
- Wilson, T. L., & Rood, R. 1994, *ARA&A*, 32, 191
- Wyrowski, F., Schilke, P., Walmsley, C. M., & Menten, K. M. 1999, *ApJ*, 514, L43
- Zhang, Q. 2005, in *IAU Symp.* 227, ed. R. Cesaroni, M. Felli, E. Churchwell, & M. Walmsley, 135
- Zinchenko, I., Henkel, C., & Mao, R. Q. 2000, *A&A*, 361, 1079

Online Material

Appendix A: Rotation diagrams per molecule for all sources

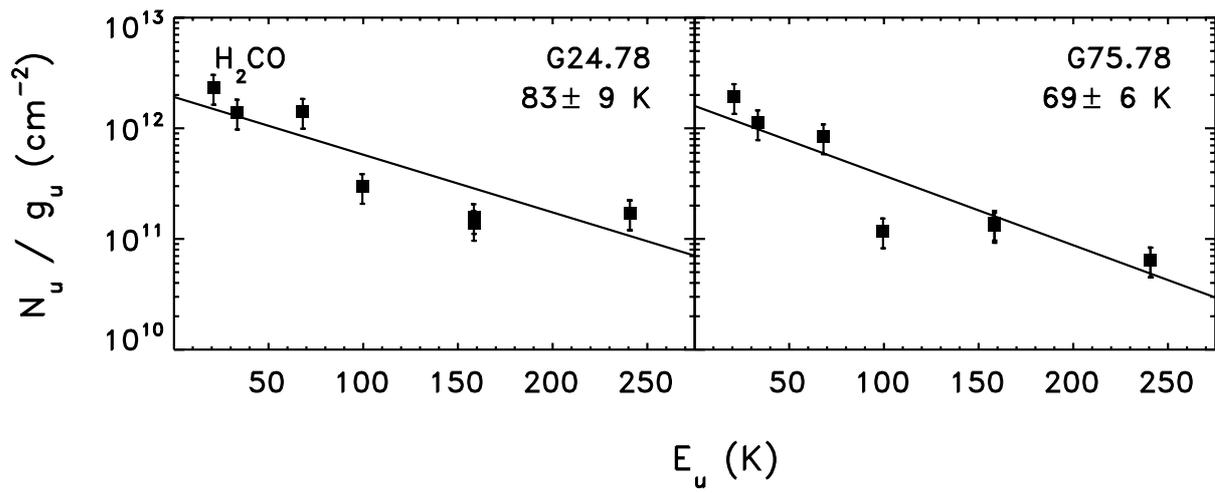


Fig. A.1. Rotation diagrams for H_2CO for G24.78, and G75.78.

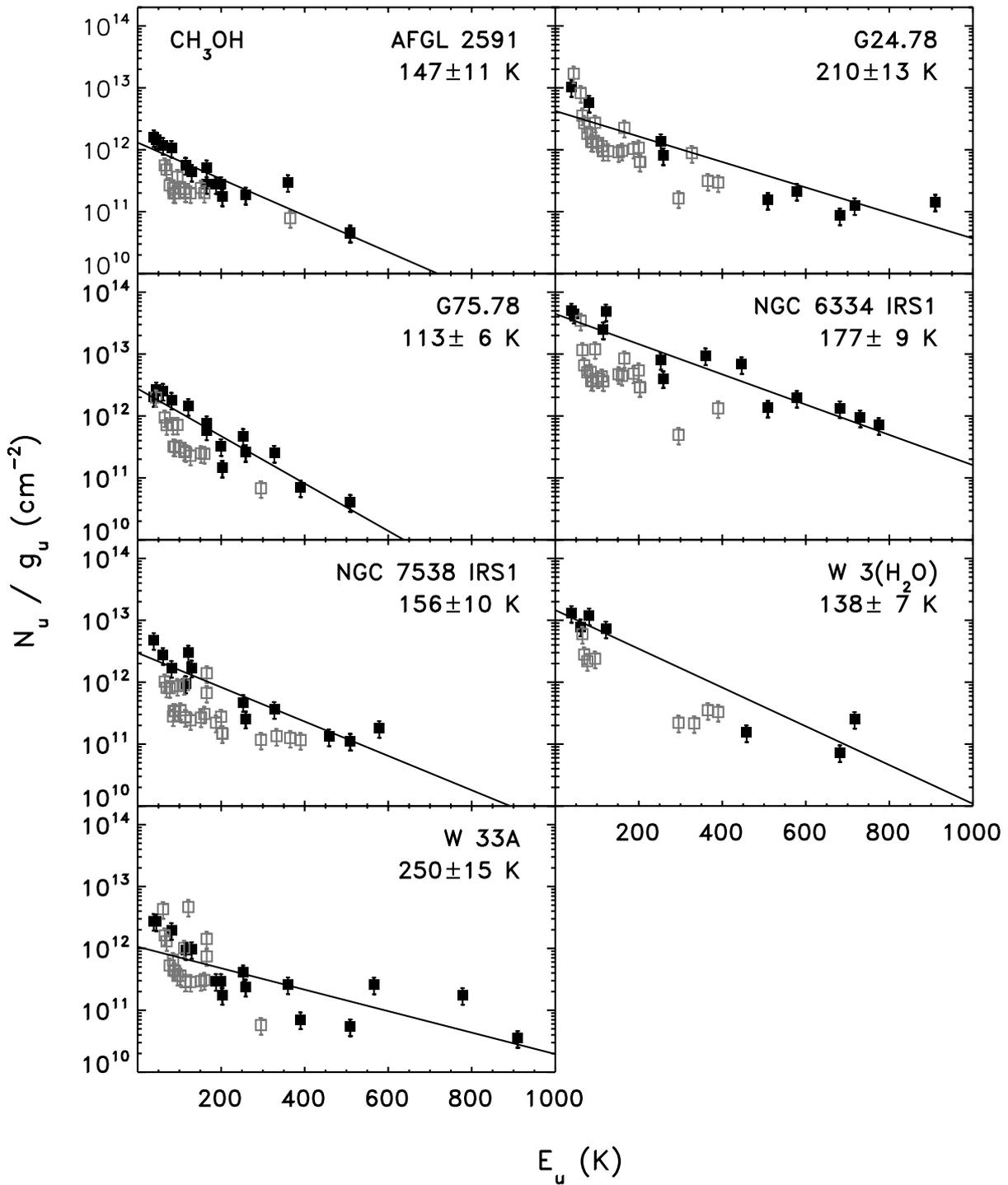


Fig. A.2. Rotation diagrams for CH_3OH for respectively AFGL2591, G24.78, G75.78, NGC 6334IRS1, NGC 7538IRS1, W3(H_2O), and W33A. The filled black squares are optically thin lines, the open gray squares optically thick lines, based on the estimated column densities. The rotation temperatures are fits to the optically thin lines only.

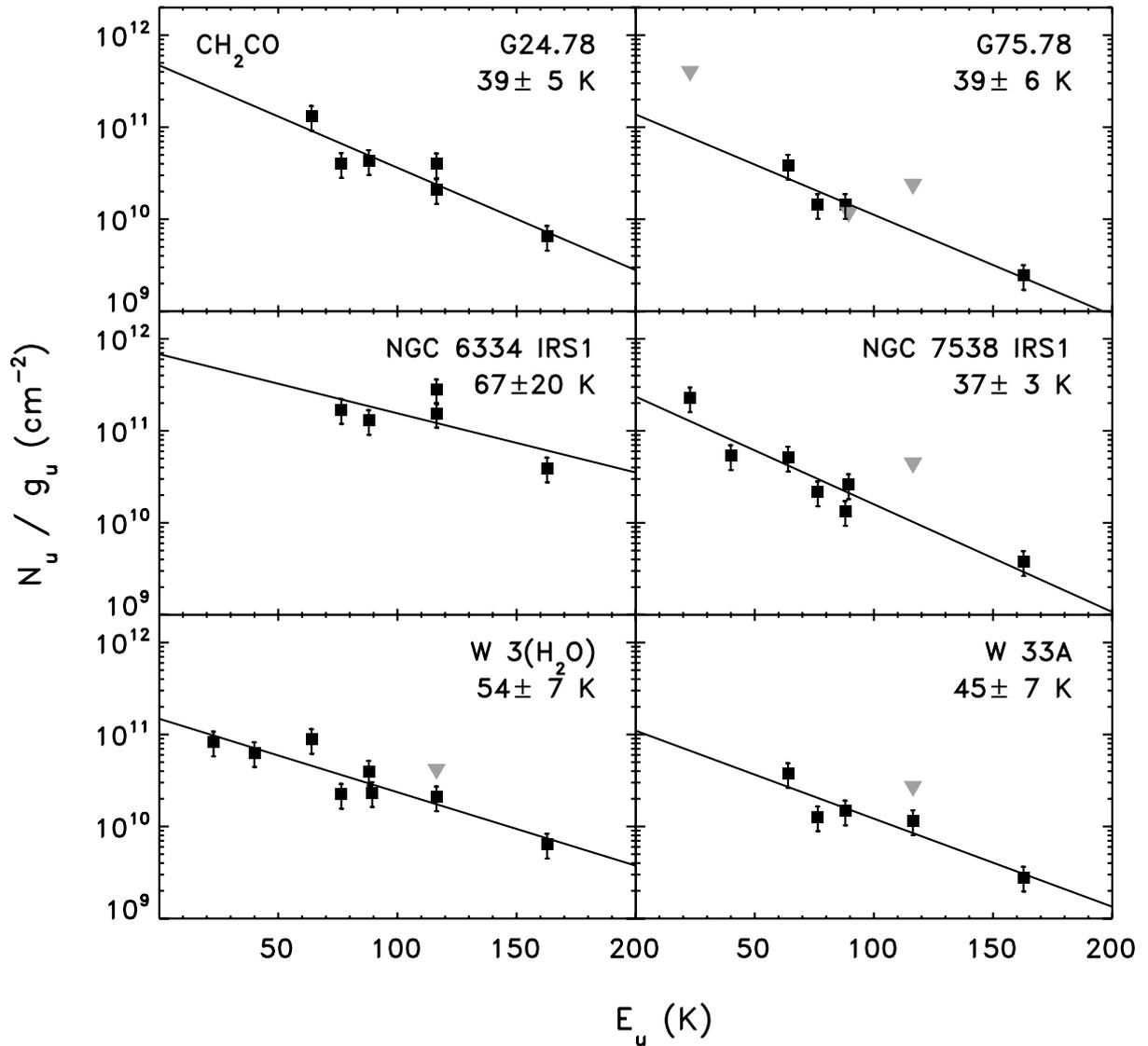


Fig. A.3. Rotation diagrams for CH_2CO for respectively G24.78, G75.78, NGC 6334IRS1, NGC 7538, W3(H_2O), and W33A. The gray triangles indicate upper limits.

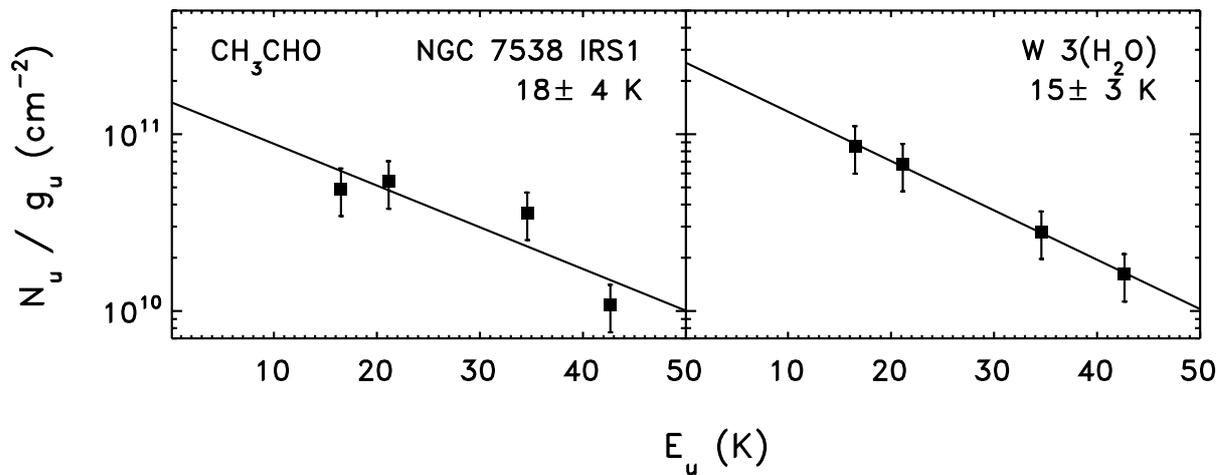


Fig. A.4. Rotation diagrams for CH_3CHO for respectively NGC 7538 IRS1, W3(H_2O), and W33A.

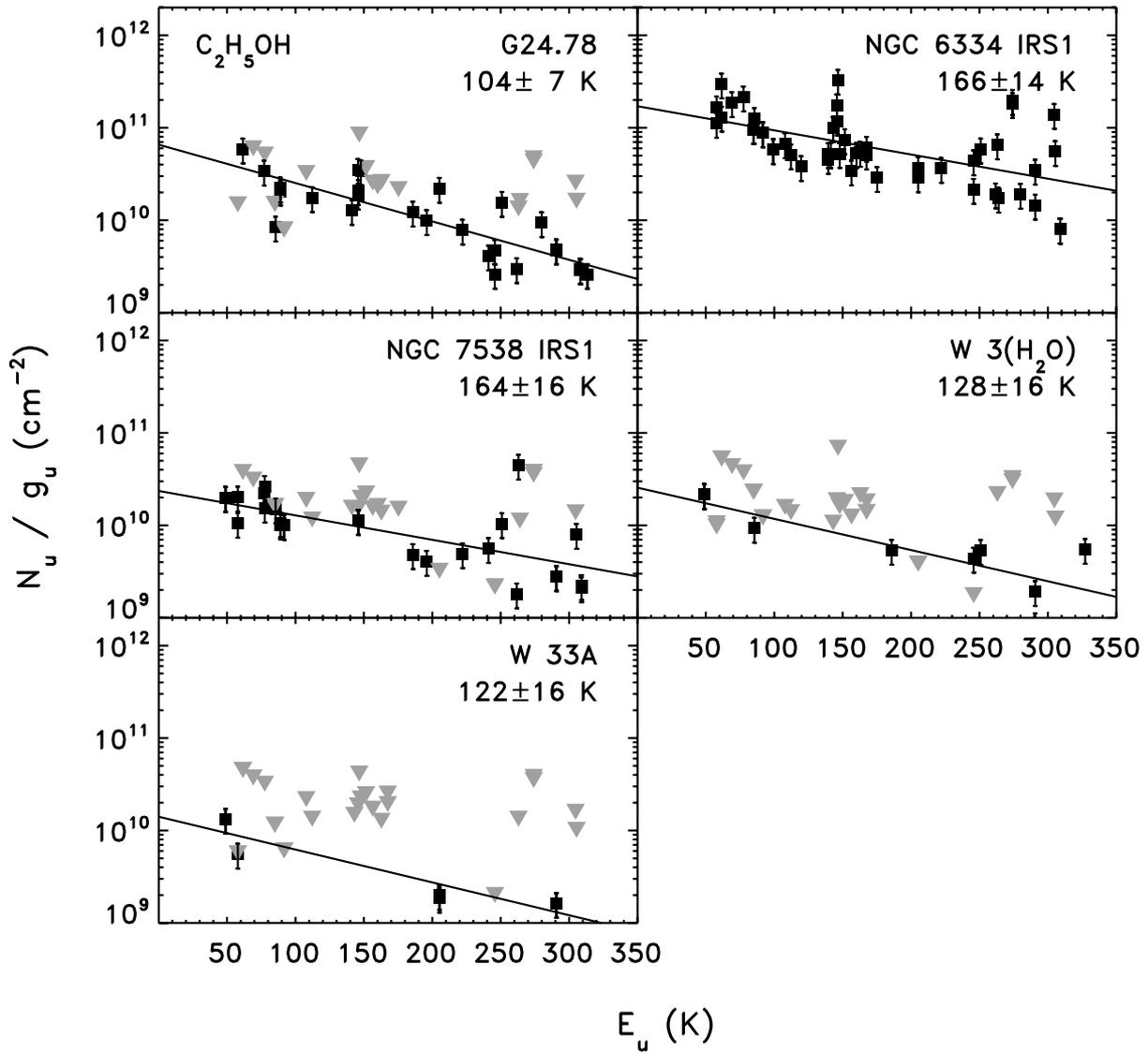


Fig. A.5. Rotation diagrams for $\text{C}_2\text{H}_5\text{OH}$ for respectively G24, NGC 6334IRS1, NGC 7538IRS1, W3(H₂O), and W33A. The gray triangles indicate upper limits.

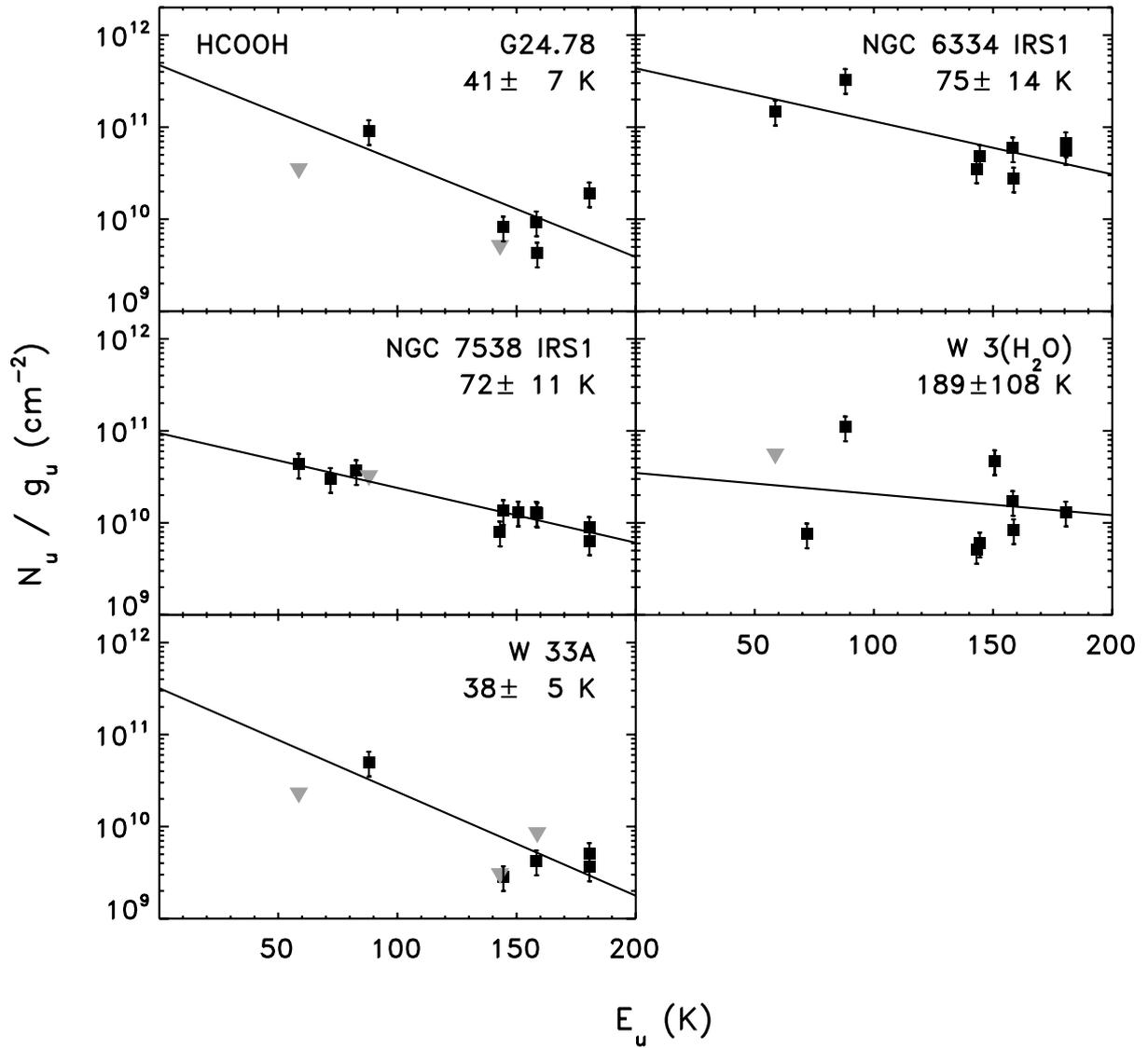


Fig. A.6. Rotation diagrams for HCOOH for respectively G24.78, NGC 6334IRS1, NGC 7538IRS1, W3(H₂O), and W33A. The gray triangles indicate upper limits.

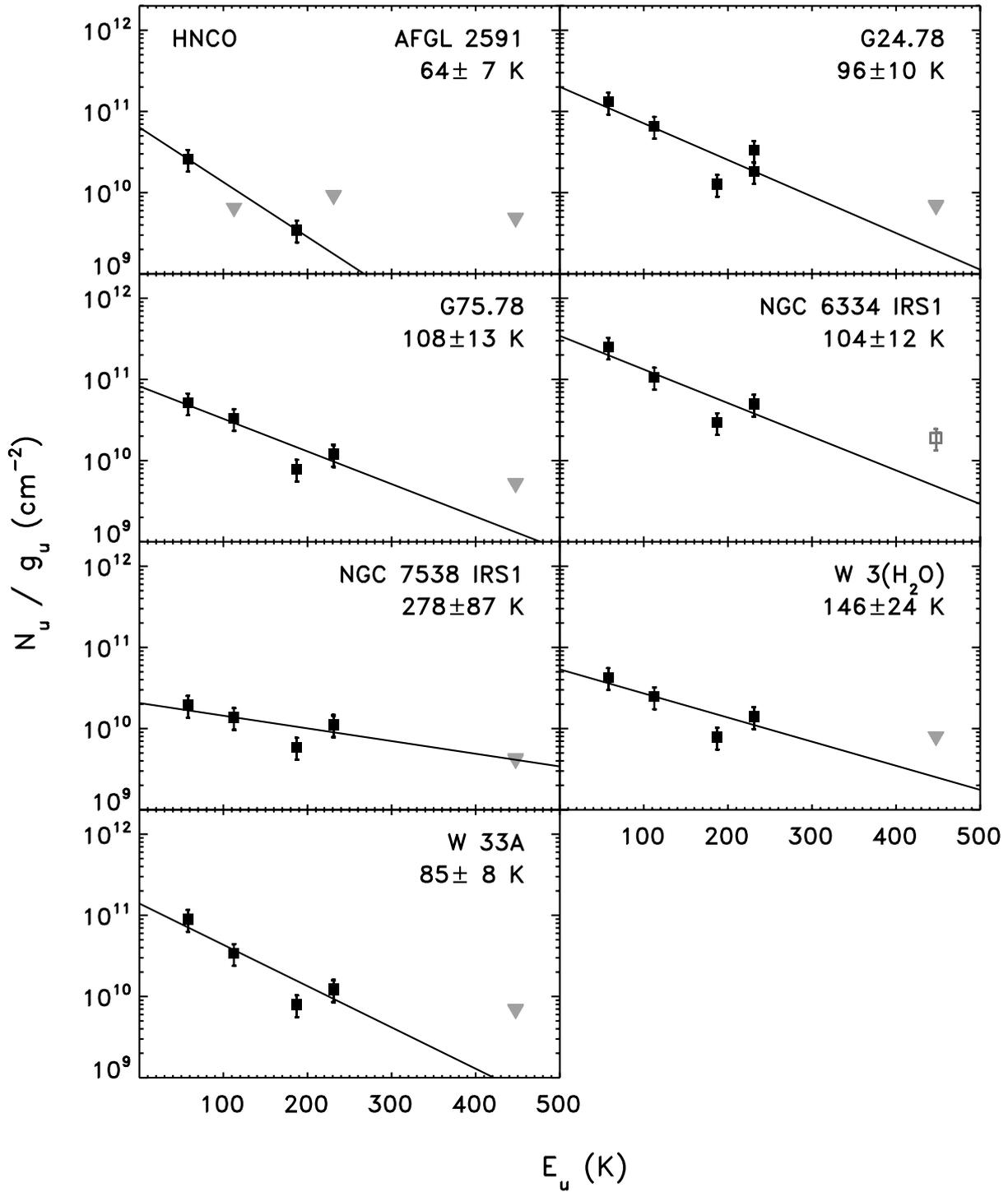


Fig. A.7. Rotation diagrams for HNC0 for respectively G24.78, G75.78, NGC 6334IRS1, NGC 7538IRS1, W3(H₂O), and W33A. The filled black squares are the points included and the gray squares excluded from the fit. The gray triangles indicate upper limits.

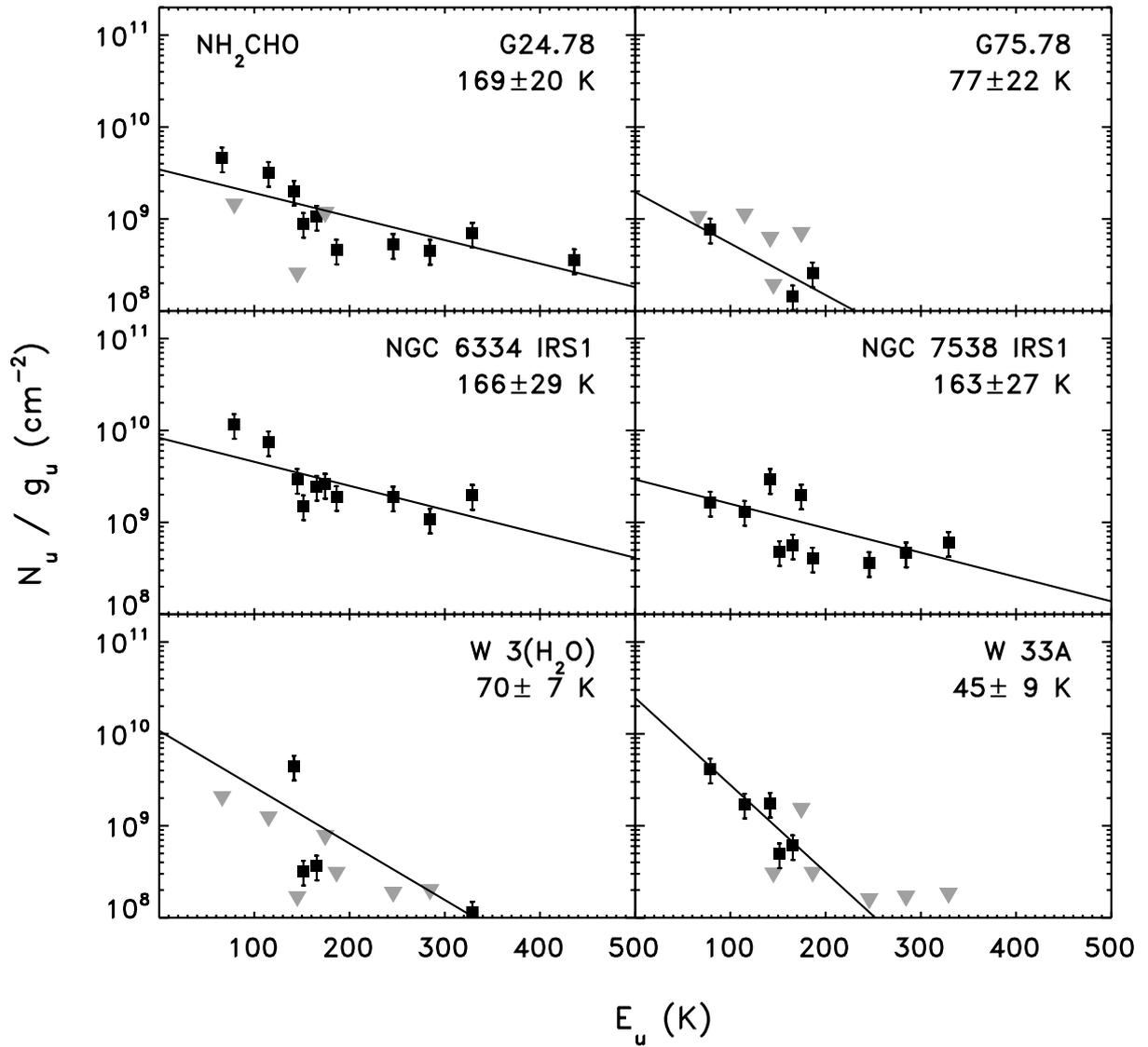


Fig. A.8. Rotation diagrams for NH_2CHO for respectively G24.78, G75.78, NGC 6334IRS1, NGC 7538IRS1, W3(H_2O), and W33A. The gray triangles indicate upper limits.

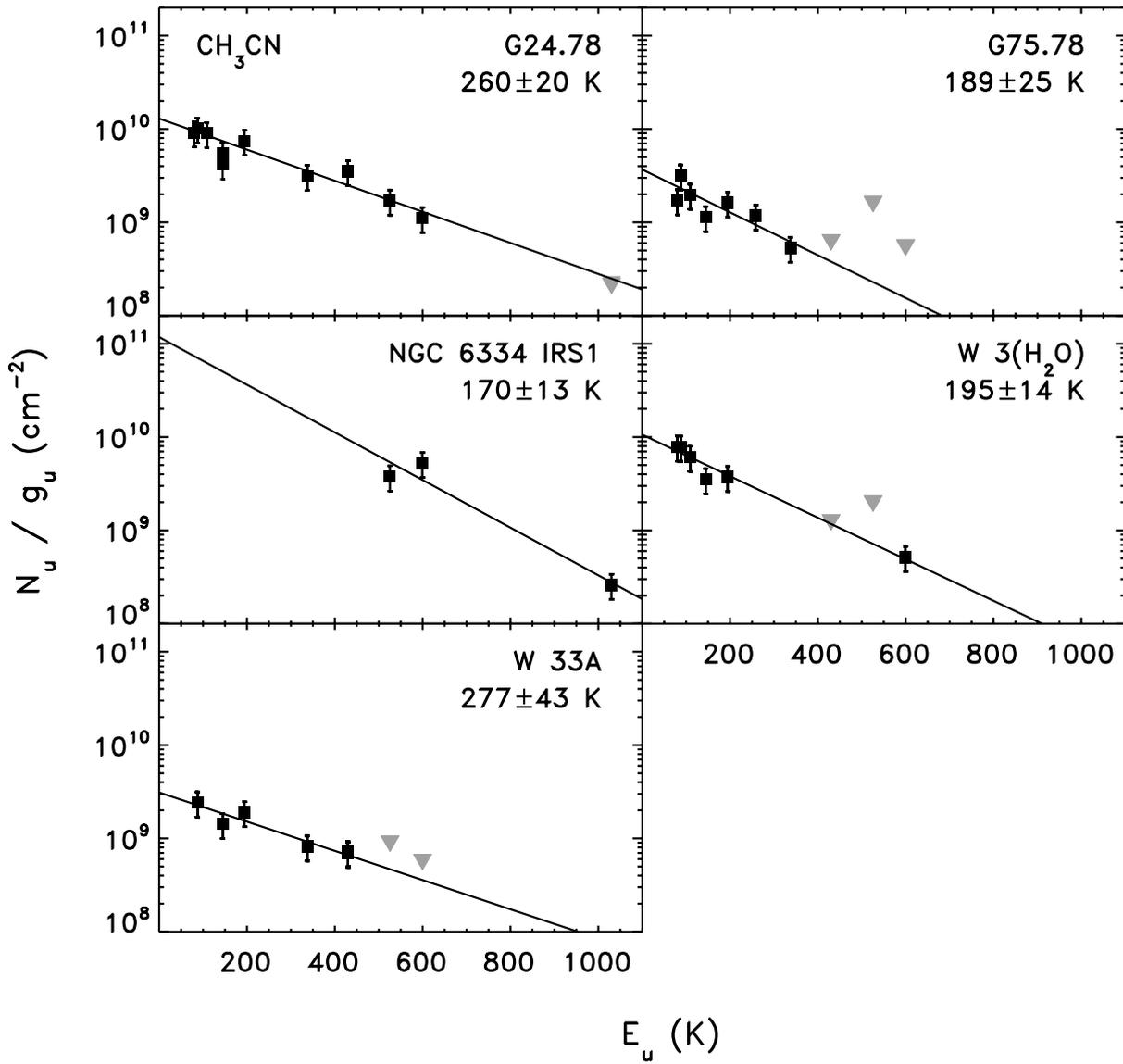


Fig. A.9. Rotation diagrams for CH₃CN for respectively G24.78, G75.78, NGC 6334IRS1, W3(H₂O), and W33A. The gray triangles indicate upper limits.

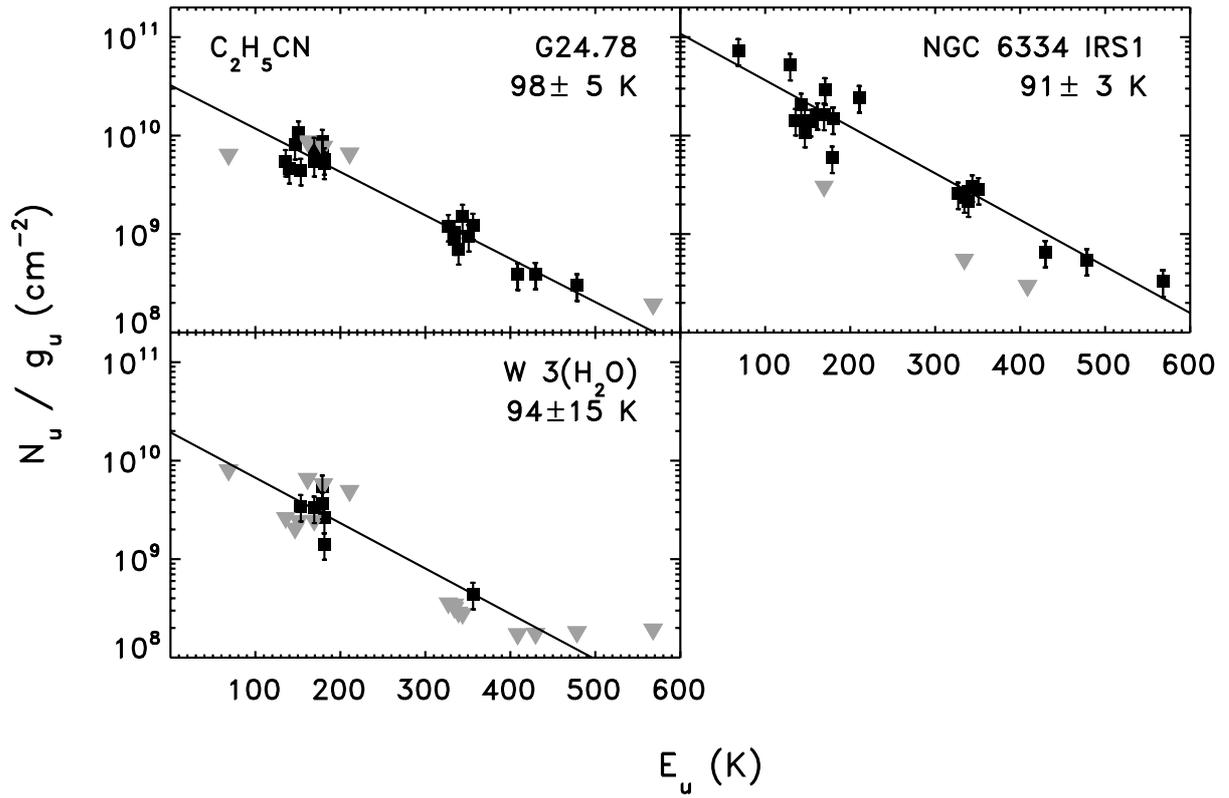


Fig. A.10. Rotation diagrams for $\text{C}_2\text{H}_5\text{CN}$ for respectively G24.78, and NGC 6334IRS1. The gray triangles indicate upper limits.

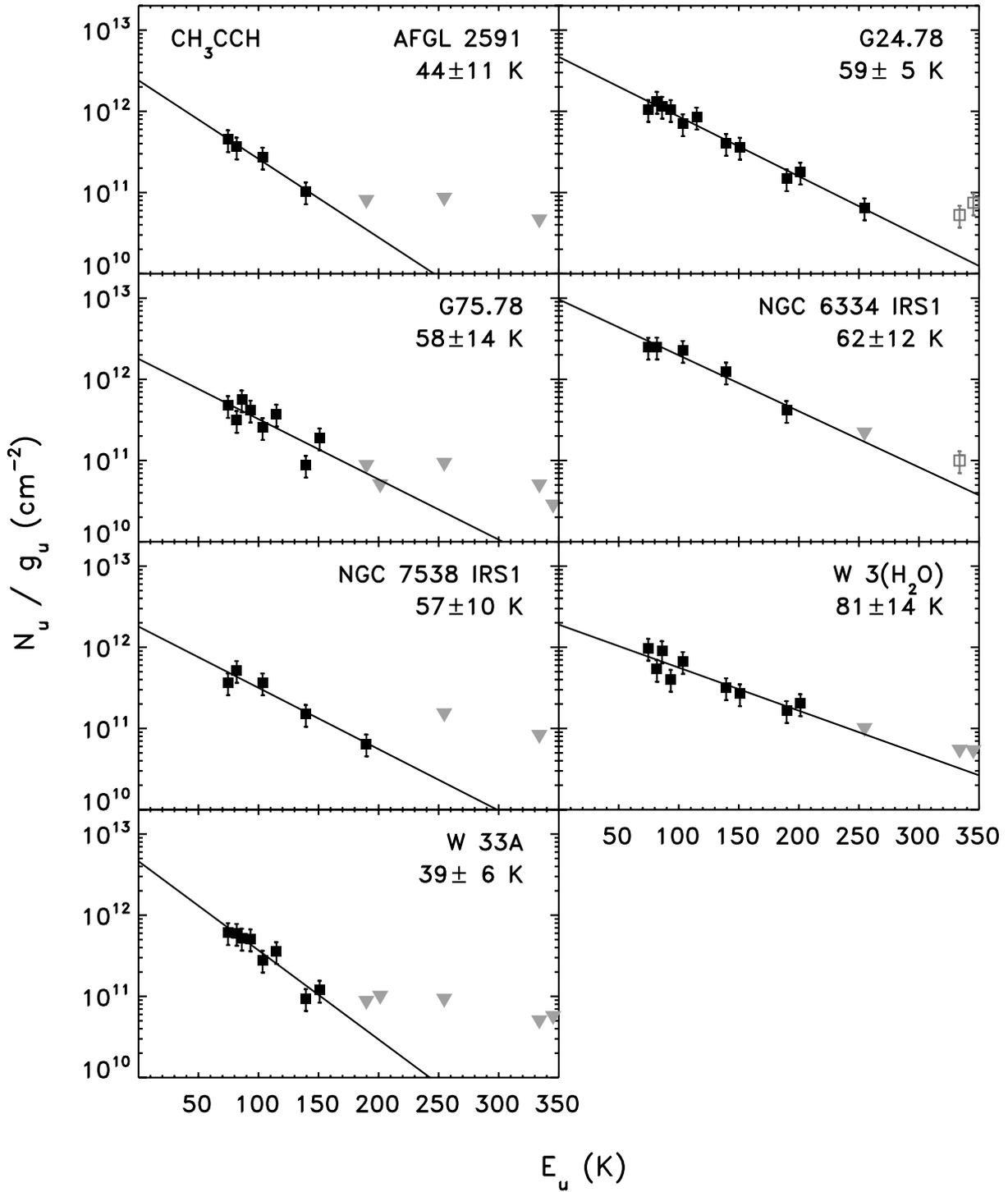


Fig. A.11. Rotation diagrams for CH_3CCH for respectively AFGL2591, G24.78, G75.78, NGC 6334IRS1, NGC 7538IRS1, W3(H_2O), and W33A. The filled black squares are the points included, the gray squares excluded from the fit and the gray triangles indicate upper limits.

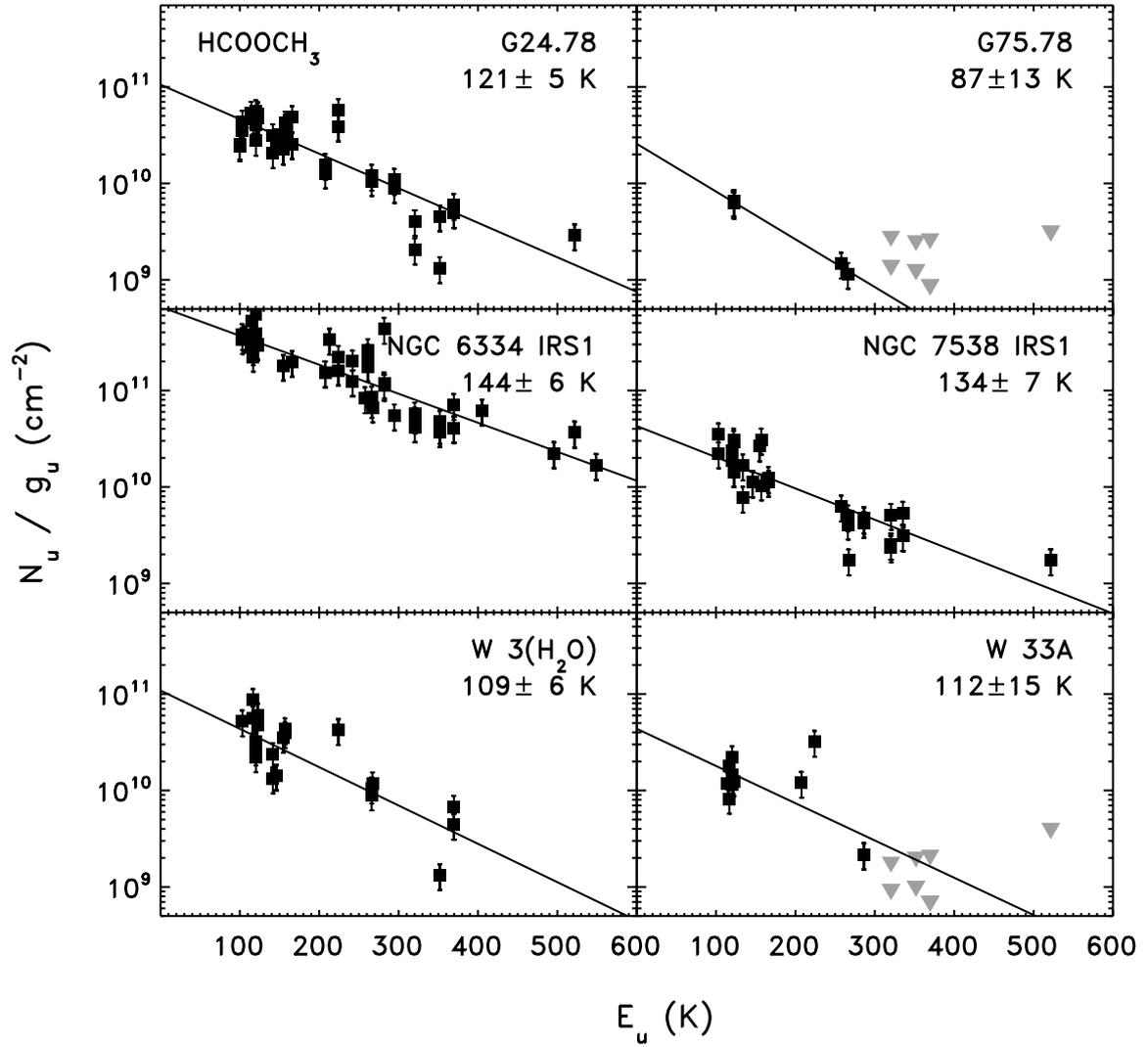


Fig. A.12. Rotation diagrams for HCOOCH_3 for respectively G24.78, G75.78, NGC 6334IRS1, NGC 7538IRS1, W3(H_2O), and W33A. The filled black squares are the points included and the gray triangles indicate upper limits.

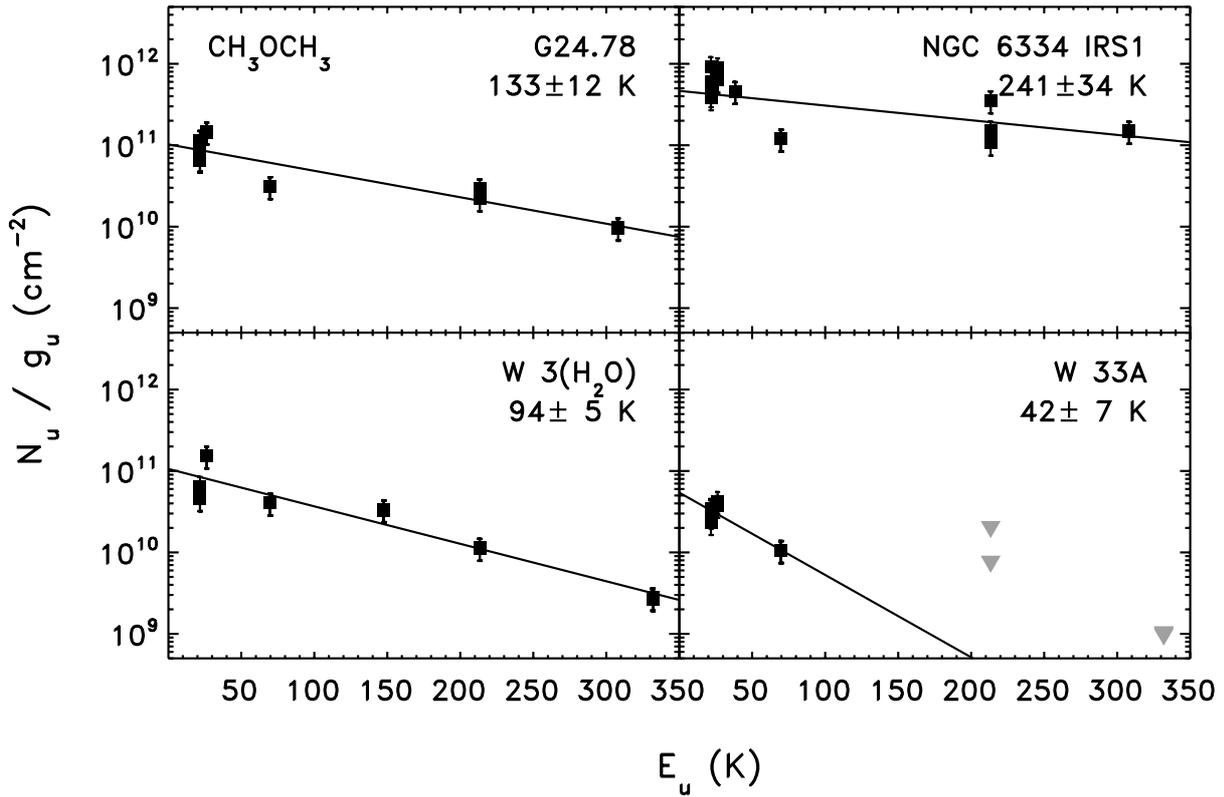


Fig. A.13. Rotation diagrams for CH_3OCH_3 for respectively G24.78, NGC 6334IRS1, W3(H_2O), and W33A. The filled black squares are the points included and the gray triangles indicate upper limits.

Appendix B: Detected lines per species for all sources**Table B.1.** Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for H₂CO and its isotopic species.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W 3(H ₂ O)	W 33A
H ₂ CO								
218.222	3 _{0,3} -2 _{0,2}	–	17	14	–	–	–	–
218.476	3 _{2,2} -2 _{2,1}	–	5.6	3.3	–	–	–	–
225.698	3 _{1,2} -2 _{1,1}	15	27	22	57	31	34	22
363.946	5 _{2,4} -4 _{2,3}	–	15	5.9	–	–	–	–
364.103	5 _{4,1/2} -4 _{4,0/1}	–	7.3	2.7	–	–	–	–
364.275	5 _{3,3} -4 _{3,2}	–	16	7.5	–	–	–	–
364.289	5 _{3,2} -4 _{3,1}	–	18	7.9	–	–	–	–
H ₂ ¹³ CO								
219.909	3 _{1,2} -2 _{1,1}	0.41	2.9	1.0	8.8	0.84	1.1	1.6
356.176	5 _{2,3} -4 _{2,2}	<0.36	2.4	0.55	6.3	<0.41	0.96	0.92
H ₂ C ¹⁸ O								
348.032	5 _{2,3} -4 _{2,2}	0.15	0.85	<0.32	2.6	0.26	<0.32	<0.26

– means frequency not observed.

Table B.2. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for CH₃OH in the 230 GHz window.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
CH ₃ OH								
218.440	4 _{3,1,0} -3 _{2,1,0}	–	12.5	3.0	–	–	–	–
219.994	23 _{14,9,0} -22 _{14,8,0}	<0.36	<0.49	<0.26	2.0	<0.32	<0.47	<0.41
220.079	8 _{4,4,0} -7 _{4,3,0}	0.55	4.1	1.1	18	1.3	3.6	0.54
227.815	16 _{9,8,0,1} -15 _{9,7,0,1}	–	2.1	0.60	–	0.87	1.8	–
229.939	19 _{12,7,0,2} -20 _{12,8,0,2}	<0.31	0.56	<0.32	5.1	0.47	<0.38	<0.25
230.027	3 _{1,3,0} -4 _{2,3,0}	0.53	3.4	0.68	17	1.6	4.3	0.92
230.293	22 ₂₂₀₀ -21 ₃₁₉₀	<0.31	0.38	<0.32	5.1 ^a	<0.41	0.68	<0.25
230.369	22 _{13,9,0} -21 _{13,8,0}	<0.31	0.26	<0.32	4.0	0.67	0.22	<0.25
231.281	10 _{6,4,0,2} -9 _{6,3,0,2}	0.34	7.6 ^b	0.93	13	0.82	4.5	1.7
232.417	10 _{6,4,0,1} -9 _{6,3,0,1}	0.64	2.8	0.72	10	<0.38	–	0.94
232.784	18 _{11,8,0,1} -17 _{11,7,0,1}	–	–	–	17	–	–	–
232.927	14 ₆₈₁ -14 ₇₇₁	–	–	–	1.7	–	–	–
232.946	10 ₄₇₀ -11 ₅₇₀	–	–	–	16	–	–	–
233.097	13 ₆₇₁ -13 ₇₆₁	–	–	–	0.83	–	–	–
234.683	4 _{3,1,0,2} -5 _{3,2,0,2}	0.62	4.3	1.3	18	1.4	4.1	2.2
234.698	5 _{1,5,0} -6 _{2,5,0,5}	<0.24	<0.36	0.31	10	0.64	1.6	1.0
240.242	5 _{4,1,0} -6 _{4,2,0,5}	0.50	3.7	0.85	12	0.79	5.6	0.92
240.741	26 _{3,23,0} -26 _{2,24,0}	<0.25	0.66	<0.3	4.8	0.47	0.33	<0.20
240.960	5 _{3,2,1,1} -4 _{3,2,1,1} ^c	0.54	3.0 ^d	0.61	17	1.1	3.4	0.94
241.043	22 _{6,16,0} -23 _{5,18,0}	<0.25	<0.40	<0.3	2.6	0.45	<0.54	<0.26
241.057	9 _{3,6,0} -10 _{0,10,0}	<0.25	<0.40	<0.3	3.3	<0.35	<0.54	<0.26
248.282	17 _{10,7,0,2} -17 _{10,8,0,1}	–	–	–	–	<0.35	3.0	–
248.885	16 _{10,7,0,2} -16 _{9,7,0,1}	0.60	2.4	0.87	–	0.96	2.7	<0.20
¹³ CH ₃ OH								
220.363	17 ₃₁₅₀ -18 ₀₁₈₀	<0.19	0.15	<0.36	1.3	<0.30	<0.46	<0.19
223.332	6 _{4,3,1,2} -7 _{5,3,1,2}	<0.25	–	–	1.3	<0.36	–	–
225.571	14 _{7,8,0} -13 _{6,8,0}	<0.32	0.64	<0.24	0.84	<0.32	0.50	<0.23
239.169	19 ₂₁₇₀ -19 ₀₁₉₀	–	<0.49	0.19	–	–	<0.38	<0.29
241.036	22 ₀₂₂₀ -21 ₃₁₈₀	<0.25	<0.40	<0.30	0.83	<0.35	<0.36	<0.42

– means frequency not observed. ^a Blend with HCOOCH₃. ^b Blend with CH₃CN. ^c Blend with 5_{2,3,2}-4_{2,3,2}. ^d Blend with HCOOCH₃.

Table B.3. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for CH₃OH in the 345 GHz window and its isotopic species.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
CH ₃ OH continued								
337.969	7 _{4,3,1,2} –6 _{4,3,1,2}	0.48	3.2	0.76	14	1.3	3.6	0.77
338.125	7 _{4,4,0} –6 _{3,3,0}	2.9	20	4.7	56	5.9	24	5.8
338.345	7 _{3,4,0} –6 _{3,4,0}	3.9	35	7.7	71	8.1	30	11
338.409	7 _{4,4,0,1} –6 _{3,3,0,1}	3.9	39	7.9	59	–	66	11
338.431	7 _{1,7,0} –6 _{0,6,0}	–	4.0	0.84	–	–	–	–
338.442	7 _{7,1,0,1/2} –6 _{6,0,0,1/2}	–	4.7	1.4	–	–	–	–
338.456	7 _{1,6,0} –6 _{1,6,0}	–	5.6	1.5	–	–	–	–
338.475	7 _{6,1,0} –6 _{6,1,0}	–	5.9	1.8	–	–	–	–
338.486	7 _{6,1,0,1/2} –6 _{6,1,0,1/2}	–	6.9	1.6	–	–	–	–
338.504	7 _{2,6,0} –6 _{1,5,0}	–	6.8	1.9	–	–	–	–
338.513	7 _{5,3,0,2} –6 _{4,2,0,2} ^a	–	14	3.3	–	–	–	–
338.530	7 _{6,2,0} –6 _{5,1,0}	–	7.2	1.8	–	–	–	–
338.542	7 _{5,2,0,1/2} –6 _{5,2,0,1}	–	17	4.7	–	–	–	–
338.560	7 _{2,5,0} –6 _{2,5,0}	–	8.6	2.1	–	–	–	–
338.583	7 _{5,2,0} –6 _{5,2,0}	–	10	2.5	–	–	–	–
338.615	7 _{4,3,0} –6 _{4,3,0}	–	20	8.0	–	–	–	–
338.640	7 _{5,3,0,1} –6 _{4,2,0,1}	–	13	3.1	–	–	–	–
338.722	7 _{3/5,5/3,0} –6 _{2/4,4/2,0}	–	27	6.5	–	–	–	–
340.141	2 _{2,0,0,1} –3 _{2,1,0,1}	0.89	5.6	1.6	27	3.2	4.1	1.7
340.394	16 _{11,5,0,1/2} –17 _{11,6,0,1/2}	0.65	2.2	0.58	19	1.6	1.1	0.78
355.964	16 _{10,7,1} –16 _{9,7,1}	<0.29	<0.47	<0.53	5.6	<0.41	<0.43	<0.68
356.007	15 _{8,7,0,2} –15 _{8,8,0,1}	2.4	9.5	3.9	29	6.8	13	3.4
¹³ CH ₃ OH								
345.084	2 _{2,0,0,1} –3 _{2,1,0,1}	<0.23	0.26	<0.30	4.1	<0.33	<0.29	<0.19
348.101	11 _{6,6,0} –10 _{6,5,0}	<0.18	2.0	0.27	10.0	0.61	<0.44	0.19

– means frequency not observed. ^a Blend with 7_{6,2,0,1/2}–6_{5,1,0,1/2}.**Table B.4.** Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for CH₂CO.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
81.586	4 _{1,3} –3 _{1,2}	0.32	–	<0.75	–	0.45	0.16	–
140.127	7 _{1,7} –6 _{1,6}	–	–	–	–	0.72	0.85	–
244.712	12 _{1,11} –11 _{1,10}	<0.39	–	<0.78	–	1.8	1.6	–
262.619	13 _{2,12} –12 _{2,11}	<0.33	–	<0.47	–	<1.7	1.5	–
220.178	11 _{1,11} –10 _{1,10}	<0.19	1.2	0.41	5.0	0.63	0.65	0.37
222.199 ^a	11 _{0,11} –10 _{0,10}	<0.31	1.3	0.38	5.9	0.51	0.87	0.37
222.229	11 _{2,10} –10 _{2,9}	<0.31	0.38	<0.22	2.7	<0.41	0.20	0.11
222.315	11 _{2,9} –10 _{2,8}	<0.31	0.20	<0.22	1.5	<0.41	<0.35	<0.25
240.187	12 _{1,12} –11 _{1,11}	<0.24	1.5	0.50	4.5	0.46	1.4	0.51
346.600	17 _{1,16} –16 _{1,15}	<0.26	1.4	0.52	8.4	0.81	1.4	0.60

– means frequency not observed. ^a Blend of lines 11_{0,11}–10_{0,10}, 11_{3,9}–10_{3,8}, and 11_{3,8}–10_{3,7}.**Table B.5.** Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for CH₃CHO.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
98.901	5 _{1,4} –4 _{1,3}	0.29	–	<0.14	–	0.22	0.38	–
112.249	6 _{1,6} –5 _{1,5}	–	–	–	–	0.40	0.50	–
149.507	8 _{1,8} –7 _{1,7}	<0.15	–	<0.36	–	0.69	0.54	–
168.093	9 _{1,9} –8 _{1,8}	<0.21	–	<0.30	–	0.30	0.48	–

– means frequency not observed.

Table B.6. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for C₂H₅OH.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
220.155	24 _{3,22,2} –24 _{2,23,2}	<0.24	<0.38	<0.36	1.1	0.78	<0.39	<0.24
222.138	19 _{7,13/12,0} –18 _{7,11/12,1}	<0.39	<0.43	<0.32	1.6	<0.32	<0.27	<0.32
225.279	13 _{5,9,1} –12 _{5,8,1}	<0.41	<0.43	<0.24	0.57	<0.26	<0.21	<0.29
225.283	13 _{5,8,1} –12 _{5,7,1}	<0.41	<0.43	<0.24	0.9	<0.26	<0.21	<0.29
225.457	13 _{3,11,1} –12 _{3,10,1}	<0.41	0.43	<0.24	0.66	<0.26	<0.21	<0.29
225.546	13 _{4,10,1} –12 _{4,9,1}	<0.41	<0.43	<0.24	0.56	<0.26	<0.21	<0.29
225.560	13 _{3,11,0} –12 _{3,10,0}	<0.41	<0.43	<0.24	1.9	<0.26	<0.21	<0.29
225.660	13 _{4,10,0} –12 _{4,9,0}	<0.41	<0.43	<0.24	0.85	<0.26	<0.21	<0.29
226.581	26 _{2,24,2} –26 _{1,25,2}	<0.30	<0.48	<0.21	2.6	<0.26	<0.35	<0.30
230.954	16 _{5,11,2} –16 _{4,12,2}	<0.32	0.59	<0.21	2.9	<0.29	<0.32	<0.42
230.991	14 _{0,14,2} –13 _{1,13,2}	<0.32	0.19	<0.21	2.8	0.34	0.21	<0.42
232.035	18 _{5,14,2} –18 _{4,15,2}	<0.30	<0.43	<0.26	0.56	<0.30	–	<0.41
232.319	23 _{5,19,2} –23 _{4,20,2}	<0.30	<0.43	<0.26	0.45	<0.30	–	<0.41
232.405	17 _{5,13,2} –17 _{4,14,2}	<0.30	<0.43	<0.26	0.95	<0.30	–	<0.41
232.491	14 _{0,14,0} –13 _{0,13,0}	<0.30	0.24	<0.26	0.99	<0.30	–	<0.41
232.809	16 _{5,12,2} –16 _{4,13,2}	–	–	<0.30	2.0	–	–	–
232.929	14 _{5,9,2} –14 _{4,10,2}	–	–	<0.30	0.54	–	–	–
233.571	13 _{5,8,2} –13 _{4,9,2}	<0.41	<0.43	<0.24	0.87	<0.25	<0.21	<0.29
234.524	25 _{5,21,2} –25 _{4,22,2}	<0.30	<0.48	<0.21	1.6	0.23	<0.35	<0.30
234.666	10 _{5,5,2} –10 _{4,6,2}	<0.30	<0.48	<0.21	<0.62	0.24	<0.35	<0.30
234.715	10 _{5,6,2} –10 _{4,7,2}	<0.30	<0.48	<0.21	2.0	0.14	<0.35	<0.30
234.853	9 _{5,4,2} –9 _{4,5,2}	<0.30	<0.48	<0.21	1.5	<0.25	<0.35	<0.30
234.984	8 _{5,3,2} –8 _{4,4,2}	<0.30	<0.48	<0.21	0.84	<0.25	<0.35	<0.30
234.992	8 _{5,4,2} –8 _{4,5,2}	<0.30	0.38	<0.21	1.9	<0.25	<0.35	<0.30
240.110	27 _{2,25,2} –27 _{1,26,2}	<0.30	<0.36	<0.26	<0.72	<0.30	0.16	<0.41
240.782	15 _{2,13,0} –14 _{1,13,1}	<0.32	<0.53	<0.30	1.1	<0.28	<0.45	<0.26
240.839	14 _{1,13,0} –13 _{1,13,1}	<0.32	<0.53	<0.30	2.0	<0.28	<0.45	<0.26
338.099	18 _{7,11,2} –18 _{6,12,2}	<0.30	1.7	<0.20	2.8	<0.25	<0.30	0.14
338.110	18 _{7,12,2} –18 _{6,13,2}	<0.30	<0.29	<0.20	2.2	<0.25	<0.30	0.15
338.163	10 _{2,8,1} –9 _{1,8,0}	<0.30	0.37	<0.20	1.1	<0.25	<0.30	<0.29
339.511	9 _{7,3/2,2} –9 _{6,4/3,2}	<0.30	0.64	<0.21	1.3	<0.28	<0.39	<0.21
339.544	8 _{7,2/1,2} –8 _{6,3/2,2}	<0.30	<0.26	<0.21	2.8	0.32	<0.39	<0.21
339.566	7 _{7,0/1,2} –7 _{6,1/2,2}	<0.30	<0.26	<0.21	1.6	<0.28	<0.39	<0.21
339.979	9 _{4,6,2} –8 _{3,5,2}	<0.30	1.2	<0.21	6.1	<0.73	<0.39	<0.21
340.189	6 _{5,1/2,2} –5 _{4,2/1,2}	<0.36	3.4	0.68	15	1.6	1.7	1.1
340.420	9 _{4,5,2} –8 _{3,6,2}	<0.36	<0.55	<0.26	4.0	0.38	<0.36	0.78
345.173	7 _{7,0/1,0} –6 _{6,0/1,1}	<0.29	<0.43	<0.30	2.4	<0.26	<0.21	<0.24
345.295	21 _{1,21,1} –20 _{1,20,1}	<0.29	0.55	<0.30	2.5	<0.26	<0.21	<0.24
345.408	21 _{0,21,1} –20 _{0,20,1}	<0.29	0.30	<0.30	5.1	<0.26	0.51	<0.24
346.620	20 _{8,12/13,0} –19 _{8,11/12,0}	<0.26	0.10	<0.36	1.3	0.34	<0.32	<0.23
346.817	20 _{7,13/14,0} –19 _{7,12/13,0}	<0.26	0.82	<0.36	3.9	0.48	0.22	0.28
346.929	20 _{6,15,1} –19 _{6,14,1}	<0.26	0.71	<0.36	1.4	<0.19	<0.32	<0.23
346.963	21 _{0,21,2} –20 _{1,20,2}	<0.26	2.0	<0.36	4.4 ^a	0.80	0.89	<0.23
347.887	20 _{5,15,0} –19 _{5,14,0}	<0.23	0.38	<0.32	2.4	0.23	<0.32	<0.26
347.916	20 _{4,17,0} –19 _{4,16,0}	<0.23	2.1	<0.32	8.1	1.4	0.74	<0.26
347.975	22 _{2,20,2} –21 _{3,19,2}	<0.23	0.57	<0.32	2.7	0.36	<0.32	<0.26
348.720	4 _{4,0/1,1} –3 _{3,0/1,0}	<0.36	0.65	<0.27	<0.36	<0.26	<0.36	<0.26
348.847	10 _{6,5/4,0} –9 _{5,5/4,1}	<0.36	0.97	<0.27	<0.36	<0.26	<0.36	<0.26
352.858	21 _{1,20,2} –20 _{2,19,2}	<0.32	1.2	<0.51	4.9	0.49	<0.29	<0.42
353.035	12 _{3,9,2} –11 _{2,10,2}	<0.32	0.97	<0.51	5.2	0.64	0.58	<0.42
363.968	21 _{7,15/14,1} –20 _{7,14/13,1}	–	0.50	<0.38	–	–	–	–
364.233	21 _{7,15/14,0} –20 _{7,14/13,0}	–	0.57	<0.38	–	–	–	–

– means frequency not observed. ^a Blend with C₂H₅CN.

Table B.7. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for HCOOH.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
223.916	10 _{2,9} -9 _{1,8}	<0.21	–	<0.17	–	0.76	0.19	–
247.514	11 _{5,6/7} -10 _{5,5/6}	–	–	–	–	0.66	2.4	–
257.975	12 _{1,12} -11 _{1,11}	<0.38	–	<1.9	–	<0.89	<0.95	–
262.103	12 _{0,12} -11 _{0,11}	–	–	–	–	<0.36	<1.3	–
220.038	10 _{0,10} -9 _{0,9}	<0.19	<0.29	<0.29	1.3	0.37	<0.46	<0.19
225.513	10 _{3,7} -9 _{3,6}	<0.32	0.73	<0.19	2.6	<0.25	0.88	0.40
338.144	15 _{4,12} -14 _{4,11}	0.23 ^a	2.1 ^a	<0.16	3.7	0.49 ^a	1.2 ^a	0.28
338.202	15 _{3,13} -14 _{3,12}	<0.24	0.53	0.26	3.4	0.74	0.97	0.24
338.249	15 _{4,11} -14 _{4,10}	<0.24	1.1	<0.16	3.1	0.35	0.72	0.20
345.031	16 _{0,16} -15 _{0,15}	<0.23	<0.32	<0.20	2.3	0.51	0.33	<0.19
346.719	15 _{2,13} -14 _{2,12}	0.21	0.49	0.19	2.9	0.81	0.36	0.17
356.137	16 _{2,15} -15 _{2,14}	0.65	0.28	0.34	1.8	0.83	0.55	<0.54

 – means frequency not observed. ^a Blend with C₂H₅CN.

Table B.8. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for HNCO.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
HNCO								
219.656	10 _{3,7/8} -9 _{3,6/7}	<0.29	<0.41	<0.31	1.2	<0.25	<0.47	<0.41
219.734	10 _{2,9} -9 _{2,8}	<0.29	0.59	0.77	3.2	0.72	0.90	0.78
219.737	10 _{2,8} -9 _{2,7}	<0.29	1.1	^a	^a	^a	^a	^a
219.798	10 _{0,10} -9 _{0,9}	0.88	4.4	1.7	8.4	0.65	1.4	3.0
240.876	11 _{1,11} -10 _{1,10}	<0.25	2.7	1.3	4.3	0.55	0.99	1.4
352.898	16 _{1,15} -15 _{1,14}	0.89	3.3	2.0	7.7	1.5	2.0	2.1
HN ¹³ CO								
219.803	10 _{2,9} -9 _{2,8}	<0.36	<0.41	<0.26	<0.98	<0.32	<0.39	<0.41
240.881	11 _{1,11} -10 _{1,10}	<0.25	<0.46	<0.36	<0.41	<0.28	<0.54	<0.26

 – means frequency not observed. ^a Blend with HNCO line at 219734 MHz.

Table B.9. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for NH₂CHO.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
227.606	11 _{0,11} -10 _{0,10}	–	0.90	<0.20	–	^a	<0.39	–
232.274	11 _{2,10} -10 _{2,9}	<0.5	<0.27	0.15	2.2	0.32	–	<0.60
233.529	11 _{6,5/6} -10 _{6,4/5}	<0.15	<0.32	<0.19	0.73	0.56	<0.21	<0.42
233.595	11 _{5,6/7} -10 _{5,5/6}	<0.15	0.64	<0.19	^b	0.93	1.4	0.56
233.736	11 _{4,8} -10 _{4,7}	<0.15	0.56	<0.19	1.3	0.23	<0.21	0.30
339.716	16 _{8,8/9} -15 _{8,7/8}	<0.3	1.3	<0.17	3.8	1.2	0.22	<0.34
339.780	16 _{7,9/10} -15 _{7,8/9}	<0.3	0.93	<0.17	2.2	0.95	<0.39	<0.34
339.904	16 _{6,10/11} -15 _{6,9/10}	<0.3	1.2	<0.17	4.1	0.81	<0.39	<0.34
340.491	16 _{3,14} -15 _{3,13}	<0.36	1.3	0.18	3.0	0.70	0.45	0.75
340.536	16 _{4,13} -15 _{4,12}	<0.36	0.55	0.31	2.3	0.49	<0.36	<0.36
345.183	17 _{0,17} -16 _{0,16}	<0.29	1.2	<0.24	2.1	0.66	0.44	0.34
345.327	16 _{1,15} -15 _{1,14}	<0.29	<0.32	<0.24	3.8	2.2	<0.21	<0.38

 – means frequency not observed. ^a Blend with C₂H₅OH. ^b Blend with C₂H₅OH at 225404 MHz in other sideband.

Table B.10. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for CH₃CN.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
CH ₃ CN								
220.476	12 _{8,0,K} –11 _{8,0,L}	<0.24	0.45	<0.43	1.0	<0.54	<0.53	<0.24
238.913	13 _{7,0,K} –12 _{7,0,L}	–	1.4	<0.25	–	–	<0.50	0.38
238.972	13 _{6,0,K} –12 _{6,0,K}	–	2.8	0.47	–	–	1.8	0.73
239.064	13 _{4,0,K} –12 _{4,0,L}	–	3.8	0.83	–	–	1.9	0.97
239.097	13 _{3,0,K} –12 _{3,0,L}	–	5.9	1.2	–	–	3.8	1.5
239.120	13 _{2,0,K} –12 _{2,0,L}	–	5.0	1.1	–	–	3.4	1.5
239.133	13 _{1,0,K} –12 _{0/1,0,L}	–	5.7	1.8	–	–	8.1	1.4
239.138	13 _{0,0,K} –12 _{0,0,L}	–	5.2	0.98	–	–	<0.50	0.69
240.090	13 _{1,2,K} –12 _{1,2,L}	<0.3	0.94	<0.31	3.0	^a	0.29	<0.32
348.645	19 _{11,0,K} –18 _{11,0,L}	<0.36	<0.36	<0.43	0.64	<0.59	<0.58	<0.24
CH ₃ ¹³ CN								
220.485	12 _{6,0,K} –11 _{6,0,L}	<0.24	0.30	<0.43	0.35	<0.54	<0.53	<0.24
238.905	13 _{5,0,K} –12 _{5,0,L}	–	0.17	<0.25	–	–	<0.50	<0.34
238.946	13 _{4,0,K} –12 _{4,0,L}	–	0.57	<0.25	–	–	<0.50	<0.34
238.978	13 _{3,0,K} –12 _{3,0,L}	–	1.06	<0.25	–	–	<0.50	0.38
239.001	13 _{2,0,K} –12 _{2,0,L}	–	0.73	<0.25	–	–	<0.50	<0.34
348.853	19 _{8,0,K} –18 _{8,0,L}	<0.36	0.36	<0.43	<0.36	<0.59	<0.58	<0.24

– means frequency not observed. K refers to main transition of the upper level i.e. 12, 13, or 19 +1, +0, or –1 L refers to the main transition of the lower level i.e. 11, 12, or 18 +1, +0, or –1. ^a Blend with CH₃OH.

Table B.11. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for C₂H₅CN.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
218.390	24 _{3,21} –23 _{3,20}	–	0.71	<0.41	–	–	–	–
219.506	24 _{2,22} –23 _{2,21}	<0.36	0.85	<0.26	2.2	<0.32	<0.39	<0.41
222.918	25 _{1,24} –24 _{1,23}	<0.32	–	–	3.4	<0.36	–	–
223.385	26 _{1,26} –25 _{1,25}	<0.32	–	–	1.9	<0.36	–	–
227.781	25 _{3,22} –24 _{3,21}	–	1.8	<0.26	–	<0.32	<0.39	–
228.483	25 _{2,23} –24 _{2,22}	<0.24	1.4	<0.36	2.4	<0.30	<0.33	<0.24
231.310	26 _{1,25} –25 _{1,24}	<0.32	0.79	<0.21	2.5	<0.36	0.61	<0.42
233.443	26 _{5,22} –25 _{5,21}	<0.15	1.2	<0.24	1.0	<0.41	0.94	<0.29
233.498	26 _{5,21} –25 _{5,20}	<0.15	1.5	<0.24	1.9 ^a	<0.41	0.63	<0.29
240.319	28 _{1,28} –27 _{1,27}	<0.30	1.4	<0.26	3.2	<0.38	0.66	<0.41
240.429	28 _{0,28} –27 _{0,27}	<0.30	1.1	<0.26	<0.58	<0.38	<0.47	<0.41
248.781	29 _{1,29} –28 _{1,28}	<0.32	1.1	<0.30	–	<0.35	0.60	<0.26
248.869	29 _{0,29} –28 _{0,28}	<0.32	1.2	<0.30	–	<0.35	1.1	<0.26
339.895	39 _{2,38} –38 _{2,37}	<0.30	1.2	<0.21	2.8	<0.36	<0.39	<0.21
339.968	38 _{2,36} –37 _{2,35}	<0.30	1.4	<0.21	2.9	<0.36	<0.39	<0.21
340.149	39 _{1,38} –38 _{1,37}	<0.36	1.0	<0.26	<0.62	<0.41	<0.36	<0.23
340.433	38 _{12,26/27} –37 _{12,25/26}	<0.36	0.63	<0.26	1.1	<0.41	<0.36	<0.23
340.483	38 _{10,28/29} –37 _{10,27/28}	<0.36	0.83	<0.26	1.4	<0.41	<0.36	<0.23
340.551	38 _{15,23/24} –37 _{15,22/23}	<0.36	<0.35	<0.26	0.63	<0.41	<0.36	<0.23
340.576	38 _{9,29/30} –37 _{9,28/29}	<0.36	0.85	<0.26	<0.62	<0.41	<0.36	<0.23
346.712	25 _{8,17/18} –25 _{7,18/19}	<0.26	<0.47	<0.36	1.9	<0.24	<0.32	<0.23
346.864	22 _{8,14/15} –22 _{7,15/16}	<0.26	<0.47	<0.36	0.96	<0.24	<0.32	<0.23
346.874	38 _{4,34} –37 _{4,33}	<0.26	0.82	<0.36	2.4	<0.24	<0.32	<0.23
346.898	21 _{8,13/14} –21 _{7,14/15}	<0.26	0.49	<0.36	1.8	<0.24	<0.32	<0.23
346.925	20 _{8,12/13} –20 _{7,13/14}	<0.26	<0.47	<0.36	0.93	<0.24	<0.32	<0.23
348.261	39 _{2,37} –38 _{2,36}	<0.23	1.8	<0.21	3.6	<0.27	<0.32	<0.26
348.345	40 _{1,39} –39 _{1,38}	<0.23	1.2	<0.21	3.5	<0.27	<0.32	<0.26
353.089	23 _{3,20} –22 _{2,21}	<0.32	0.27	<0.51	0.58	<0.30	<0.29	<0.42
353.235	11 _{6,5/6} –10 _{5,6/5}	<0.32	<0.23	<0.51	2.9	<0.30	<0.29	<0.42

– means frequency not observed. ^a Blend with NH₂CHO.

Table B.12. Observed line fluxes $\int T_{\text{MB}} dV$ (K km s⁻¹) for CH₃CCH.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
222.014	13 ₆ –12 ₆	<0.23	<0.27	<0.25	0.51	<0.41	<0.27	<0.25
222.061	13 ₅ –12 ₅	<0.23	0.18	<0.25	<0.59	<0.41	<0.27	<0.25
222.099	13 ₄ –12 ₄	<0.23	0.44	<0.25	1.2	0.19	0.49	<0.25
222.129	13 ₃ –12 ₃	0.63	2.5	0.54	7.6	0.92	2.0	0.58
222.150	13 ₂ –12 ₂	0.87	2.2	0.81	7.2	1.2	2.1	0.89
222.163	13 ₁ –12 ₁	1.2	4.3	1.0	8.1	1.7	1.7	1.9
222.167	13 ₀ –12 ₀	1.5	3.4	1.5	8.1	1.2	3.2	2.0
239.088	14 ₆ –13 ₆	–	0.46	<0.17	–	–	<0.32	<0.34
239.138 ^a	14 ₅ –13 ₅	–	0.69	0.98	–	–	<0.32	<0.34
239.179	14 ₄ –13 ₄	–	0.62	<0.17	–	–	0.70	<0.34
239.211	14 ₃ –13 ₃	–	2.6	1.4	–	–	1.9	0.86
239.234	14 ₂ –13 ₂	–	3.2	1.4	–	–	8.0 ^b	1.3
239.248	14 ₁ –13 ₁	–	4.0	1.6	–	–	1.5	1.9
239.252	14 ₀ –13 ₀	–	4.4	2.1	–	–	3.4	2.0

– means frequency not observed. ^a Blend with CH₃CN. ^b Blend with OCS at 231062 MHz in the main sideband.

Table B.13. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for HCOOCH₃ for the 230 GHz band.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
211.537	17 _{6,11} –16 _{6,10} E	<0.36	–	–	8.6	–	–	<0.14
211.575	17 _{6,11} –16 _{6,10} A	<0.36	–	–	6.3	–	–	0.19
211.771	17 _{5,13} –16 _{5,12} E	<0.36	–	–	6.2	–	–	<0.14
211.785	17 _{5,13} –16 _{5,12} A	<0.36	–	–	6.2	–	–	<0.14
218.281	17 _{1,16} –16 _{1,15} E	–	0.48	<0.41	–	–	–	–
218.297	17 _{3,14} –16 _{3,13} A	–	0.46	<0.41	–	–	–	–
220.167	17 _{2,15} –16 _{4,12} E	<0.24	0.82	<0.36	7.1	0.42	0.98	<0.24
220.190	17 _{4,13} –16 _{4,12} A	<0.24	0.67	<0.36	6.4	0.66	<0.33	<0.24
221.979	18 _{9,9/10} –17 _{9,8/9} A	<0.39	0.71	<0.32	5.7	0.84	1.1	<0.32
223.125	18 _{7,11} –17 _{7,10} E	<0.32	–	–	7.2	0.30	–	–
223.135	18 _{7,12} –17 _{7,11} E	<0.32	–	–	8.9	<0.43	–	–
223.163	18 _{7,11} –17 _{7,10} A	<0.32	–	–	<1.3	0.14	–	–
225.609	19 _{3,16} –18 _{3,15} E	<0.41	<0.27	<0.24	4.9	<0.32	1.2	0.39
225.619	19 _{3,17} –18 _{3,16} A	<0.41	1.0	<0.24	5.7	<0.32	1.9	0.18
226.713	20 _{2,18} –19 _{2,18} E	<0.30	0.65	<0.21	9.1	0.56	0.60	<0.30
226.719	20 _{2,19} –19 _{2,18} A	<0.30	1.3	<0.21	14	0.66	0.73	0.52
226.773	20 _{3,18} –19 _{3,17} E	<0.30	<0.30	<0.21	<1.4	0.44	<0.35	<0.30
226.779	20 _{1,19} –19 _{1,18} A	<0.30	<0.30	<0.21	<1.4	0.53	<0.35	<0.30
227.561	21 _{1,21} –20 _{1,20} E ^a	–	5.3	0.73	–	1.4	6.1	0.34
233.247	19 _{16,3/4} –18 _{16,2/3} A	<0.41	<0.27	<0.24	1.5	<0.32	<0.21	<0.29
233.256	19 _{16,3} –18 _{16,2} E	<0.41	<0.27	<0.24	3.0	<0.32	<0.21	<0.29
233.269	19 _{16,4} –18 _{16,3} E	<0.41	<0.27	<0.24	1.3	<0.32	<0.21	<0.29
233.310	19 _{15,4/5} –18 _{15,3/4} A	<0.41	<0.27	<0.24	3.1	<0.32	<0.21	<0.29
233.316	19 _{15,4} –18 _{15,3} E	<0.41	<0.27	<0.24	2.0	<0.32	<0.21	<0.29
233.331	19 _{15,5} –18 _{15,4} E	<0.41	<0.27	<0.24	2.3	<0.32	<0.21	<0.29
233.395	19 _{14,5/6} –18 _{14,4/5} A ^b	<0.41	<0.27	<0.24	4.0	<0.32	<0.21	<0.29
233.414	19 _{14,6} –18 _{14,5} E	<0.41	<0.27	<0.24	2.1	<0.32	<0.21	<0.29
233.505	19 _{13,6} –18 _{13,5} E ^c	<0.41	1.4	<0.24	6.1	<0.32	1.6	<0.29
233.525	19 _{13,7} –18 _{13,6} E	<0.41	0.72	<0.24	2.8	<0.32	<0.21	0.40
233.649	19 _{12,7} –18 _{12,6} E	<0.41	0.22	<0.24	<0.74	<0.32	<0.21	<0.29
233.655	19 _{12,7/8} –18 _{12,6/7} A	<0.41	1.5 ^d	<0.24	4.4	<0.32	<0.21	<0.29
233.671	19 _{12,8} –18 _{12,7} E	<0.41	0.18	<0.24	2.3	<0.32	<0.21	0.17
233.754	18 _{2,16} –17 _{2,15} E	<0.41	1 .1	<0.24	7.5	<0.32	<0.21	<0.29
234.502	19 _{9,10/11} –18 _{9,9/10} A	<0.30	0.94	<0.21	7.3	0.41	<0.35	<0.30
240.021	19 _{1,18} –18 _{1,17} E	<0.30	1.1	<0.26	6.8	0.72	<0.42	<0.41
240.035	19 _{3,16} –18 _{3,15} A	<0.30	1.1	<0.26	7.0	0.68	1.1	0.29
248.617	20 _{7,14} –19 _{7,13} A	<0.32	0.69	<0.30	–	<0.35	0.91	<0.26
248.634	20 _{7,14} –19 _{7,13} E	<0.32	0.94	<0.30	–	<0.35	0.96	<0.26
248.744	20 _{7,13} –19 _{7,12} E	<0.32	<0.33	<0.30	–	0.69	<0.45	<0.26
248.750	23 _{2,22} –22 _{1,22} E ^e	<0.32	3.8	<0.30	–	1.3	2.0	<0.26
248.787	20 _{7,13} –19 _{7,12} A	<0.32	<0.33	<0.30	–	0.24	<0.45	<0.26
249.031	20 _{5,16} –19 _{5,15} E	<0.32	0.51	<0.30	–	<0.35	0.33	<0.26
249.047	20 _{5,16} –19 _{5,15} A	<0.32	0.77	<0.30	–	<0.35	0.59	<0.26

– means frequency not observed. ^a Blend with 21_{2,20}–20_{2,19} E, 21_{1,21}–20_{1,20} A, and 21_{0,21}–20_{0,20} A. ^b Blend with line at 233397 19_{14,5}–18_{14,4} E. ^c Blend with 19_{13,6}–18_{13,5} A, and 19_{13,7}–18_{13,6} A. ^d Blend with C₂H₅CN. ^e Blend with 23_{1,23}–22_{1,22} E, 23_{2,22}–22_{2,21} E, 23_{1,23}–22_{2,21} E.

Table B.14. Observed line fluxes $\int T_{\text{MB}}dV$ (K km s⁻¹) for HCOOCH₃ in the 345 GHz band.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
338.356	27 _{8,19} –26 _{8,18} E	<0.30	<0.30	<0.20	8.8	0.23	1.6	<0.29
338.397	27 _{7,21} –26 _{7,20} E	<0.30	<0.30	0.13	11	0.85	<0.30	<0.29
340.221	31 _{13,19} –31 _{12,19} E ^a	<0.36	<0.35	<0.26	2.7	<0.41	<0.36	<0.23
345.069	28 _{14,14} –27 _{14,13} E ^b	<0.29	1.7	<0.30	14	<0.33	1.5	<0.24
345.090	28 _{14,15} –27 _{14,14} E	<0.29	0.69	<0.30	8.2	<0.33	0.78	<0.24
345.281	18 _{13,5/6} –18 _{12,6/7} A	<0.29	<0.27	<0.30	3.5	<0.33	<0.21	<0.24
345.462	28 _{13,15} –27 _{13,14} E	<0.29	<0.27	<0.30	4.8	<0.33	<0.21	<0.24
345.467	28 _{13,15/16} –27 _{13,14/15} A	<0.29	1.1	<0.30	8.9	<0.33	0.32	<0.24
345.486	28 _{13,16} –27 _{13,15} E	<0.29	0.16	<0.30	5.7	<0.33	<0.21	<0.24
345.975	28 _{12,16} –27 _{12,15} E	–	–	–	–	0.68	<0.29	–
345.985	28 _{12,16/17} –27 _{12,15/16} A	–	–	–	–	0.78	2.0	–
346.001	28 _{12,17} –27 _{12,16} E	–	–	–	–	0.39	0.64	–
346.660	28 _{11,17} –27 _{11,16} E	<0.26	<0.47	<0.36	7.5	0.67	<0.32	<0.23
346.675	28 _{11,17/18} –27 _{11,16/17} A	<0.26	1.1	<0.36	11	0.67	<0.32	<0.23
346.687	28 _{11,18} –27 _{11,17} E	<0.26	0.27	<0.36	6.7	0.31	<0.32	<0.23
348.050	28 _{4,24} –27 _{4,23} E	<0.23	1.8	0.11	12	0.73	1.5	<0.26
348.066	28 _{6,23} –27 _{6,22} A	<0.23	1.6	<0.32	11	0.60	1.3	<0.26
348.909	28 _{9,20} –27 _{9,19} E	<0.23	1.2	<0.32	<0.43	<0.33	<0.36	<0.30
348.915	28 _{9,20} –27 _{9,19} A	<0.23	1.5	<0.32	7.7	<0.33	<0.36	<0.30
356.215	29 _{21,8/9} –28 _{21,7/8} A	<0.36	<0.47	<0.53	2.6	<0.41	<0.36	<0.68
356.297	29 _{20,9/10} –28 _{20,8/9} A	<0.36	0.49	<0.53	6.3	0.30	<0.36	<0.68
356.404	29 _{19,10/11} –28 _{19,9/10} A	<0.36	<0.47	<0.53	4.2	<0.41	<0.36	<0.68

– means frequency not observed. ^a Blend with 31_{13,18}–31_{12,19} A, and 31_{13,19}–31_{12,20} A. ^b Blend with 28_{14,14}–27_{14,13} A, and 28_{14,15}–27_{14,14} A.

Table B.15. Observed line fluxes $\int T_{\text{mb}}dV$ (K km s⁻¹) for CH₃OCH₃.

Frequency [GHz]	Transition	Sources						
		AFGL 2591	G24.78	G75.78	NGC 6334 IRS1	NGC 7538 IRS1	W3(H ₂ O)	W33A
222.029	21 _{2,20,2/3} –21 _{1,21,2/3}	<0.39	0.26	<0.32	3.1	<0.41	0.42	<0.25
222.035	21 _{2,20,0/1} –21 _{1,21,0/1}	<0.39	0.72	<0.32	2.5	<0.41	0.37	<0.25
222.239	4 _{3,2,2} –3 _{2,1,2}	<0.39	0.16	<0.32	1.6	<0.41	<0.27	<0.25
222.248	4 _{3,2,1/3} –3 _{2,1,1/3}	<0.39	0.86	<0.32	5.3	<0.41	0.37	0.37
222.255	4 _{3,2,0} –3 _{2,1,0}	<0.39	0.53	<0.32	4.7	<0.41	0.50	0.18
222.259	4 _{3,1,1/2} –3 _{2,1,1/2}	<0.39	0.60	<0.32	2.6	<0.41	0.48	0.18
222.326	25 _{3,23,K} –24 _{4,20,0}	<0.39	0.15	<0.32	2.3	<0.41	<0.27	<0.25
225.599	12 _{1,12,K} –11 _{0,11,K}	<0.41	3.1	0.73	11.8	0.83	4.0	1.0
226.347	14 _{1,13,K} –13 _{2,12,K}	–	1.5	<0.41	–	–	–	–
227.648	26 _{5,21,K} –26 _{4,22,K}	–	0.40	<0.26	–	<0.32	<0.39	–
227.658	24 _{3,22,0/1} –24 _{2,23,0/1}	–	0.81	<0.26	–	<0.32	<0.39	–
230.140	17 _{2,15,K} –16 _{3,14,K}	<0.39	1.7	<0.32	3.3	<0.41	1.4	<0.25
233.633	25 _{5,20,K} –25 _{4,21,K}	<0.15	1.1	0.24	2.6	<0.32	0.46	<0.23
240.978	5 _{3,3,2} –4 _{2,2,2}	<0.32	^a	<0.30	2.6	<0.35	<0.45	<0.26
240.985	5 _{3,3,1} –4 _{2,2,1}	<0.32	3.4 ^a	<0.30	8.3	0.41	2.8	0.56
240.990	5 _{3,3,0} –4 _{2,2,0}	<0.32	^a	<0.30	3.3	<0.35	<0.45	0.19

– means frequency not observed. K is either 0,1,2, or 3. ^a 5_{3,3,1}–4_{2,2,1} transition is blended with 5_{3,3,2}–4_{2,2,2} and 5_{3,3,0}–4_{2,2,0}.